This document is downloaded from DR-NTU (https://dr.ntu.edu.sg) Nanyang Technological University, Singapore.

Perovskite-ion beam interactions: toward controllable light emission and lasing

Wang, Yue; Gu, Zhiyuan; Ren, Yinjuan; Wang, Ziming; Yao, Bingqing; Dong, Zhili; Adamo, Giorgio; Zeng, Haibo; Sun, Handong

2019

Wang, Y., Gu, Z., Ren, Y., Wang, Z., Yao, B., Dong, Z., . . . Sun, H. (2019). Perovskite-ion beam interactions: toward controllable light emission and lasing. ACS Applied Materials & Interfaces, 11(17), 15756-15763. doi:10.1021/acsami.9b01592

https://hdl.handle.net/10356/142815

https://doi.org/10.1021/acsami.9b01592

This document is the Accepted Manuscript version of a Published Work that appeared in final form in ACS Applied Materials & Interfaces, copyright © American Chemical Society after peer review and technical editing by the publisher. To access the final edited and published work see https://doi.org/10.1021/acsami.9b01592

Downloaded on 20 Mar 2024 17:45:15 SGT



Processing all-inorganic halide perovskites for controlled light emission and lasing by focused ion beam

Yue Wang^{1,2}, Zhiyuan Gu², Yinjuan Ren³, Ziming Wang¹, Bingqing Yao⁴, Zhili Dong⁴, Giorgio Adamo⁵, Haibo Zeng¹*, Handong Sun^{2,5,6}*

¹MIIT Key Laboratory of Advanced Display Materials and Devices, Institute of Optoelectronics & Nanomaterials, College of Materials Science and Engineering, Nanjing University of Science and Technology, Nanjing 210094, China

²Division of Physics and Applied Physics, School of Physical and Mathematical Sciences,

Nanyang Technological University, Singapore 637371, Singapore

³Department of Chemistry, National University of Singapore, 3 Science Drive 3, Singapore 117543, Singapore

⁴School of Materials Science and Engineering, Nanyang Technological University, Nanyang Avenue, Singapore 639798, Singapore

⁵Centre for Disruptive Photonic Technologies (CDPT), Nanyang Technological University, Singapore 637371, Singapore

⁶MajuLab, CNRS-UCA-SU-NUS-NTU International Joint Research Unit, Singapore

* Authors to whom correspondence should be addressed, electronic email: hdsun@ntu.edu.sg; zeng.haibo@njust.edu.cn

Abstract: All-inorganic cesium lead halide perovskites hold great promise for the

development of next-generation optoelectronics. However, it remains unexplored how the

energetic ions will impact CsPbX₃, which may largely limit the application potentials. In this

work, we for the first time investigate the interaction between the CsPbX₃ and high-energy

gallium ions in a broad range of ion doses provided by a focused ion beam (FIB) system. We

found that the optical properties of CsPbX₃ are highly sensitive to the energetic Ga⁺ ions due

to the relatively vulnerable ionic bonding. Specifically, even low-dose Ga⁺ irradiation

(~1×10¹⁵ ions/cm²) can lead to more than one-order-of-magnitude reduction in the

photoluminescence (PL) intensity, which can be attributed to the combined effects of the

formation of vacancy/interstitial defects, generation of metallic Pb-related nonradiative

recombination centers and crystal-to-amorphization transition. With the increase of ion dose

(~10¹⁷ ions/cm²), the morphology of CsPbX₃ can be dramatically altered due to the ion

sputtering effect. We demonstrate that both low- and high-dose FIB treatment can be

important for realizing the application prospects of CsPbX3 in optical security protection and

system-on-a-chip compatible microlasers. Our results offer significant information about the

ion impacts on CsPbX₃ and offer an enabling tool to manipulate the emission and lasing from

CsPbX₃, which could push ahead the potential of CsPbX₃ in photonics and optoelectronics.

Keywords: Inorganic perovskites, focused ion beam, ion-pervoskite interaction, microlaser.

Organic-inorganic lead halide perovskites have been attracting intense interest by virtue of

their intriguing potential in photovoltaics, light-emitting diodes (LEDs) and lasers.^[1-5]

However, these organic-inorganic perovskites are notorious to degrade upon exposure to heat, oxygen and moisture, which hinders their practical applications and commercialization.^[1, 4] Notably, all-inorganic cesium lead halide perovskites (CsPbX₃, X=Cl, Br, I) had been demonstrated to possess superior optical/electric properties and much enhanced stability compared to the organic-inorganic analogues, thus holding great promise for the development of next-generation optoelectronics.^[6-8] Up till now, solar cells based on CsPbI₃ have reached 10.77% in power conversion efficiency (PCE) and the CsPbBr₃-based LEDs have manifested a high external quantum efficiency of up to 20.3%.^[9, 10]

Recently, the interaction between CsPbX₃ and high-energy X-rays as well as gamma-rays (γ-rays) has been explored and led to the discovery of the highly sensitive nuclear radiation detectors and scintillators based on CsPbX₃.^[11-13] As another high-energy radiation, energetic ions are well-known to be able to tailor the material structure and properties, resulting in a large number of important applications including microelectronics, radiotherapy treatment, and engineering of nanomaterials.^[14-16] For example, dramatic enhancement of near-band-edge emission and suppression of the deep-level emission was demonstrated in ZnO by ion irradiation.^[16] Moreover, energetic ions are also overwhelmingly used in semiconductor technology to introduce electrical doping in a controlled fashion and thereby tuning the electronic properties.^[14]

Nowadays, the focused ion beam (FIB) technology has become mature and popular, especially in semiconductor industry, providing an enabling tool for surface modification, ion implantation, and integrated circuit repair. [14] Therefore, it is significant to gain knowledge about the impact of energetic ions on the photophysical properties for any emerging and potentially important materials in terms of both fundamental physics and practical applications.

Despite that tremendous efforts had been devoted to the research of CsPbX₃ perovskites, it remains unknown how the energetic ions will impact CsPbX₃, which, otherwise, may

contribute to the improved performance and new functionalities of CsPbX₃-based optoelectronic devices. [12, 17] In this work, we for the first time explore the interaction between the CsPbX₃ and high-energy gallium ions (Ga⁺ ions at 30 keV) in a broad range of ion doses provided by a commercial FIB system. We found that the optical properties of CsPbX₃ are highly sensitive to the energetic Ga⁺ ions due to the relatively vulnerable ionic bonding. Specifically, even low-dose Ga^+ irradiation ($\sim 1 \times 10^{15}$ ions/cm²) leads to more than one-orderof-magnitude reduction in the photoluminescence (PL) intensity from CsPbBr₃. According to the electronic microscopy, X-ray photoelectron spectroscopy and comprehensive optical characterizations, the PL quenching phenonenon can be attributed to the combined effects of the formation of vacancy/interstitial defects, generation of metallic Pb-related nonradiative recombination centers and crystal-to-amorphization transition. Interestingly, such spatially controllable PL fits for the microscale security protection applications and can be adopted to in-situ tune the laser emission from CsPbBr₃. With the increase of ion dose (~10¹⁷ ions/cm²), the morphology of CsPbX₃ can be dramatically altered due to the ion sputtering effect. By taking advantage of the mature FIB technique and optimizing the ion parameters, we demonstrate that the CsPbBr₃ crystals can be sculpted into precisely controlled shapes and sizes, enabling the top-down fabrication of the customizable and reproducible microlasers that are unattainable by the bottom-up synthetic routes. Our results offer essential information about the ion impacts on CsPbX3 and provide a practical tool to manipulate the emission and lasing from CsPbX₃, which could greatly push ahead the prospects of CsPbX₃ in photonic and optoelectronic applications.

The CsPbX₃ microcrystals with lateral dimensions of several tens to hundreds of micrometers and thickness of hundreds of nanometers were fabricated by the chemical vapor deposition method (see Experimental Section for detailed fabrication process).^[18] The elemental mapping analysis by the energy dispersive X-ray spectroscopy (EDX) coupled with scanning electron microscopy (SEM) on individual CsPbBr₃ microplates manifests the

uniform spatial distribution of Cs, Pb, and Br and confirms the formation of CsPbBr₃ compound (Figure S1, Supporting Information). The X-ray diffraction (XRD) measurment reveals the cubic phase of the CsPbBr₃ crystals (Figure S2, Supporting Information), which is consistent with previous reports.^[18, 19] The Helios 600 NanoLab FIB was employed to deliver the energetic gallium ions (Ga⁺ at 30 keV) and the sample was placed in the SEM/FIB dual beam system consisted of both electron- and ion-columns (Figure 1a). To test the impacts of the Ga⁺ ions on CsPbBr₃ microcrystals, we start with low dose ion irradiation on the sample in range of ~10¹⁴-10¹⁵ ions/cm². Figure 1b shows the SEM image of the FIB-treated CsPbBr₃ surface. The dark squares correspond to the FIB-treated area, where the ion dose gradually increases from $1.8 \times 10^{14} \text{ ions/cm}^2$ to $1.5 \times 10^{15} \text{ ions/cm}^2$ with equal intervals from 1 to 8 as labelled in Figure S3, Supporting Information, and the square becomes more and more darker with the increase of ion dose. The corresponding opitcal image of the FIB-treated sample is presented in Figure 1c, which does not show any observable change on the surface due to the low ion dose treatment. Interestingly, the fluoresent image under laser excitation (wavelength: 442 nm) clearly manifests the FIB-treated squares as being much less emissive than the pristine parts, indicating that the PL was quenched by the ion irradiation (Figure 1d). In order to quantitatively analyze the PL change in CsPbBr₃ by ion irradiation, we performed the iondose dependent steady-state PL measurements. As shown in Figure 2a, the PL intensity keeps decreasing with the increase of ion dose and the PL intensity with ion irradiation of $\sim 1 \times 10^{15}$ ions/cm² reduces by more than one order of magnitude (inset in Figure 2a). Simultaneously, the time-resolved PL measurement reveals that the PL decay rate rises with the increase of ion dose (Figure 2b), which suggests that more carrier trap defects are formed under higher ion dose irradiation. As the binding energy of the atoms in the CsPbBr₃ target sample (several eV)^[20] is much smaller than the kinetic energy of Ga⁺ ion at 30 keV, the atoms in CsPbBr₃ will be readily displaced from their lattice positions by the elastic core collision resulting from energy and momentum exchanges when the Ga+ ion beam is directed onto the target

sample.[14, 21] Moreover, the incident ions after collision and the target atoms knocked free from the lattice sites may still have enough energy to displace more target atoms, which in turn continue to dislocate yet other atoms. [14, 22] As a result, a "displacement cascade" is formed upon Ga+ ion irradiation, which generates plenty of defects including vacancies and interstitials, leading to the PL quenching phenomenon. In general, the penetration depth of the Ga⁺ ion beam lies in the range of several to tens of nanometers, so that the irradiation induced quenching effects primarily occurs on the surface of sample.^[14] To investigate the composition change upon Ga⁺ ion irradiation, we performed the surface-sensitive X-ray photoelectron spectroscopy (XPS) measurements on the CsPbBr₃ sample before and after FIB treatment. The presence of Cs, Pb and Br element is consistent with the material composition of CsPbBr₃ (Figure S4, Supporting Information). Whilst, upon the FIB treatment $(1.5 \times 10^{15}$ ions/cm²), new peaks locating near Pb 4f_{7/2} and Pb 4f_{5/2} peaks were observed (Figure 2c), indicating the formation of metallic Pb (Pb⁰).^[23, 24] The presence of Pb⁰ can effectively serve as the nonradiative recombination centers, [24] which may further contribute to the PL reduction. It is noted that we do not observe the signal related to the gallium element, indicating that gallium is not present above 0.1% concentration (the detection limit for XPS instrument) in the FIB-treated CsPbBr₃ surface.^[25]

We also performed the transmission electron microscopy (TEM) to access the change of crystal structure by the Ga⁺ ion irradiation. To do so, we scratched the CsPbBr₃ microcrystals and collected the tiny fragments onto the copper grids. Afterwards, the edge of the CsPbBr₃ fragment was treated by FIB (1.5×10¹⁵ ions/cm²) and then examined by TEM. As displayed in Figure 2e, clear lattice fringes can be observed for the CsPbBr₃ fragment before FIB treatment, indicating the high crystalline nature. Whlist, many nanometer-sized particles are formed and the lattice fringes disappear after the FIB treatment, indicating that the ion irradiation can induce the transition from crystal to amorphous phase.

As a result of the high energy of Ga⁺ ions, some of the target atoms can be ejected from the surface of the sample, the process is known as ion sputtering.^[14] Notably, the surface morphology of the CsPbBr₃ sample can be dramatically transformed due to ion sputtering effect. Figure 2d shows the atomic force microscopy (AFM) image of CsPbBr₃ after FIB treatment with relatively low ion dose (2.5×10¹⁴ ions/cm²). It is found that the concave square with depth of ~5 nm was formed. Deeper concave with microscale depth can be obtained by the higher-dose FIB treatment. Figure 1b displays the SEM image of the representative deep concave square fabricated by high-dose FIB treatment (1.25×10¹⁷ ions/cm²) as numbered "0". In the following part, we will show that the high-dose FIB treatment can be adopted as an enabling tool to sculpt the CsPbBr₃ into microlasers with on-demand resonators so as to overcome the limitations of the CsPbX₃-microlasers made from the bottom-up approach.

The ability to spatially control the emission from CsPbX₃ in a programmable manner allows the potential applications in security protection ranging from the information encryption to anticounterfeiting.^[26, 27] Quick response (QR) Code is a 2D matrix code that contains a high capacity as the dot-based 2D binary code scheme can provide a higher density of binary code information than a line-based 1D barcode scheme.^[27] As a proof-of-concept experiment, we designed a QR Code whose mirror image corresponds to the text "perovskite" and then duplicated it on CsPbBr₃ microcrystals by low-dose FIB treatment (see the SEM image in Figure 3g). Figure 3a shows the optical image of the CsPbBr₃ microcrystals under microscope and the QR Code is invisible due to small topographical modification (thickness contrast: <10 nm) of the surface by the low-dose ion treatment (2.5×10¹⁴ ions/cm²). It is worth noting that the realization of photopatterns without notably changing the morphology renders an ideal security-protection strategy. When the CsPbBr₃ microcrystals are exposed to the laser excitation, the QR Code emerges (Figure 3d). The size of the QR Code can be easily controlled by the FIB technique as demonstrated in Figs. 3d, e and f. By virtue of the high

photo-emissive property of CsPbBr₃, a feature that makes CsPbBr₃ attractive in light-emitting applications, and the favorable PL contrast induced by ion irradiation, the QR Code is readily observable by the cheap and commercially available lamps as demonstrated in Figures S5, where the QR is well decoded by a Xenon lamp (~10 mW/cm²) in combination with a band pass filter (390-420 nm). Moreover, leveraging on the facile emission wavelength tunability of CsPbX₃ by tailoring the composition, the blue and red QR Code (Figure S6, Supporting Information) can be made based on the CsPb(Cl/Br)₃ and CsPb(Br/I)₃, respectively.

Besides the QR Code, the text information can also be encoded by simply pre-defining it in the FIB system. As an example, the word "encrytion" was written on the CsPbBr₃ sample and the word was clearly resolved by the laser excitation (Figure 3h). It is well-known that FIB can reach nanometer spatial resolution, thus patterns or texts with subwavelength scale can be encrypted. Figure 3i shows the stripes fabricated by FIB treatment, it is found that the width of the stripe can be made as small as several tens of nanometers, making the subwavelength encryption feasible. Figure 3j shows the code "0101" in nanometer scale written on the sample (see Figure S7 for large-area SEM image, Supporting Information). Noting that the code with such tiny dimension cannot be distinguished by the common optical imaging system due to the diffraction limit. Nevertheless, the more advanced imaging techniques, such as the scanning near-field optical microscopy (SNOM), ^[28] may be feasible for the decryption.

In addtion to being excellent for spontaneous emission, CsPbX₃ perovskites have also been recognized as superior laser materials because of the large absorption coefficient and defect-tolerant nature.^[7, 29-32] Inspired by the effective PL quenching in CsPbX₃, we speculate that lasing emssion from CsPbX₃ could be manipulated by purposely introducing optical loss via selective FIB treatment, which would be not only interesting for fundamental research but also important for practical applications. To confirm the hyperthesis, we adopt the cubic CsPbBr₃ microplate which had been demonstrated to serve as favorable microlasers^[19, 33] as

the model system. Figure 4g shows the pump-fluence dependent micro-PL (µ-PL) spectra from the individual CsPbBr₃ microplate (see details of μ -PL measurement in Experimental Section). Under relatively low pump fluence (<~16.0 µJ cm⁻²), the PL is dominated by a broad emission band with full-width at half maximum (FWHM) of ~20 nm peaking at ~528 nm. With the increase of pump fluence, evenly-spaced spikes appear. The plot of the integrated intensity over the sharp peak spectral range as a function of the pump fluence exhibits a super-linear behavior (inset in Figure 4g), indicating the achievement of lasing action with low threshold (Pth) of ~16.0 μJ cm⁻². [34] The fluorescent image of CsPbBr₃ microplate exhibits nearly uniform green emission when the excitation fluence is below threshold (Figure 4a). While, a bright periphery of the microplate is observed as the pump fluence exceeds the threshold (Figure 4b), indicating the whispering gallery mode (WGM) lasing occurred in the perovskite microplates.^[35] In the cubic WGM cavity, the free spectral range (FSR) is given by: $FSR = \frac{\lambda^2}{2\sqrt{2}Ln_z}$, [34] where λ is the lasing wavelength and L is the edge length of the microplate. Accordingly, the group refractive index $n_{\rm g}$ is derived to be ~4.2, which agrees well with previous reports.^[19] In fact, it is not surprising to observe WGM lasing from the CsPbBr3 miroplates since the cubic microcrystal naturally behalves as the high-Q WGM resonator. [19, 33, 34] To demonstrate the feasibility of lasing manipulation by FIB, we partially treat the perovskite microplate by Ga^+ ion irradiation with low dose of $\sim 3\times 10^{15}$ ions/cm². As shown in Figure 4d, the fluorescent image of the perovskite microplate under low excitation fluence (~5 µJ cm⁻²) exhibits obvious PL quenching phenomenon in the FIBtreated area. The corresponding μ -PL spectra from the FIB-treated microplate are presented in Figure 4h. Similarly, the PL spectra are dominated by the broad spontaneous emission under relatively low pump fluence (< ~24 µJ cm⁻²). As pump fluence further increases, sharp spikes emerge, and the plot of the integrated PL intensity over pump fluence exhibits the threshold-

like behavior, which suggests the occurrence of lasing action in the FIB-treated CsPbBr₃ microplate. Notably, in sharp contrast to the pristine CsPbBr₃ microplate, only a pair of bright edges in the untreated area were observed for the FIB-treated microplate (Figure 4e), which indicates the development of Fabry–Pérot (F-P) lasing instead of WGM lasing. By taking the measured FSR of 1.33 nm into the formula describing F-P oscillation: $FSR = \frac{\lambda^2}{2Ln_g}$, the group refractive index is derived to be ~4.1, further confirming the F-P lasing mechanism in the FIB-treated microplate. In principle, as a result of the optical loss induced by the FIB treatment, the CsPbBr₃ microplate fails to provide sufficient optical gain for WGM oscillation, thereby leading to the transition from WGM to F-P lasing in the FIB-treated microplate. The corresponding resonant oscillations in the pristine and FIB-treated CsPbBr₃ microplate can be well retrieved by numerical simulation using finite element method (FEM) (see details in Experimental Secion) as shown in Figures 4c and 4f, respectively.

Till now, almost all of CsPbX₃-based microlasers are developed from the bottom-up method.^[19, 32] However, the bottom-up growth of the CsPbX₃ microcrystals lacks the controllability of the crystal size and shape.^[31] Thus, the lasing characteristics of one particular CsPbX₃-microlaser are difficult to duplicate in another, which fails to meet the demands of on-chip integrated photonic applications. Moreover, these as-grown microcrystals have limited morphologies, such as the triangles, squares, and hexagons,^[19, 31, 32] which restricts the diversity and potential of CsPbX₃-based microlasers. As shown above, the high-dose ion irradiation could dramatically alter the morphology of the pervoskites, such finding triggers us to explore the fabrication of CsPbX₃ microlasers by top-down approach via FIB treatment, aiming at overcoming the limitations of the bottom-up produced CsPbX₃ microlasers. Figure 5b shows the typical SEM image of the microdisk fabricated by high-dose FIB sputtering (1.25×10¹⁷ ions/cm²). By optimizing the parameters including current (22 pA), beam diameter (15 nm) and ion incident angle (90°), smooth surface of the perovskite

microdisks can be obtained (Figure S8, Supporting Information). It is known that the smooth surface can reduce the optical loss and enable the high-Q microcavities.^[36] Moreover, the vertical incidence of ions onto the sample can suppress the ion-induced optical loss on the surface, thus the bright green-color microdisk (Figure S9, Supporting Information) can be observed under low excitation fluence (~5 µJ cm⁻²). To check the lasing performance, we performed the pump-fluence dependent µ-PL measurement from the individual microdisk. Figure 5a shows the μ-PL spectra from a typical microdisk with diameter of 27.1 μm (Figure 5c for the SEM image). Again, the PL spectra are dominated by the broad spontaneous emission under low excitation intensities (< ~20 µJ cm⁻²). As the pump fluence keeps increasing, evenly-spaced sharp peaks appear, and the plot of the integrated PL intensity over pump fluence exhibits nonlinear behavior (inset in Figure 5a), indicating the development of lasing action in the CsPbBr3 microdisk. To get deeper insights, we carried out the timeresolved µ-PL measurements on the individual CsPbBr₃ microdisk with different excitation fluences (Figure 5i). It is seen that the spontaneous emission lifetime is ~460 ps at low pump fluence ($\sim 0.2 \text{ P}_{th}$). As the pump flence increase to $\sim 0.8 \text{ P}_{th}$, the lifetime decreases to $\sim 240 \text{ ps}$. Notably, when the pump fluence exceeds the threshold (~1.2 P_{th}), a much faster decay channel with lifetime of $< \sim 50$ ps (limited by the temporal response of the streak camera system) is revealed, which corresponds to the stimulated emission process and further confirms the occurrence of lasing action. To the best of our knowledge, it represents the first report of the laser emission from circular-shaped CsPbBr₃ microdisk.

The corresponding fluorescent image of the CsPbBr₃ microdisk above lasing threshold (Figure 5f) exhibits bright ring around the periphery, which suggests the WGM lasing mechanism. Hence, we tentatively examined the lasing peaks using WGM model. Considering the first radial mode order (q = 1), the resonant condition is given by: $m\lambda_m = \pi n_g D$, where m and λ_m are the angular mode number and the resonant wavelength at

m, respectively, and D is the diameter of the microdisk. Following the procedure reported by Ta et al., [35] the lasing peaks are found to match well with the mode numbers indexed as 756–763 (Figure S10a, Supporting Information). The Q-factor (calculated by $Q = \lambda/\Delta\lambda$, where λ and $\Delta\lambda$ are the lasing peak wavelength and the corresponding linewidth of the lasing peak, respectively) is derived to be as high as ~2700. The resonances in circular-shaped microdisks can be retrieved via numerical studies as shown in the inset of Figure 5h and the electric field is found to be effectively confined close to the inner boundary of CsPbBr₃ microdisk (Figure 5h), which is beneficial for the development of high-Q laser emission. [37]

By means of the mature FIB technology, the top-down fabricated perovskite microlasers are highly reproducible and controllable. Figure S11 shows the lasing spectra from another FIB-fabricated 27.1 μm-sized CsPbBr₃ microdisk. It is found that the lasing wavelength, *FSR* and Q-factor are nearly the same as those of the one shown in Figure 5a, indicating that the lasing characteristics can be duplicated by the FIB approach, which is essential for applications including on-chip integrated optoelectronics. The size of CsPbBr₃ microdisks can be readily manipulated by FIB method. Figures 5b, c and d display the SEM images of CsPbBr₃ microdisks with different sizes and the corresponding lasing images are shown in Figures 5e (Diameter: 30.2 μm), f (Diameter: 27.1 μm) and g (Diameter: 5.0 μm), respectively. The size dependent lasing spectra disclose that the *FSR* increases with the decrease of the microdisk diameter (Figure S10b, Supporting Information), which agrees with WGM lasing behavior.^[37, 38] Besides, other resonator shapes, such as the deformed CsPbBr₃ microdisk (Figure S12, Supporting Information), can also be fabricated and the lasing emission with nonuniform angular distribution is achieved, which provides an excellent platform to explore light-matter interaction in customizable laser resonators.

In summary, we investigated the interaction between $CsPbX_3$ and high-energy gallium ions (Ga⁺ ions at 30 keV) in a broad range of ion doses for the first time. We revealed that both low-dose (~10¹⁴ ions/cm²) and high-dose (~10¹⁷ ions/cm²) ion irradiations can serve as

effective tools to manipulate the light emission and lasing from $CsPbX_3$ and the underlying physical mechanisms are provided. Our results represent a significant step towards applying the emerging $CsPbX_3$ for photonic applications.

Experimental Section

Synthesis of CsPbX₃ mcirocrystals: The CsPbX₃ microcrystals were fabricated by the chemical vapor deposition method. The ITO-glass substrate was placed inside the downstream of quartz tube in Ar atmosphere. The precusor was prepared by mixing powder of PbX₂ and CsBr with molar ratio of 1:1. The reaction temperature was set at 575 °C and hold for 15 min. The CsPbX₃ microcrystals with lateral dimensions of several tens to hundreds of micrometers and thickness of hundreds of nanometers can be fabricated.

FIB treating on CsPbX₃ microcrystals: The Helios 600 NanoLab FIB was adopted to deliver gallium ions (Ga⁺ at 30 keV) and the CsPbX₃ sample was placed in the SEM/FIB dual beam system that consisted of both electron- and ion-columns. The stage has five degrees of freedom for movement (X, Y, Z, rotation, tilt). The currents used were 22 pA.

Optical characterization: For the μ-PL measurement, the sample was pumped at 400 nm by frequency-doubling of the fundamental wavelength (800 nm) from the Ti: sapphire regenerative amplifier with pulsewidth of 100 fs. The laser beam is focused to cover the whole sample by a microscope objective and the PL signal is collected by the same objective and then analyzed by a charge coupled device (CCD). For the time-resolved PL, the same laser source is adopted to excited the sample and the PL signal is recorded by an Optronis streak camera with an optimized temporal resolution of ~50 ps.

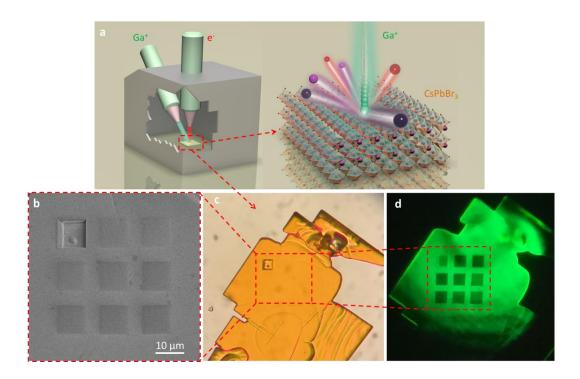


Figure 1. a) Schematic illustration of FIB treatment on CsPbBr₃ microcrystals. b) SEM image of the FIB-treated CsPbBr₃ microcrystal. The dark squares correspond to the FIB treated area. c) Fluorescent image of the FIB-treated CsPbBr₃ microcrystal. The dark squares correspond to the FIB treated area.

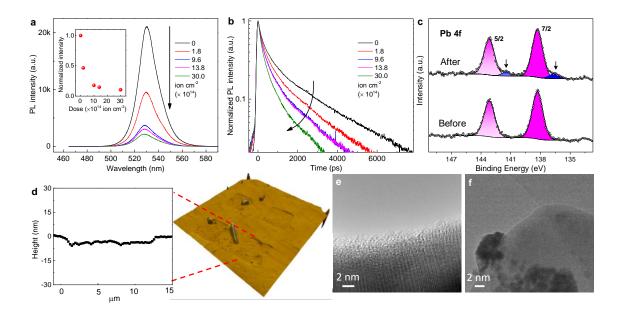


Figure 2. a) PL spectra from CsPbBr₃ with different ion dose irradiation. The inset shows the corresponding ion dose dependent integrated PL intensity from CsPbBr₃. b) PL decay from CsPbBr₃ with different ion dose irradiation. c) XPS signals corresponding to Pb 4f before and after FIB treatment. d) Atomic force microscopy (AFM) image and the corresponding cross-section line of CsPbBr₃ surface with FIB treatment. e) and f) Transmission electron microscopy (TEM) of CsPbBr₃ fragment before and after FIB treatment.

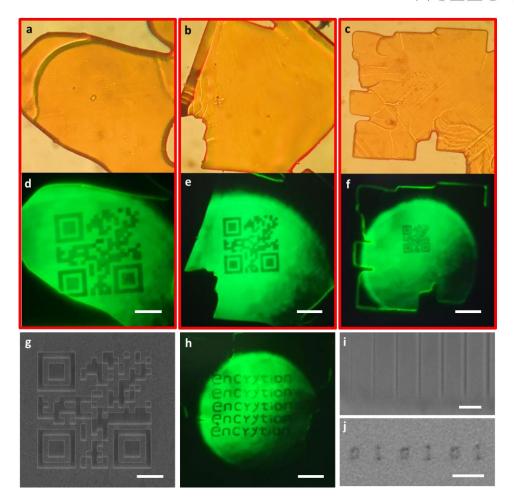


Figure 3. a-c) Optical images of the FIB-treated CsPbBr₃ mcirocrystal with QR Code encoding. d-f) Fluorecent images of the FIB-treated CsPbBr₃ mcirocrystal, revealing the encoded QR Code. Scale bar: 20 μm. g) SEM image of QR Code encoded on CsPbBr₃ by FIB-treatment. Scale bar: 5 μm. h) Fluorecent image of the the FIB-treated CsPbBr₃ mcirocrystal, revealing the encoded text information. Scale bar: 20 μm. i) SEM image of stripes made by FIB treatment. Scale bar: 2 μm. j) Nanometer sized charaters (0101) encoded on CsPbBr₃ by FIB treatment. Scale bar: 1 μm.

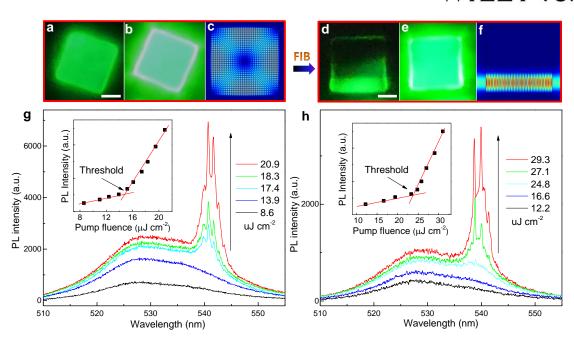


Figure 4. a) Below-threshold fluorescent image of the CsPbBr₃ microplate. Scale bar: 10 μm. b) Above-threshold fluorescent image of the CsPbBr₃ microplate. c) Simulated electric field distribution inside the square perovskite cavity. d) Below-threshold fluorescent image of the FIB-treated CsPbBr₃ microplate. Scale bar: 10 μm. e) Above-threshold fluorescent image of the FIB-treated CsPbBr₃ microplate. f) Simulated electric field distribution inside the FIB-treated perovskite cavity. g) Pump-fluence dependent PL spectra from the CsPbBr₃ microplate. The inset shows the integrated PL intensity over the sharp peak spectral range as a function of pump fluence. h) Pump-fluence dependent PL spectra from the FIB-treated CsPbBr₃ microplate. The inset shows the integrated PL intensity over the sharp peak spectral range as a function of pump fluence.

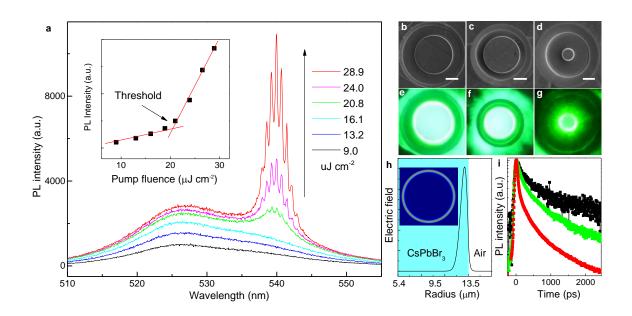


Figure 5. a) Pump-fluence dependent PL spectra from individual CsPbBr₃ microdisk with diameter of 27.1 μm. The inset shows the integrated PL intensity over the sharp peak spectral range as a function of pump fluence. b-d) SEM images of CsPbBr₃ microdisks with diameter of b) 30.2 μm, c) 27.1 μm and d) 5.0 μm. Scale bar: b) 10 μm, c) 10 μm, d) 5 μm. The corresponding above-threshold fluorescent images of the microdisks are shown in e-g). h) 2D plot of the electric field distribution near the edge of the mcirodisk. The inset shows the electric field distribution inside the microdisk. i) Pump fluence dependent PL decay from individual CsPbBr₃ microdisk with diameter of 27.1 μm.

References

- [1] B.R. Sutherland, E.H. Sargent, *Nature Photonics*, **2016** *10* 295.
- [2] X. Yang, X. Zhang, J. Deng, Z. Chu, Q. Jiang, J. Meng, P. Wang, L. Zhang, Z. Yin, J. You, *Nature Communications*, **2018** 9 570.
- [3] W. Zhang, L. Peng, J. Liu, A. Tang, J.-S. Hu, J. Yao, Y.S. Zhao, *Advanced Materials*, **2016** 28 4040-4046.
- [4] G. Xing, N. Mathews, S.S. Lim, N. Yantara, X. Liu, D. Sabba, M. Grätzel, S. Mhaisalkar, T.C. Sum, *Nature Materials*, **2014** *13* 476.
- [5] N. Zhang, W. Sun, S.P. Rodrigues, K. Wang, Z. Gu, S. Wang, W. Cai, S. Xiao, Q. Song, *Advanced Materials*, **2017** 29 1606205.
- [6] M.V. Kovalenko, L. Protesescu, M.I. Bodnarchuk, Science, 2017 358 745-750.
- [7] Y. Wang, H. Sun, Small Methods, 2018 2 1700252.
- [8] Y. Wang, X. Li, J. Song, L. Xiao, H. Zeng, H. Sun, Advanced Materials, 2015 27 7101-7108.
- [9] K. Lin, J. Xing, L.N. Quan, F.P.G. de Arquer, X. Gong, J. Lu, L. Xie, W. Zhao, D. Zhang, C. Yan,
- W. Li, X. Liu, Y. Lu, J. Kirman, E.H. Sargent, Q. Xiong, Z. Wei, *Nature*, 2018 562 245-248.
- [10] A. Swarnkar, A.R. Marshall, E.M. Sanehira, B.D. Chernomordik, D.T. Moore, J.A. Christians, T. Chakrabarti, J.M. Luther, *Science*, **2016** *354* 92-95.
- [11] Y. He, L. Matei, H.J. Jung, K.M. McCall, M. Chen, C.C. Stoumpos, Z. Liu, J.A. Peters, D.Y. Chung, B.W. Wessels, M.R. Wasielewski, V.P. Dravid, A. Burger, M.G. Kanatzidis, *Nature Communications*, **2018** 9 1609.
- [12] Q. Chen, J. Wu, X. Ou, B. Huang, J. Almutlaq, A.A. Zhumekenov, X. Guan, S. Han, L. Liang, Z. Yi, J. Li, X. Xie, Y. Wang, Y. Li, D. Fan, D.B.L. Teh, A.H. All, O.F. Mohammed, O.M. Bakr, T. Wu, M. Bettinelli, H. Yang, W. Huang, X. Liu, *Nature*, **2018** *561* 88-93.
- [13] J.H. Heo, D.H. Shin, J.K. Park, D.H. Kim, S.J. Lee, S.H. Im, *Advanced Materials*, **2018** *30* 1801743.
- [14] N.M. M Nastasi, J Mayer, JK Hirvonen, M James Cambridge Univ. Press, Cambridge, 1996.
- [15] R.M. Papaléo, R. Thomaz, L.I. Gutierres, V.M. de Menezes, D. Severin, C. Trautmann, D. Tramontina, E.M. Bringa, P.L. Grande, *Physical Review Letters*, **2015** *114* 118302.
- [16] B. Yadian, R. Chen, H. Liu, H. Sun, Q. Liu, C.L. Gan, Z. Kun, C. Zhao, B. Zhu, Y. Huang, *Nano Research*, **2015** 8 1857-1864.
- [17] Y. Wang, X. Li, V. Nalla, H. Zeng, H. Sun, *Advanced Functional Materials*, **2017** 27 1605088-n/a.
- [18] C. Huo, X. Liu, X. Song, Z. Wang, H. Zeng, *The Journal of Physical Chemistry Letters*, **2017** 8 4785-4792.
- [19] Q. Zhang, R. Su, X. Liu, J. Xing, T.C. Sum, Q. Xiong, *Advanced Functional Materials*, **2016** 26 6238-6245.
- [20] A. Varadwaj, P.R. Varadwaj, K. Yamashita, ChemSusChem, 2018 11 449-463.
- [21] J. Gierak, Nanofabrication 2014.
- [22] G.V. Voznyuk, I.V. Levitskii, M.I. Mitrofanov, D.N. Nikolaev, V.P. Evtikhiev, *Journal of Physics: Conference Series*, **2018** *1038* 012080.
- [23] L. Zhang, X. Yang, Q. Jiang, P. Wang, Z. Yin, X. Zhang, H. Tan, Y. Yang, M. Wei, B.R. Sutherland, E.H. Sargent, J. You, *Nature Communications*, **2017** 8 15640.
- [24] W. Zhang, S. Pathak, N. Sakai, T. Stergiopoulos, P.K. Nayak, N.K. Noel, A.A. Haghighirad, V.M. Burlakov, D.W. deQuilettes, A. Sadhanala, W. Li, L. Wang, D.S. Ginger, R.H. Friend, H.J. Snaith, *Nature Communications*, **2015** 6 10030.
- [25] Y. Ren, K. Yuan, X. Zhou, H. Sun, K. Wu, S.L. Bernasek, W. Chen, G.Q. Xu, *Chemistry A European Journal*, 0.
- [26] Y. Wang, X. Li, S. Sreejith, F. Cao, Z. Wang, M.C. Stuparu, H. Zeng, H. Sun, *Advanced Materials*, **2016** 28 10637-10643.
- [27] R. Arppe, T.J. Sørensen, Nature Reviews Chemistry, 2017 1 0031.
- [28] E. Betzig, R.J. Chichester, Science, 1993 262 1422-1425.
- [29] Y. Wang, X. Li, X. Zhao, L. Xiao, H. Zeng, H. Sun, Nano Letters, 2016 16 448-453.
- [30] Y. Wang, D. Yu, Z. Wang, X. Li, X. Chen, V. Nalla, H. Zeng, H. Sun, *Small*, **2017** *13* 1701587-n/a.

- [31] H. Zhou, S. Yuan, X. Wang, T. Xu, X. Wang, H. Li, W. Zheng, P. Fan, Y. Li, L. Sun, A. Pan, *ACS Nano*, **2017** *11* 1189-1195.
- [32] S.W. Eaton, M. Lai, N.A. Gibson, A.B. Wong, L. Dou, J. Ma, L.-W. Wang, S.R. Leone, P. Yang, *Proceedings of the National Academy of Sciences*, **2016** *113* 1993-1998.
- [33] X. He, P. Liu, H. Zhang, Q. Liao, J. Yao, H. Fu, Advanced Materials, 2017 29 1604510.
- [34] A.K. Bhowmik, Appl. Opt., 2000 39 3071-3075.
- [35] V.D. Ta, R. Chen, H.D. Sun, Advanced Materials, 2012 24 OP60-OP64.
- [36] S.-J. Tang, S. Liu, X.-C. Yu, Q. Song, Q. Gong, Y.-F. Xiao, *Advanced Materials*, **2018** *30* 1800262.
- [37] Y. Wang, K.S. Leck, V.D. Ta, R. Chen, V. Nalla, Y. Gao, T. He, H.V. Demir, H. Sun, *Advanced Materials*, **2015** 27 169-175.
- [38] Y. Wang, K.E. Fong, S. Yang, Van D. Ta, Y. Gao, Z. Wang, V. Nalla, H.V. Demir, H. Sun, *Laser & Photonics Reviews*, **2015** *9* 507-516.