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Undergraduate Research Experience on CAmpus

Category: 6 Student: Zhang Jie Project ID: SCBE13033

School of Chemical and Biomedical Engineering

Self-assembled Curcumin-PAH Nanoplex with Enhanced Solubility

INTRODUCTION

Self-assembled drug-polyelectrolyte complexation is a novel formulation of a nanoscale amorphous drug from its poorly soluble crystalline form, resulting in greater solubility and enhanced bioavailability of the drug. Nanoscale amorphous drug is formed by dispersing the ionised drug particles onto an oppositely charged polyelectrolyte (PE). The drug-PE complex is held together by strong electrostatic interaction.



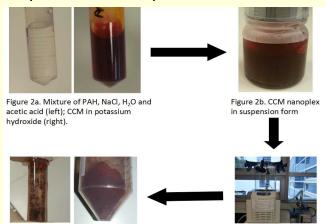
Curcumin (CCM) is a substance in turmeric that exhibits anti-inflammatory and chemo-preventive effects. However, research has shown poor absorption of the weakly acidic CCM by the body due to its poor solubility and fast metabolism. Increasing the solubility and prolonging drug release are critical for enhancing its absorption by the body.

Poly(allylamine hydrochloride) (PAH) is a cationic PE. It has several biomedical applications such as cell encapsulation and drug delivery.

This research aims to examine the solubility and duration of drug release of the self-assembled freeze-dried CCM-PE nanoplex.

METHODOLOGY

1. Preparation of CCM Nanoplex



2. Quantification of Supersaturation

Figure 2d. Dried CCM nanoplex (left);

Dried CCM nanoplex with hydroxypropyl methylcellulose (HPMC) (right).



15mg of CCM-PAH nanoplex and 30 mg of CCM-PAH nanoplex with hydroxypropyl methylcellulose (HPMC) were dissolved in 8.5mL of 1X phosphate buffered saline (PBS) respectively and incubated at 37°C. 300uL of the each incubated sample was withdrawn and filtered before 200uL of the withdrawn sample was diluted 10 folds with 1X PBS at specific time interval. The drug concentration was then measured to determine dissolution of the nanoplex.

Figure 2c. Freeze-dryer

Figure 3. Shaking incubator

RESULTS & DISCUSSION

Supersaturation Profile

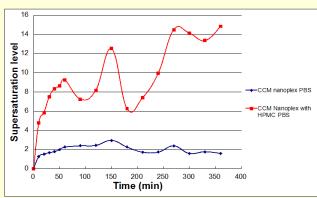


Figure 4. Supersaturation profile of CCM nanoplex in PBS and CCM nanoplex with HPMC in PBS.

CCM Nanoplex in PBS

- -Measured apparent solubility is slightly higher than the saturation solubility. The supersaturation level is between 1~3.
- -Increased dissolution rate due to the presence of PAH. But rapid precipitation prevent high level of supersaturation from being quantified.

CCM Nanoplex with HPMC in PBS

- -The supersaturation level increases to $^{\sim}9$ in 1 h and remains high though the supersaturation level fluctuates after 1 h. High supersaturation level over the next 5 h provides sufficient time for drug absorption.
- -HPMC slowers precipitation of CCM, allowing high supersaturation level of CCM nanoplex to be determined.

CONCLUSION

- The solubility of CCM is enhanced after complexation with PAH.
- High and prolonged supersaturation level of CCM nanoplex with HPMC provides sufficient time for drug absorption.

Future studies

The explanation for fluctuation of supersaturation level is unknown. More studies need to be done to examine the cause.

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