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Te-seeded Growth of Few- Quintuple Layer Bi_2Te_3 Nanoplates

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Te-seeded Growth of Few-Quintuple Layer Bi_2Te_3 Nanoplates

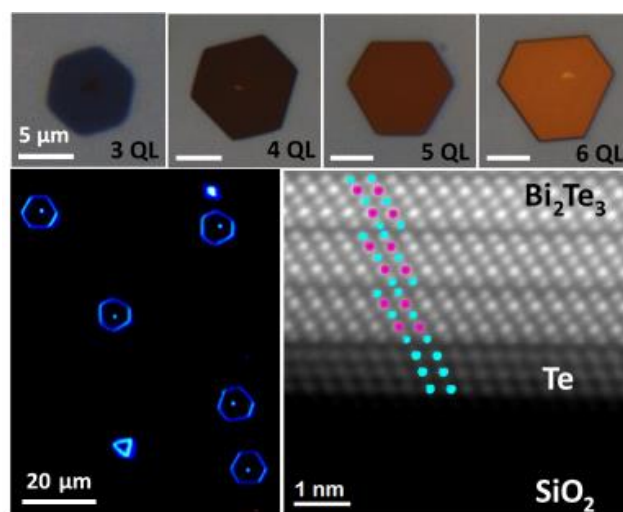
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Ultrathin (down to 3QL) Bi_2Te_3 nanoplates have been synthesized through a vapor transport method, where a Te-seeded epitaxial growth mechanism has been investigated. High optical contrast of few-QL Bi_2Te_3 on SiO_2/Si substrates has been demonstrated experimentally and computationally.

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ABSTRACT

We report on a Te-seeded epitaxial growth of ultrathin Bi_2Te_3 nanoplates (down to three quintuple layers) with large planar sizes (up to tens of micrometers) through vapor transport. Optical contrast has been systematically investigated for the as-grown Bi_2Te_3 nanoplates on the SiO_2/Si substrates, experimentally and computationally. The high and distinct optical contrast provides a fast and convenient method for the thickness determination of few-quintuple layer (QL) Bi_2Te_3 nanoplates. By aberration corrected scanning transmission electron microscopy, a hexagonal crystalline structure has been identified for the Te seeds, which form naturally during the growth process and initiate an epitaxial growth of the rhombohedral-structured Bi_2Te_3 nanoplates. The epitaxial relationship between Te and Bi_2Te_3 is identified to be perfect along both in-plane and out-of-plane directions of the layered nanoplate. Similar growth mechanism might be expected for other bismuth chalcogenide layered materials.

Introduction

Two-dimensional (2D) crystals have attracted tremendous interest in the past few years for their

richness in unusual physical and chemical properties and the potentials of novel applications [1]. 2D crystals exist in many categories of layered materials, including graphite, boron nitride,

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vanadium oxide, a large family of transition metal dichalcogenides, some group III, IV, and V metal chalcogenides, *etc.* [1-3]. Bi₂Te₃ is an important layered compound which has been historically well known for its excellent thermoelectric properties [4]. Recently, Bi₂Te₃, together with Bi₂Se₃ and Sb₂Te₃, were demonstrated as 3D topological insulators, a new state of quantum matter which has insulating bulk states but conducting surface states that are robust against any nonmagnetic disorder scatterings [5-7]. The intriguing surface states make topological insulators promising for unprecedented applications in spintronics, low-power electronics and quantum computing [8, 9]. Topological insulator nanostructures, especially 2D crystals, are advantageous compared to their bulk counterparts because of (1) the enhanced surface state contribution due to the extremely large surface-to-volume ratio; and (2) the enlarged bulk band gap caused by quantum size effect, which allows wider operation ranges for spintronic devices. Angular-resolved photoemission spectroscopy (ARPES) demonstrated that in the 2D limit, the nontrivial topological insulators would turn to trivial insulators [10, 11]. Recent transport measurements on topological insulator thin films and nanostructures have reported the ambipolar field effect, Aharonov-Bohm (AB) effect and Shubnikov-de Haas (SdH) oscillations that are associated with the surface states [12-16]. The thermoelectric performance has also been demonstrated to be largely enhanced in the ultrathin nanoplate-formed Bi₂Te₃/Bi₂Se₃ binary or ternary compounds due to the strong anisotropy [17-21]. Thus, the synthesis of high-quality ultrathin bismuth chalcogenide topological insulator 2D crystals is highly important for both the fundamental investigations and technological applications.

Most of the synthesized Bi₂Xe₃ (X=Se, Te) nanostructures come in two dimensional forms (nanobelt, nanoplate, nanofilm) because of their layered crystal structure with rhombohedral symmetry (D_{3d}^5) [22]. Five atomic layers (X-Bi-X-Bi-X) are covalently bonded, forming one non-polar quintuple layer (QL) with a thickness of around 1 nm. Adjacent QLs are weakly coupled via the weak van der Waals interactions and hence easy

cleavage is allowed perpendicular to the *c* axis along which the QLs pile up. Bi₂X₃ 2D crystals have been produced through a variety of methods, such as mechanical exfoliation [23, 24], molecular beam epitaxy (MBE) [25], vapor transport [26-29], chemical solution synthesis [17, 19, 21, 30], *etc.* Vapor transport is cost-effective, contamination-free, and feasible to achieve ultrathin Bi₂X₃ thin films. Different choices of substrates can lead to diverse products during the vapor transport process. Oriented Bi₂X₃ nanoplate arrays were prepared on mica and graphene through van der Waals epitaxy [27, 28, 31]. Si substrate has been commonly adopted for the synthesis of Bi₂X₃ nanostructures with diverse morphologies [32]. Inspired by the good optical contrast in the graphene/SiO₂/Si system [33], Si substrates capped with an oxide layer might enable the visibility and easy location of few-QL Bi₂X₃ crystals under optical microscope. Besides, SiO₂/Si substrate is also favorable for the *in-situ* field effect transistor (FET) device fabrication. Despite of a large number of reports on the vapor transport grown Bi₂X₃ 2D crystals, there still lacks a comprehensive investigation on the growth mechanism, which however, is essentially important in terms of controllable synthesis, structural design and functional engineering. Here, we report the synthesis of few-QL Bi₂Te₃ (down to 3 QLs) nanoplates and reveal a Te-seeded epitaxial growth process, supported by a comprehensive characterization using optical and transmission electron microscopy and spectroscopy. The optical contrast of few-QL Bi₂Te₃ nanoplates on SiO₂/Si substrates has been investigated, which provides a fast and convenient approach for the thickness determination.

Results and discussion

Bi₂Te₃ nanoplates were synthesized in a home-built vapor transport system (see Methods) [34, 35]. Figure 1A shows the bright field optical images of as-grown Bi₂Te₃ nanoplates on the 100 nm SiO₂/Si substrates. Most nanoplates exhibit hexagonal, triangular, or truncated triangular shapes with lateral sizes from several microns up to tens of microns. The typical thickness is about a few nanometers and the thinnest we could achieve is 3

nm, corresponding to 3 QLs. It is worth to note that a bright dot-like feature is found in the center of many nanoplates, as can be clearly seen in the dark field image (Figure 1B). Atomic force microscopy (AFM) measurements indicate that the bright dot is thicker than the rest of the nanoplates (Figure 1C). The central features are further investigated by scanning electron microscopy (SEM) and found to possess a particle or thin film morphology (see Figure S-1 in the Electronic Supplementary Material), likely corresponding to the nucleation center and a growing top layer, respectively. The as-grown nanoplates exhibit different colors under optical microscope, an indication of different thicknesses, similar with the case of other 2D crystals [36, 37]. Extensive AFM measurements confirm a one-to-one correlation between the colors and thicknesses. Figure 1D shows the bright field optical images of 3-9 QL and even thicker Bi₂Te₃ nanoplates. The colors (optical contrasts) for 8 QL and below are very distinct for each thickness and can be easily distinguished by eyes. The excellent optical contrasts for few-QL Bi₂Te₃ provides a quick and convenient approach for the nanoplate location and thickness identification under the optical microscope, which is highly favorable for further optical characterizations and electronic device fabrications.

Optical contrast in the three-layer thin film system (Bi₂Te₃/SiO₂/Si, Figure 2A) originates from the interference of the reflected light from different interfaces. The presence of the opaque Bi₂Te₃ thin film adds an optical path, changing its interference color respect to the empty SiO₂/Si substrates [33]. The optical contrast of Bi₂Te₃ thin flakes has previously been studied by Li *et al.*, in which they focused on the contrast optimization under different illumination wavelengths, while the experimental contrast data of few-QLs was not comprehensively presented [38]. In our study, we demonstrate a distinctive optical contrast under white light illumination for each thickness in the few-QL regime, experimentally and theoretically. The optical contrast of the Bi₂Te₃ nanoplates can be quantitatively represented by the Michelson contrast [39]

$$\text{optical contrast} = \frac{R_{\text{Bi}_2\text{Te}_3} - R_{\text{SiO}_2}}{R_{\text{Bi}_2\text{Te}_3} + R_{\text{SiO}_2}},$$

where $R_{\text{Bi}_2\text{Te}_3}$ is the reflected light intensity from the nanoplate while R_{SiO_2} is that from the bare SiO₂/Si substrates, both of which can be measured via a micro-reflection spectrometer (Craic 20) in our experiments. The value of optical contrast ranges from -1 to 1, while the positive (negative) sign indicates a stronger (weaker) reflection from the Bi₂Te₃ nanoplate than from the substrates. Figure 2B displays the measured optical contrast spectra in the visible region (400~750 nm) for few-QL and bulk Bi₂Te₃ nanoplates. A contrast minimum can be seen in the spectra for 3-8QL, exhibiting a blue shift while increasing the thickness, as denoted by the arrow. For even thicker nanoplates (12-QL and above), the contrast minimum probably shifts out of the visible region and cannot be identified. The 3-QL and 4-QL nanoplates exhibit negative optical contrast through the whole visible region, consistent with the fact that they appear darker than the substrates in the optical images. From 5QL, positive contrast becomes more and more dominant with increased thickness and covers the whole visible range in the limit of bulk, explaining why thicker nanoplates appear brighter to eyes under the optical microscope.

To explain the observed optical contrast in our Bi₂Te₃/SiO₂/Si system, calculations were carried out based on the Fresnel's equations with a simple normal incidence geometry (as shown in Figure 2A) [33]. The refractive indices of the four media (air, Bi₂Te₃, SiO₂, and Si) required for the calculations are obtained from existing literatures [40-42]. The calculated optical contrasts for 1-200 QL Bi₂Te₃ on 100 nm SiO₂/Si substrates are shown in Figure 2C. The contrast spectra for 50 QL and above are identical, indicating that a 50-QL Bi₂Te₃ nanoplate should appear the same as the bulk crystal. The calculated results show a blue shift of the contrast minimum as the thickness increases, qualitatively in agreement with the experiments. Quantitatively, the experimental and calculated results are in good agreement for bulk Bi₂Te₃ but not for few-QL nanoplates. This discrepancy indicates that the refractive index of few-QL Bi₂Te₃ might be different from that of their bulk counterpart, given that the refractive index of bulk Bi₂Te₃ was used in the calculations for all thicknesses. Another factor that

could contribute to the discrepancy is the simplified normal incidence model employed for our contrast calculations, while in the experiments an objective with numerical aperture of 0.5 was used. A more complex model [39] considering the incident light angle might be able to characterize our system better but it is beyond the scope of this manuscript. The thickness of the SiO₂ layer is the key factor to determine the optical contrast in the Bi₂Te₃/SiO₂/Si system. Our calculations found that 100 nm SiO₂ gives the highest contrast under visible illumination for few-QL Bi₂Te₃ (see Figure S-2 in the Electronic Supplementary Material), consistent with previous calculated results for the same system [38].

A detailed scanning transmission electron microscopy (STEM) characterization has been conducted on the Bi₂Te₃ nanoplates. A PMMA transfer method was used for the TEM sample preparation (see Method). Figure 3 shows the released Bi₂Te₃ samples on lacey carbon studied under aberration corrected high-angle annular dark-field (HAADF) STEM conditions. The image contrast is proportional to the nanoplate thickness, where the brighter hexagon (A) is thicker than the darker one (B). The nanoplates crystallize in the Tellurobismuthite $R\bar{3}m$ phase with the hexagonal facets corresponding to the $\{01\bar{1}0\}$ planes, and the growth direction is along the $[0001]$ axis perpendicular to the hexagon plane, as identified from the selected-area electron diffraction (SAED) pattern (Figure 3C). It is worth to note that small holes were found in the center of the nanoplate and notches on the edges. The notches on the edges were probably caused by the accidental damage during the sample preparation process. Cheng *et al.* have reported similar center-hollowed Bi₂Te₃ nanoplates broken from T-shaped Bi₂Te₃(nanoplate)-Te(nanorode) heterojunctions [43]. In our as-prepared TEM samples, the holed features were commonly present in the released Bi₂Te₃ nanoplates and we tentatively propose that they might be related to the nucleation seeds (bright spots in Figure 1), which are left over during the releasing process, causing the holes in the released nanoplates.

To further characterize the possible nucleation seeds in the center of the Bi₂Te₃ nanoplates, planar view TEM samples were prepared from the

as-grown Bi₂Te₃ nanoplates on SiO₂/Si substrates by thinning the backside of the substrates using conventional electron microscopy sample preparation methods. Figure 4A shows the HAADF STEM image of a typical nanoplate with truncated triangular morphology from the studied area (Figure 4C) and its surface plot is presented in Figure 4B. A darker contrast (denoted by the black square in Figure 4A and the orange square in Figure 4B) was revealed in the nanoplate, and is clearly shown in the intensity profile along the direction indicated by black arrow (Figure 4D). The contrast difference can be due to a thickness change or to a compositional variation. The HRTEM analysis of this darker contrast region shows a particle-like agglomerate with a size of around 50 nm (Figure 4E), which is likely to correspond to the nucleation seeds that appear as bright dots in the Bi₂Te₃ nanoplates under the optical microscope (see Figure 1). Through the study of the fast Fourier transform (FFT) (Figure 4F) of the image E, we can identify not only the Bi₂Te₃ and the Si phases, but also a hexagonal tellurium phase ($P3_121$). The appearance of the Te phase can explain the darker contrast region in Figure 4A, since the image intensity is proportional to the squared atomic number ($Z^2_{\text{Bi}} = 6889$; $Z^2_{\text{Te}} = 2704$). Moreover, the Te phase shows a $(11\bar{2}0) [0001] \text{ Te} // (11\bar{2}0) [0001] \text{ Bi}_2\text{Te}_3$ epitaxial relationship with the Bi₂Te₃ phase, as denoted in Figure 4F. Notice that the green hexagon in Figure 4F represents the $\{1\bar{2}10\}$ lattice planes of both the Bi₂Te₃ and Te phases while the yellow hexagon only represents the $\{10\bar{1}0\}$ planes of Te, in agreement with our simulated diffraction patterns (Figure S-3 in the Electronic Supplementary Material). The epitaxial relationship between Te and Bi₂Te₃ can be further identified as perfect epitaxy along the c axis since the in-plane lattice mismatch is only 1.6% for the Te ($a=4.447 \text{ \AA}$) and Bi₂Te₃ ($a=4.375 \text{ \AA}$) phases [44], in agreement with previous reports on the Bi₂Te₃/Te heterostructures [43, 45]. The crystalline Te phase and the Te/Bi₂Te₃ epitaxy have also been observed in several other Bi₂Te₃ nanoplates in the studied area (Figure 4C).

The appearance of the Te phase in the Bi₂Te₃ nanoplates can be understood from analyzing the detailed vapor transport growth process-Bi₂Te₃ sublimation and recrystallization. Early

measurements on the vapor pressure of crystalline Bi_2Te_3 indicated that the sublimation occurred congruently below the melting temperature 585°C , following the decomposition reaction $\text{Bi}_2\text{Te}_3(\text{s}) \rightarrow 2\text{BiTe}(\text{g}) + 1/2\text{Te}_2(\text{g})$ [46–48], while later Brebrick *et al.* casted doubt on the validity of the proposed reaction for an equilibrium sublimation and suggested an incongruent sublimation process with Te-rich (atomic percent > 60%) vapor species [49]. A Te-rich composition will lead to a phase segregation during the crystallization process and thus the formation of two equilibrium crystalline phases, Bi_2Te_3 and Te, according to the Bi-Te binary phase diagram [50]. During the vapor transport growth, BiTe and Te_2 molecules (major vapor species) sublimating from the Bi_2Te_3 powder source are carried with the gas flow to a lower temperature zone and deposit on the substrates into Bi_2Te_3 nanoplates, according to the synthesis reaction $2\text{BiTe}(\text{g}) + 1/2\text{Te}_2(\text{g}) \rightarrow \text{Bi}_2\text{Te}_3(\text{s})$. The wide presence of the Te crystalline seeds in as-grown Bi_2Te_3 nanoplates implies an excess of the Te_2 molecules in the sublimated vapor species, in agreement with the report from Brebrick *et al.* [49]. The identification of the Te/ Bi_2Te_3 epitaxial relationship as elucidated by transmission electron microscopy, suggests a Te-seeded epitaxial growth of Bi_2Te_3 nanoplates in our vapor transport process.

To further examine the epitaxial growth mechanism proposed, cross-section samples have been analyzed under HAADF STEM conditions by using an aberration-corrected microscope. Figure 5A shows a HAADF STEM image of a nanoplate with thickness of around 300 nm, along with its surface plot below to show the nanoplate morphology. Electron energy loss spectroscopy (EELS) analysis was performed to study the elemental constitution of the nanoplate and the mapping results are presented as Figure 5B. The bright cyan color on the right side of the nanoplate corresponds to a pure Te structure aggregated at the lateral facet of the plate. Furthermore, it was found that the Te phase exists in the pinholes in the SiO_2 layer until it reaches the Si substrate, while the Bi signal is restricted only to the nanoplate. Figure 5C (pink squared region in Figure 5A) shows the atomic resolution HAADF STEM image of the

interface between the Bi_2Te_3 nanoplate and the SiO_2 layer of the substrates, where five atomic layers Te-Bi-Te-Bi-Te can be identified as a QL (the Bi and Te atoms are highlighted as pink and cyan dots, respectively). Between the bottom QL of the Bi_2Te_3 nanoplate and the SiO_2 substrates, three atomic layers of Te can be seen. [A van der Waals gap can be identified between the Te phase and the bottom \$\text{Bi}_2\text{Te}_3\$ QL, implying the lowest interface energy in this case.](#) Again, the perfect epitaxial relationship can be directly identified between the Te atomic layers and the bottom QLs of the Bi_2Te_3 nanoplate along the c axis. The Te crystalline layer at the interface is only present in the pinholed region and should be closely related to the Te nucleation seed as discussed above. We tentatively propose that most of the Te nucleation happens at the pinhole pits, since thermodynamically the substrate imperfections could facilitate the heterogeneous nucleation by vapor condensation [51]. The pinholed feature embedded in the SiO_2 layer may result in an attachment between the nucleation seed and the Si substrates. During the nanoplate releasing process through HF etching, the SiO_2 layer gets etched away following the reaction $\text{SiO}_2 + 4\text{HF} \rightarrow \text{SiF}_4(\text{g}) + 2\text{H}_2\text{O}$, while the Te pinhole remains. The Te seed might be tore off the rest of the nanoplate due to its attachment to the Si substrate, thus leaving a holed feature in some released Bi_2Te_3 nanoplates (Figure 3A).

Figure 5D (blue squared region in Figure 5A) displays the atomic resolution interface between the Bi_2Te_3 nanoplate and the laterally attached Te structure, where an imperfect epitaxial relationship can be observed with the presence of some misfit dislocations at the interface. This epitaxial relationship can be further studied from the FFT analysis of the blue and green squared regions as $(0001)[\bar{1}2\bar{1}0] \text{ Bi}_2\text{Te}_3 // (0001)[\bar{1}2\bar{1}0] \text{ Te}$ (see Figure 5G & H), with a slight angle between the (0001) planes of the two phases. The lattice mismatch along the c axis between the Te and Bi_2Te_3 phases is around 2.7% ($c=30.39 \text{ \AA}$ for Bi_2Te_3 ; $c=5.92 \text{ \AA}$ for Te) [44], and a perfect in-plane epitaxy is expected for such a small mismatch, as the case out of plane. The observed imperfect in-plane epitaxy can be explained by the inhomogeneity of Bi_2Te_3 along the c axis. The atomic layers are evenly spaced along

the c axis in Te but not in Bi_2Te_3 due to the presence of the van der Waals gaps. In fact, 2.7% is underestimating the lattice mismatch assuming an even atomic-layer spacing along the c axis for both Bi_2Te_3 and Te, and the misfit dislocations observed at the interface can be attributed to a real lattice mismatch that is much bigger. After a few nanometers from the interface, the Te phase relaxes from the strain caused by the interface and aligns perfectly in plane with the Bi_2Te_3 phase. Thus far, an epitaxial relationship between the six-fold symmetry Te ($P3_121$) and Bi_2Te_3 ($R\bar{3}m$) phases has been confirmed both along and perpendicular to the c axis. It is worth to note that two different Bi_2Te_3 crystalline orientations were observed in the orange squared region (Figure 5E) and the blue squared region (Figure 5D), indicating the presence of a twin boundary somewhere in between. In polycrystalline Bi_2Te_3 , (0001) basal twin boundaries have been observed with a termination at the van der Waals gaps between two adjacent QLs [52, 53]. This twin boundary configuration has also been confirmed to be energetically favorable, according to the *ab initio* calculations [52]. Therefore, we believe that the same (0001) basal twin boundary also presents in the nanoplate studied here. Considering the fact that the crystalline orientation keeps the same in a range of tens of QLs (between the orange and pink regions), we assume that twin boundaries are unlikely to present in the case of few-QL nanoplates.

The nanoplate studied in Figure 5 well represents some percentage of as-grown nanoplates where the central Te nucleation seeds cannot be identified from the top-view optical and STEM images. Note that during the vapor transport growth, the volume of both Bi_2Te_3 and Te crystalline phases will gradually increase with a continuous Te-rich vapor supply and condensation, as a result of the Bi_2Te_3 -Te phase segregation, as discussed above. In the condition of fast Bi_2Te_3 phase formation, the central Te seed might get merged in the Bi_2Te_3 nanoplate. Consequently, the growth of the nucleation seed will be cut off with a lack of Te supply. Instead, the excess Te adatoms will crystallize into a new Te crystalline phase on the surface of the nanoplate, corresponding to the Te aggregation shown in Figure 5B. We tentatively

believe that extra Te aggregation is most likely to present on large and thick nanoplates, where the nanoplate growth is more rapid and larger quantity of Te phase is expected, in comparison with the few-QL cases.

Thus far, combining all the above characterizations (optical microscopy, AFM, SEM, aberration corrected HAADF STEM), we propose a nanoplate growth model: during the early growth stages, the Te-rich vapor species sublimated from Bi_2Te_3 powder source condensate on the substrates (preferably at the pinhole locations), forming Te crystalline seeds. During the condensation process, some vapors are inclined to diffuse through the SiO_2 pinholes and then solidify, resulting in an attachment between the nucleation seeds and the Si substrates. Meanwhile, the BiTe and Te_2 vapors will get adsorbed to the Te nucleation seeds and react into Bi_2Te_3 , growing epitaxially into nanoplates. The epitaxial growth occurs both laterally and vertically, following a layer-by-layer manner. The thickness of the nanoplate is determined by the size of the nucleation seed as well as the supply of BiTe and Te_2 vapors. The absence of the nucleation seeds in some nanoplates likely suggest that the Te seeds either get merged into the Bi_2Te_3 nanoplates due to the fast growth process, or react with BiTe vapors and are completely converted to Bi_2Te_3 .

Conclusions

Few-QL topological insulating Bi_2Te_3 nanoplates with large planar sizes have been synthesized through vapor transport. Optical contrast in the $\text{Bi}_2\text{Te}_3/\text{SiO}_2/\text{Si}$ system was systematically studied both experimentally and computationally. The high optical contrast provided by the 100 nm SiO_2/Si substrates provides a fast and convenient approach for the location and thickness determination of few-QL Bi_2Te_3 nanoplates. The nanoplate growth mechanism has been comprehensively investigated and a Te-seeded growth model was revealed, where crystalline Te seeds firstly form thermodynamically and then initiate the epitaxial growth of the Bi_2Te_3 nanoplates. The epitaxial relationship between Bi_2Te_3 and Te can be taken advantage of in different types of growth methods, shedding light on the potential growth and research on the $\text{Bi}_2\text{Te}_3/\text{Te}$

superlattices. This simple and effective growth mechanism might be extended to other topological insulator thin film growth like Bi_2Se_3 and Sb_2Te_3 .

Methods

Vapor transport growth

The Bi_2Te_3 nanoplates were synthesized in a 2-inch quartz tube placed inside of a tube furnace (Lindberg/Blue M) [34, 35]. Bi_2Te_3 power (99.99%, Alfa Aesar) was positioned in the center of the quartz tube during the growth. Commercial (100) Si substrates capped with 100 nm thermal oxidized layer were cleaned by Isopropyl alcohol and placed around 15 cm downstream from the powder source during the growth. The growth chamber was firstly evacuated to ~ 5 mTorr and then flushed several times with the carrier gas (Ar or H_2). The nanoplate growth was conducted at 480°C with a 30 sccm, 100 Torr Ar/ H_2 flow for 10 min. The adoption of Ar or H_2 as carrier gas gives similar growth products.

Bi_2Te_3 nanoplate release

SiO_2/Si substrates with as-grown Bi_2Te_3 nanoplates on top were firstly spin coated a layer of polymethyl methacrylate (PMMA) with the thickness of around 300 nm and then baked under 180°C for 5 min. After immersing the PMMA coated substrates in buffered HF solution for a few hours, the SiO_2 layer would be etched away and the PMMA film with Bi_2Te_3 nanoplates embedded in was released from the Si substrates. The PMMA film was then rinsed off using deionized water and transferred to TEM grids covered with lacey carbon. PMMA was then removed by acetone, leaving Bi_2Te_3 nanoplates on the lacey carbon.

Planar and cross-section STEM sample preparation

The cross-section samples have been prepared by gluing two pieces of substrate containing the grown nanoplates, placing the nanoplates face-to-face. Then, conventional polishing has been laterally performed up to getting an around 20 μm thick sample, which is then attached to a copper grid. Finally, precision ion polishing was carried out in a Precision Ion Polishing System (PIPS) to obtain an

electron transparent sample.

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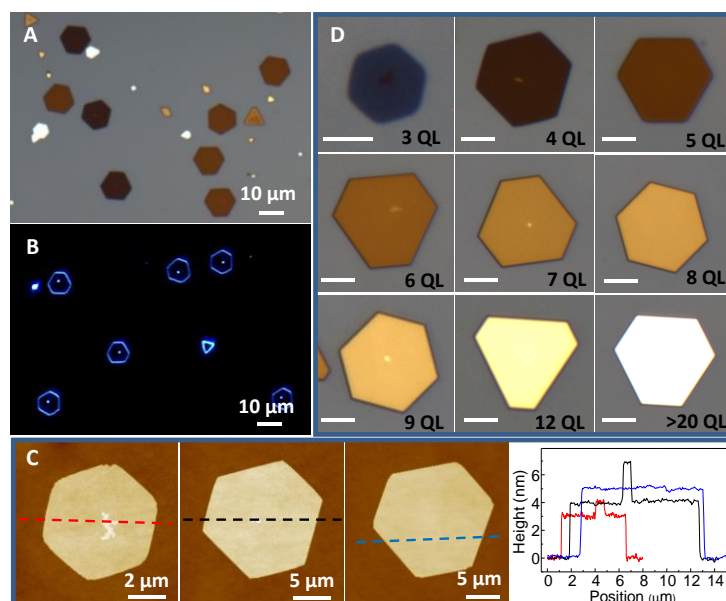


Figure 1. (A) Bright field optical image of as-grown Bi_2Te_3 nanoplates on 100 nm SiO_2/Si substrates. Nanoplates of various thicknesses show different colors in the bright field image. (B) Dark field optical image of as-grown Bi_2Te_3 nanoplates. A bright spot can be seen in the center of many nanoplates, which works as the nucleation seed to initiate the nanoplate growth. (C) AFM images of 3-QL, 4-QL and 5-QL Bi_2Te_3 nanoplates. The thickness profiles are taken from the corresponding colored dashed lines. (D) Optical images of Bi_2Te_3 nanoplates with thicknesses from 3 QLs to more than 20 QLs. All the scale bars in (D) are 5 μm .

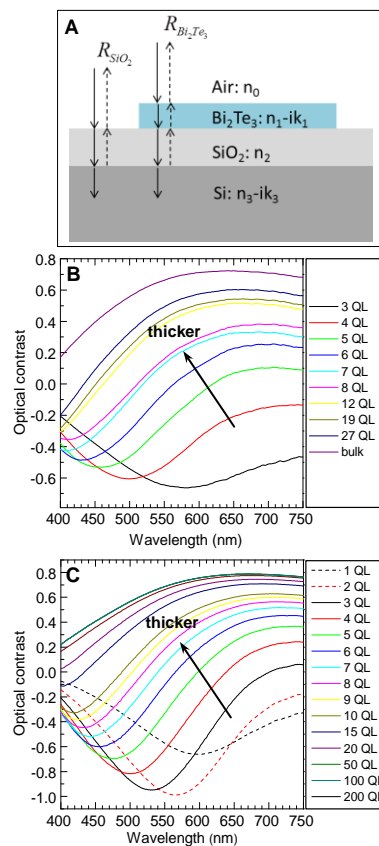


Figure 2. (A) Schematic of the optical reflection and transmission with normal incidence in the three-layer thin film system. Solid lines indicate the incident light and dashed lines stand for the reflected light. (B) Measured optical contrast spectra of as grown few-QL Bi_2Te_3 nanoplates on 100 nm SiO_2/Si substrates. (C) Calculated optical contrast of Bi_2Te_3 thin films with thicknesses from 1 QL to bulk. The spectra for 1-QL and 2-QL are shown in dashed lines while those for 3-QL and above are in solid lines.

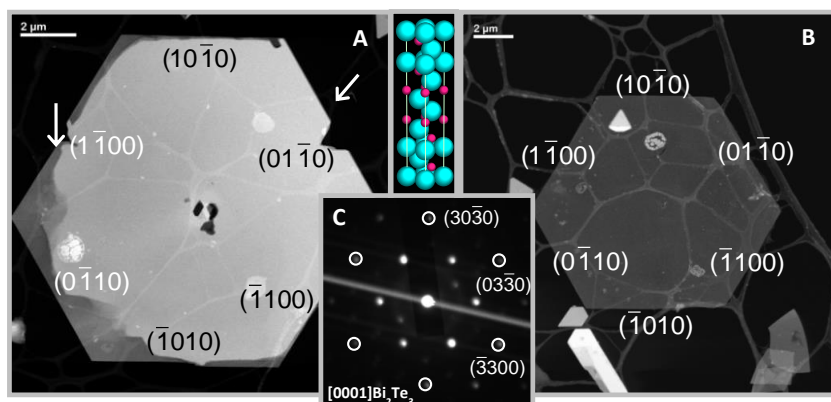


Figure 3. HAADF STEM images of two hexagonal Bi_2Te_3 nanoplates on lacey carbon. The image contrast is proportional to the plate thickness. (A) a thicker nanoplate with holes in the center, where the seed was located. (B) a thinner nanoplate. The nanoplates crystallize in the Tellurobismuthite $R\bar{3}m$ phase. The facets of the hexagons are $\{01\bar{1}0\}$ planes, as observed from the SAED pattern in (C).

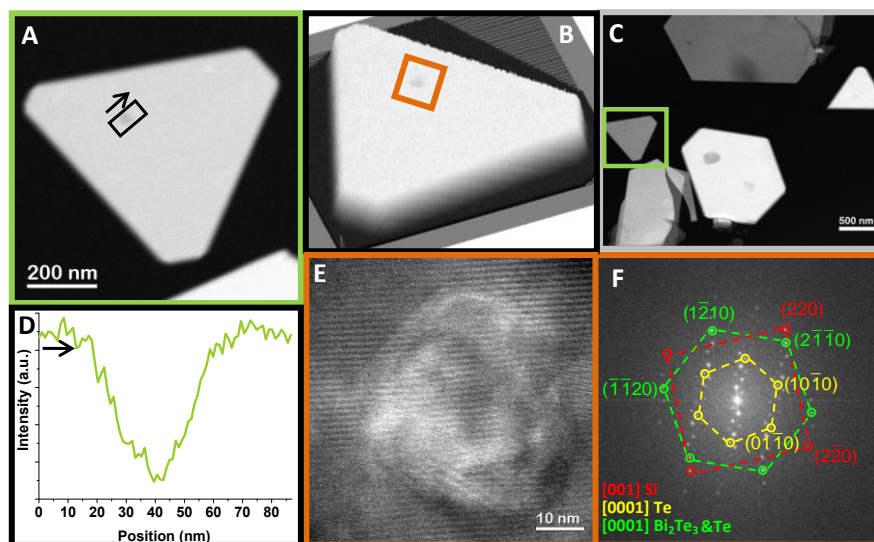


Figure 4. Planar view of Bi_2Te_3 nanoplates on SiO_2/Si substrates. (A) HAADF of the studied nanoplate. (B) Surface plot of the plate in (A). (C) Low magnification HAADF image of the studied area. (D) Intensity profile taken from the black square in (A) along the arrow indicated direction. (E) HRTEM image of the darker feature in the black squared region in (A). (F) FFT of image (E).

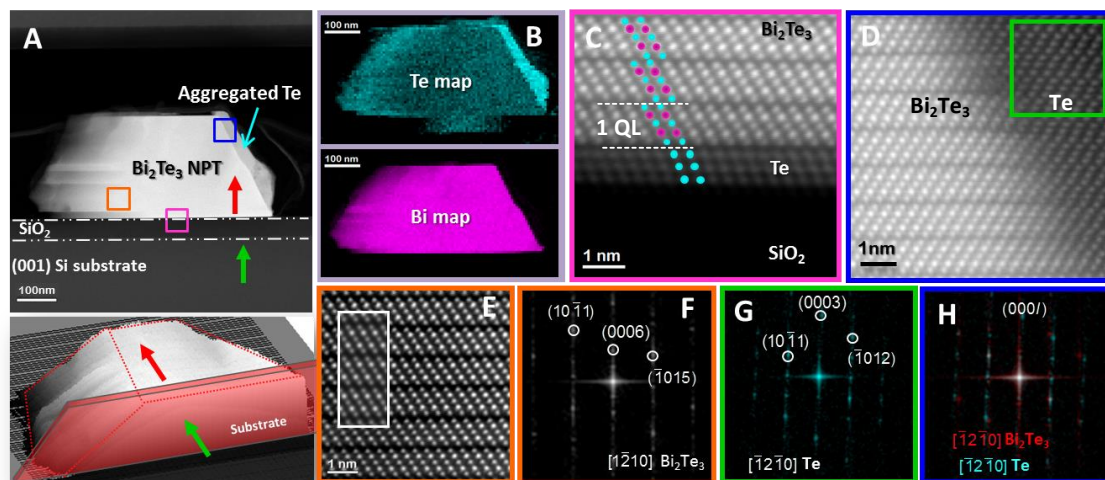


Figure 5. STEM cross-section analysis of an as-grown Bi_2Te_3 nanoplate on SiO_2/Si substrates. (A) HAADF image of the nanoplate studied, displayed also as surface plot below. The colored frames indicate the regions where the atomic resolution HAADF STEM analyses were performed. (B) Mapping of the EELS Te signal (cyan colored) and Bi signal (pink colored). (C) Bottom interface between the Bi_2Te_3 nanoplate and SiO_2/Si substrates taken in the region pink squared in (A), showing a few pure Te layers in between. One QL of Bi_2Te_3 is denoted by the white dashed lines. (D) Interface between the Bi_2Te_3 nanoplate and the attached Te structure on the lateral side, taken in the blue squared region in (A). (E) Atomic resolution HAADF STEM image of the Bi_2Te_3 structure, taken in the region orange squared in (A). The inset corresponds to the simulated structure by using the STEM-CELL software [54]. (F) FFT of (E). (G) FFT of the Te aggregation, green squared in (D). (H) FFT of (D), showing the epitaxy between the Bi_2Te_3 and the Te aggregation.

Electronic Supplementary Material

Te-seeded Growth of Few- Quintuple Layer Bi_2Te_3 Nanoplates

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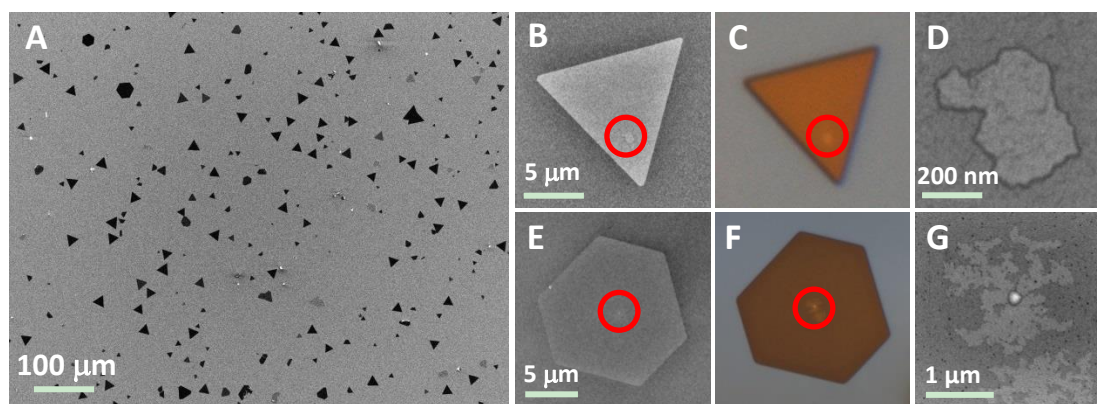


Figure S-1. (A) low-magnification SEM image of as-grown Bi_2Te_3 nanoplate. (B) High-magnification image of a triangular nanoplate with a dot-like feature in the center of the nanoplate which is highlighted by the red circle. (C) Corresponding optical image of the nanoplate in (B). (D) Zoom-in image of the dot-like feature in (B), showing a growing top layer. (E) High-magnification image of a hexagonal nanoplate with a dot-like feature in the center of the nanoplate which is highlighted by the red circle. (F) Corresponding optical image of the nanoplate in (E). (G) Zoom-in image of the dot-like feature in (E), showing a growing top layer and the nucleation center.

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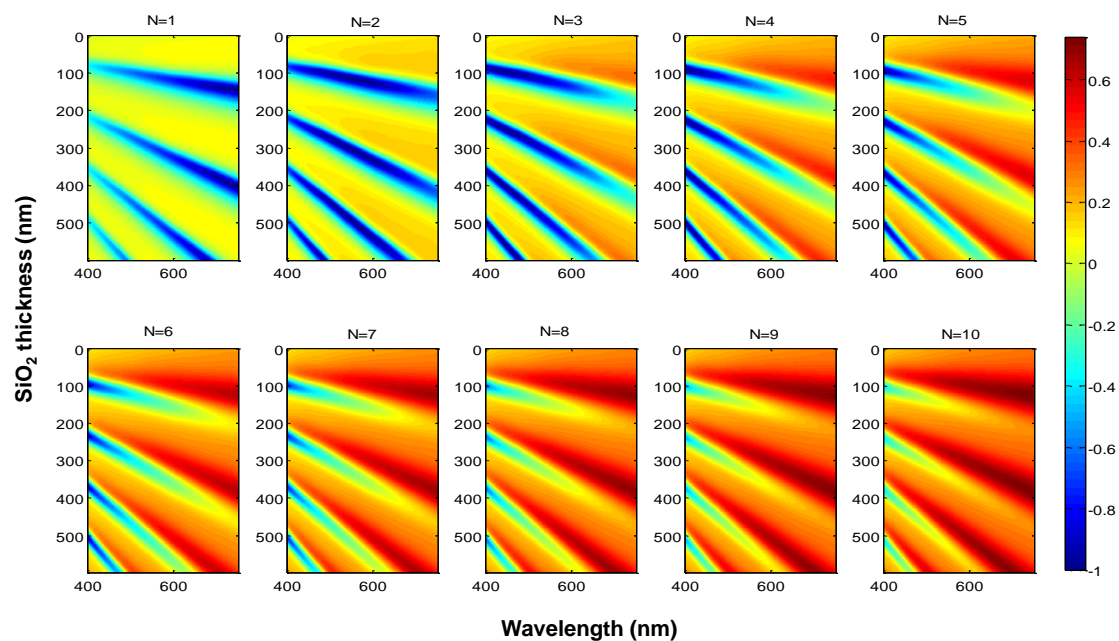


Figure S-2. Two dimensional plot of the calculated optical contrast for the 1-10QL Bi_2Te_3 on SiO_2/Si substrates as a function of the wavelength and the thickness of the SiO_2 layer. It shows that the 100 nm SiO_2 layer gives the best optical contrast for few-QL Bi_2Te_3 nanoplates.

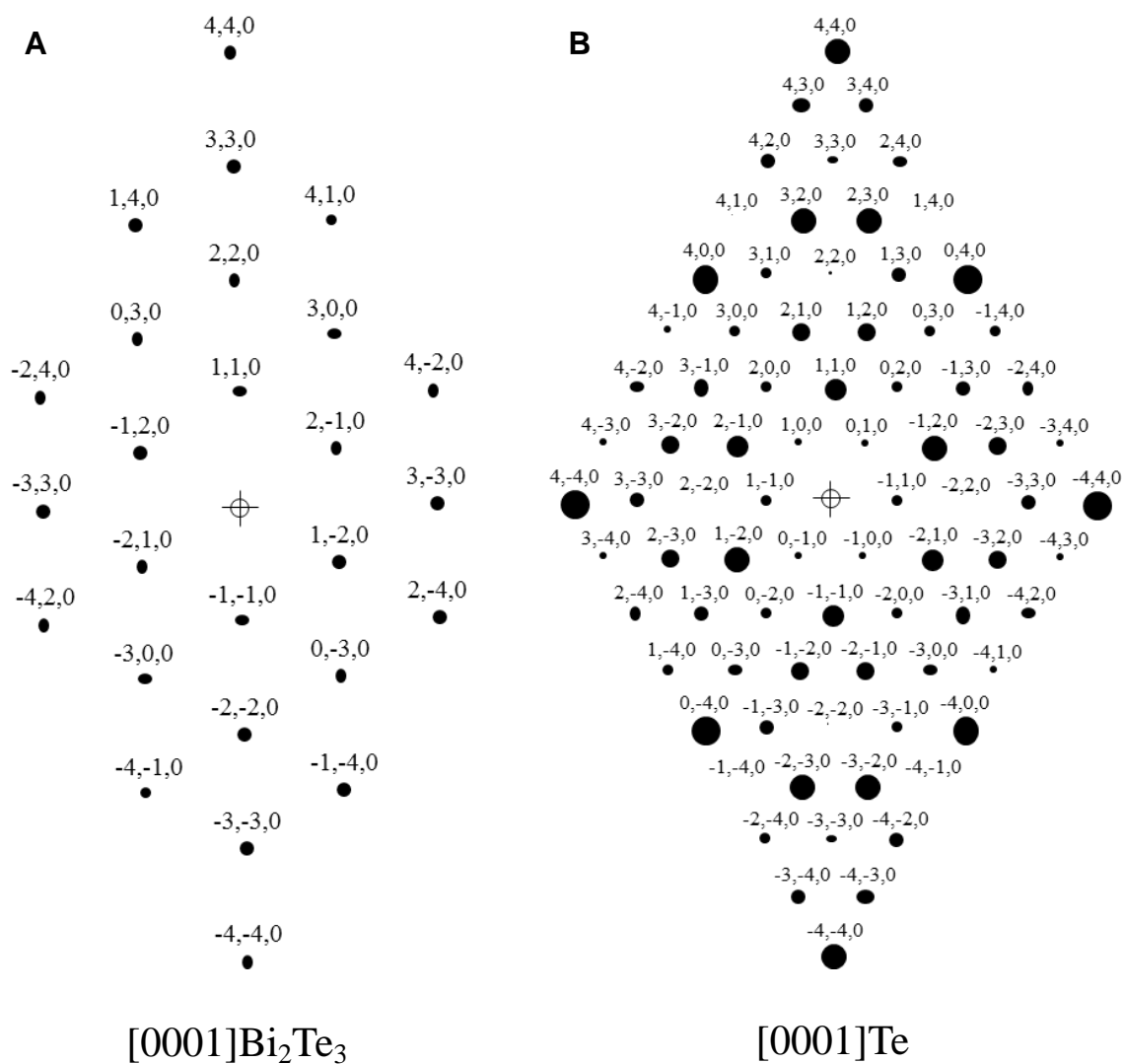


Figure S-3. Simulated diffraction patterns for crystalline (A) Bi_2Te_3 ($R\bar{3}m$) and (B) Te ($P3_121$). The size of the diffracted spot is proportional to the diffraction intensity. Visualizing along the $[0001]$ zone axis, the $\{10\bar{1}0\}$ planes of Bi_2Te_3 should extinguish while those of the Te phase are visible with a high intensity. Notice that the simulated patterns are expressed in 3-index notation while we are using 4-index in the whole manuscript. Transformation from 3-index planes to 4-index notations can be easily obtained by following: $(h, k, l) = (h, k, -(h+k), l)$. In this way, the $\{10\bar{1}0\}$ planes correspond to the $\{100\}$ in the simulated patterns.