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Title: Impact of Salt Accumulation in the Bioreactor on the Performance of Nanofiltration Membrane Bioreactor (NF-MBR)+Reverse Osmosis (RO) Process for Water Reclamation

Article Type: Research Paper

Keywords: Membrane bioreactor; Membrane fouling; Nanofiltration; Reverse Osmosis; Salt accumulation; Water reclamation

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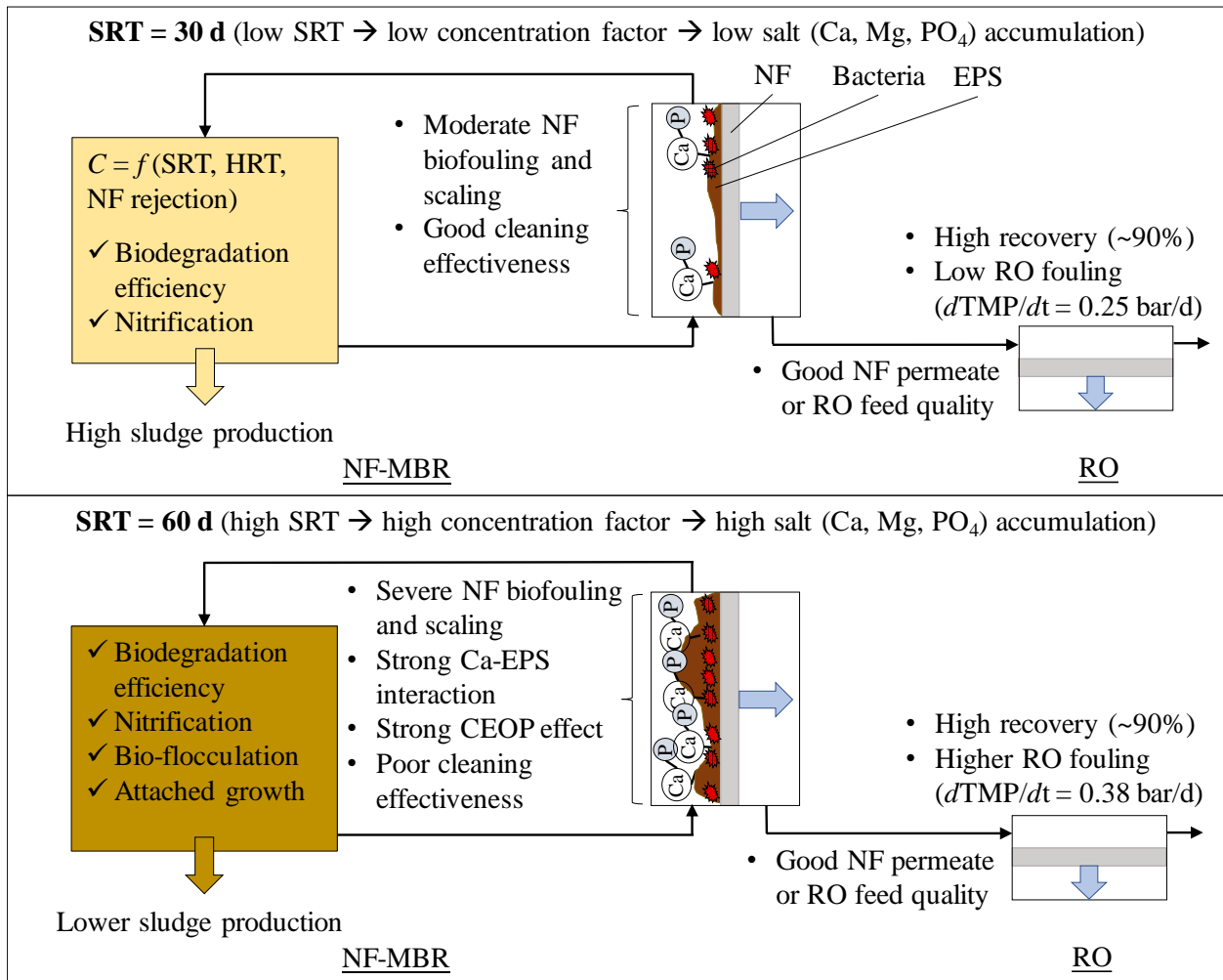
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Abstract: The impacts of salt accumulation, through adjusting the solid retention time (SRT), in the bioreactor on the bioprocess as well as membrane performance of a high retention nanofiltration membrane bioreactor (NF-MBR) and subsequent reverse osmosis (RO) process for water reclamation are addressed in this study. The build-up of salts (i.e., Ca, Mg, PO₄) is a function of SRT, hydraulic retention time (HRT) and membrane rejection. Despite the accumulation of salts, both NF-MBRs at SRT of 30 and 60 d, achieved (i) similar biodegradation efficiency; (ii) excellent organic removal (>97%); and (iii) ammonia removal (>98%). Extending the SRT could improve the microbial bio-flocculation capability, but did not influence the microbial activity, viability, and community structure. However, more severe membrane fouling was observed in the NF-MBR with elevated salt levels, which was attributed to the greater formation of calcium phosphate scale and Ca-polysaccharides complex (i.e., irreversible fouling layer) as well as the cake-enhanced-osmotic-pressure (CEOP) effect. Although both NF-MBRs produced comparable quality of permeate, a higher RO membrane fouling rate was observed when permeate of NF-MBR with SRT at 60 d was fed to the RO system, implying organic compositions in NF-MBR permeate may influence RO performance.

Highlights:

- Salt accumulation in the NF-MBR had no adverse impact on the bioprocess.
- The elevated divalent ions level promoted bio-flocculation.
- Longer SRT caused severe Ca-polysaccharides combined fouling in NF-MBR.
- Integration of NF-MBR with RO process for high recovery water reclamation.

Graphical Abstract:



1 **Impact of Salt Accumulation in the Bioreactor on the Performance of Nanofiltration**
2 **Membrane Bioreactor (NF-MBR)+Reverse Osmosis (RO) Process for Water Reclamation**

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24 **Abstract**

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26 bioreactor on the bioprocess as well as membrane performance of a high retention nanofiltration
27 membrane bioreactor (NF-MBR) and subsequent reverse osmosis (RO) process for water
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29 SRT, hydraulic retention time (HRT) and membrane rejection. Despite the accumulation of salts,
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36 cake-enhanced-osmotic-pressure (CEOP) effect. Although both NF-MBRs produced comparable
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38 MBR with SRT at 60 d was fed to the RO system, implying organic compositions in NF-MBR
39 permeate may influence RO performance.

40 **Keywords:** Membrane bioreactor; Membrane fouling; Nanofiltration; Reverse Osmosis; Salt
41 accumulation; Water reclamation

42 **Abbreviations**

43 CEOP Cake enhanced osmotic pressure
44 CF Concentration factor
45 CLSM Confocal laser scanning microscopy

46	COD	Chemical oxygen demand
47	tCOD	Total chemical oxygen demand
48	sCOD	Soluble chemical oxygen demand
49	DOC	Dissolved organic carbon
50	EPS	Extracellular polymeric substances
51	F/M	Food to microorganism ratio
52	HR-MBR	High retention membrane bioreactor
53	HRT	Hydraulic retention time
54	LC-OCD	Liquid chromatography – organic carbon detection
55	LMW	Low molecular weight
56	MBR	Membrane bioreactor
57	MLSS	Mixed liquor suspended solids
58	MLVSS	Mixed liquor volatile suspended solids
59	MWCO	Molecular weight cut-off
60	NF-MBR	Nanofiltration membrane bioreactor
61	SMP	Soluble microbial products
62	SOUR	Specific oxygen uptake rate
63	SRT	Solids retention time
64	TDS	Total dissolved solids
65	TOC	Total organic carbon

66

67 **Symbols**

68	C_f	Feed salt concentration
69	C_{ml}	Mixed liquor salt concentration
70	η_{NF}	Nanofiltration membrane rejection rate

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74 **1. Introduction**

75 Recently, high retention membrane bioreactors (HR-MBRs), integrating a biological
76 process with a high rejection membrane such as nanofiltration (NF), forward osmosis (FO), and
77 membrane distillation (MD), have received significant attention in the field of wastewater
78 reclamation due to their improved permeate quality as compared to conventional microfiltration
79 (MF) or ultrafiltration (UF) based MBR (Lay et al. 2010, Luo et al. 2014). In MD-MBR, clean
80 water can be achieved in a single step while additional step is required to recover the clean water
81 from the draw solution in FO-MBR. Whereas in NF-MBR, with the improved MBR permeate as
82 RO feed, the subsequent RO process can be potentially operated at higher recovery rate due to
83 less RO fouling (Tay et al. 2018). In HR-MBRs, the high rejection membranes effectively retain
84 the smaller size and refractory organic substances, leading to enhanced biodegradation
85 (improved removal rates). However, the major disadvantage is the retention that leads to the
86 accumulation of dissolved ions in the bioreactor by the high rejection membranes (Qiu and Ting
87 2013). It is noted that the concentration factor, CF, defined as the ratio of concentration of mixed
88 liquor in the bioreactor to the feed concentration, is related to the sludge retention time (SRT) and
89 hydraulic retention time (HRT), in which the maximum value, $CF_{max} = SRT/HRT$ (Lay et al.
90 2010). Thus, SRT is an important parameter in determining salt accumulation ratio in HR-MBRs.

91 In conventional MF/UF-MBR, extending the SRT generally promoted lower soluble
92 microbial products (SMP), lower specific bound EPS production, better bio-flocculation, and
93 formation of a more permeable cake layer, therefore, resulting in less membrane fouling
94 (Iorhemen et al. 2017, Massé et al. 2006, Ng et al. 2006, Van den Broeck et al. 2012, Wu et al.
95 2011). It is noted that longer SRT was often adopted to maintain high biomass concentrations
96 (lower F/M ratio), reduce solids production, and minimize reactor volume (Pollice et al. 2004,

97 Xiao et al. 2019). In addition, longer SRT was preferred for more effective nitrification since the
98 autotrophic nitrifiers are very slow-growing microorganisms among other aerobic bacteria (Ge et
99 al. 2015).

100 On the other hand, in HR-MBRs, extending the SRT essentially means increasing the salt
101 accumulation, which can potentially impact the physical-chemical properties of the feed solution,
102 bioprocess, and membrane operation (Lay et al. 2010, Luo et al. 2014). For example, when
103 salinity increases, solubility of oxygen decreases linearly, which leads to lower oxygen transfer
104 rate (Colt 2012). This substantially reduces the effectiveness of aeration, which is translated into
105 higher energy requirement to achieve the desired DO level. Regarding the impact of salinity on
106 bioprocess, in both FO-MBR and MD-MBR, it has been found that elevated salinity could lead
107 to a shift in microbial community, lower growth yield and lower microbial kinetics of the
108 activated sludge necessary for the biological treatment and eventually resulted in process
109 deterioration (Luján-Facundo et al. 2018, Nawaz et al. 2013, Wijekoon et al. 2014). In term of
110 impact of salinity on membrane operation, in general, higher feed salinity means reduced net
111 driving force due to higher osmotic pressure difference and greater transport of solutes across the
112 membrane that lower the permeate flux and quality. In addition, accumulation of dissolved ions
113 such as Ca^{2+} , Mg^{2+} , SO_4^{2-} , PO_4^{3-} , etc. can increase the risk of membrane scaling and aggravate
114 colloidal fouling due to higher supersaturation index. Furthermore, the increase in extracellular
115 polymeric substances (EPS) production under high salinity environment as a protective
116 mechanism under salinity stress was found to increase the fouling propensity in FO-MBR (Qiu
117 and Ting 2013).

118 It shall be noted that most of the work reported on impacts of salt accumulation were
119 related to FO-MBR and MD-MBR, the in-depth study to address this issue for NF-MBR

120 application was not available. Indeed, research work on NF-MBR in the literature was rather
121 limited, possibly associated with the poor permeability of NF membranes, typical flux was less
122 than 5 L/m²h (Choi et al. 2006, Choi et al. 2007, Phan et al. 2016, Zaviska et al. 2013), except in
123 our previous work where flux of 10 L/m²h was attained with the use of a low pressure in-house
124 fabricated NF membrane (Tay et al. 2018). Therefore, it is unclear whether such observations
125 found in FO-MBR and MD-MBR are applicable to NF-MBR. For instance, the driving force and
126 membrane properties such as structure, permeability and rejection of FO, MD, and NF
127 membranes differ from each other, which could have imposed different effects arisen from salt
128 accumulation in the bioreactors on the bioprocess and membrane operation. Thus, in this study,
129 the impacts of salt accumulation, through the manipulation of SRT, on microbial behaviors,
130 bioreactor performance, permeate quality and membrane fouling of NF-MBR as well as the
131 subsequent RO performance for water reclamation were investigated.

132 **2. Materials and methods**

133 2.1. NF-MBR setup and operating conditions

134 Two parallel lab-scale moving bed biofilm reactors employing side-stream NF membrane
135 modules were used in this work. The schematic diagram of the NF-MBR is illustrated in Figure
136 S1 (Appendix). The NF-MBRs were operated at SRT of 30 and 60 days (hereinafter denoted as
137 NF-MBR_30-LS and NF-MBR_60-HS), respectively. Each bioreactor has a total effective
138 working volume of 7.0 L and filled with 234 K3 bio-carriers (~20% of packing ratio). The
139 dimension of bio-carrier is 12 mm × Ø 25 mm with surface area of 500 m²/m³ (Ping Yin
140 Chemical Packing, China). The biofilm reactor was inoculated with aerobic activated sludge
141 collected from a municipal wastewater treatment plant in Singapore at a mixed liquor suspended

142 solid (MLSS) concentration of 1 g/L. The aeration rate was set at 500 mL/min to maintain
 143 dissolved oxygen (DO) level above 4 mg/L. The hydraulic retention time (HRT), pH, and
 144 temperature were kept at 22 h, 6.8±0.5, and 20±0.5°C, respectively. The permeate flux of NF-
 145 MBR was maintained constant throughout the study at 10 L/m²h via a mass flow controller
 146 installed on the permeate stream. The feed, retentate and permeate pressures were recorded by
 147 digital pressure transducers (Ashcroft, USA) connected to a data logging system (LabVIEW,
 148 National Instrument, USA). The municipal wastewater (after sieving with a 1 mm opening mesh)
 149 was collected weekly and stored in cool room at 4°C prior use as feed to NF-MBRs. The
 150 operating parameters of the NF-MBRs and the characteristics of the municipal wastewater are
 151 summarized in Tables 1 and 2, respectively.

152 **Table 1. Operating parameters of NF-MBRs**

Parameter	NF-MBR_30-LS	NF-MBR_60-HS
HRT (h)	22	22
SRT (d)	30	60
Flux (L/m ² h)	10	10
Organic loading rate (g Eq/L.day)	0.41	0.41
F/M ratio (g COD/g MLSS.d)	0.23	0.22
NF membrane crossflow velocity (m/s)	0.10	0.10
pH	6.2 – 6.9	6.2 – 6.9
Aeration (mL/min)	500	500
DO level (mg/L)	4.3-5.0	4.2-4.9
Biocarrier filling ratio (%)	20	20

153
 154 **Table 2. Characteristics of municipal wastewater (n=21)**

Parameter	Level
tCOD (mg/L)	378.6 ± 116.5
sCOD(mg/L)	120.5 ± 28.3
DOC (mg/L)	55.3 ± 27.7
Ca ²⁺ (mg/L)	26.1 ± 2.6
Mg ²⁺ (mg/L)	10.4 ± 2.9
Na ⁺ (mg/L)	99.0 ± 17.5
Fe ³⁺ (mg/L)	0.45 ± 0.18
TN (mg/L)	39.7 ± 7.5

NH ₄ -N (mg/L)	39.3 ± 5.2
NO ₃ ⁻ -N (mg/L)	0.4 ± 0.1
NO ₂ ⁻ -N (mg/L)	N.D.
PO ₄ ³⁻ (mg/L)	15.8 ± 2.6
TDS (mg/L)	666.4 ± 72.1
Conductivity (µs/cm)	970.7 ± 87.3

Note: tCOD: total chemical oxygen demand; sCOD: soluble chemical oxygen demand;
 DOC: dissolved organic carbon; TN: total nitrogen; N.D.: not detectable.

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158 In this study, the in-house fabricated glutaraldehyde cross-linked layer-by-layer
 159 polyelectrolyte hollow fiber NF membranes with polyethersulfone (PES) UF as substrate were
 160 used and their characteristics are summarized in Table S1 (Appendix). The details of fabrication
 161 of this NF membrane can be found in the previous publication (Liu et al. 2015). Unlike most
 162 commercial NF membranes with high rejection (>90%) of divalent ions and partial rejection (i.e.,
 163 30 - 90%) of monovalent ions, this NF membrane does not reject monovalent ions while
 164 maintains high rejection of divalent ions, thus has low osmotic pressure difference across the
 165 membrane which allows low pressure operation, i.e., < 2 bar. Each membrane module consists of
 166 40 (pieces of) hollow fibers (0.34 m in length with a total effective area of 0.031 m²). The
 167 filtration was performed in an inside (lumen)-out (shell) configuration. Periodic physical
 168 cleaning (every two days) of membrane was performed in order to maintain constant permeate
 169 flux at 10 L/m²h throughout the test. In detail, the fouled NF module was first removed from the
 170 NF-MBR and physical cleaning was conducted in a separate batch setup according to the
 171 following steps, i) flushing with tap water at 500 or 800 mL/min (equivalent to crossflow
 172 velocity of 0.5 or 0.8 m/s) for 30 s to remove clogged particles inside the lumen; ii) backwashing
 173 with ultrapure water at pressure of 2 or 3 bar for 15 min to detach the cake layer on the
 174 membrane surface; iii) flushing with tap water at 500 or 800 mL/min for 3 min to remove the
 175 residual foulants inside the lumen. After each physical cleaning cycle, the recovery of membrane
 176 permeability was checked by pure water flux measurement.

177 During first 30 days of operation, the mixed liquor in the bioreactor was directly fed to
178 the NF membrane, denoted as NF1, which caused severe membrane fouling. Then, a sponge
179 column (sponge pore size of ~ 0.5 mm, refer to Figure S1 in Appendix) was installed at the
180 effluent outlet of the bioreactor before the NF module, aiming to retain the larger sized particles
181 of >0.5 mm; a new NF membrane (NF2) was used in the subsequent test.

182 2.2. RO setup and operating conditions

183 Two identical RO systems were used to study the fouling potential of permeates from
184 both NF-MBRs. The RO system consists of a feed tank, a high-pressure pump (Hydra Cell D-03-
185 S, USA), a cooling device equipped with temperature controller (PolyScience 9106, USA), a flat
186 sheet RO test cell, a back pressure regulator (SS-4R3A, Swagelok), a mass flow controller
187 (LIQUI-FLOW L30, Bronkhorst, Netherlands), pressure transducers (EUTECH Instruments,
188 USA), and conductivity meters (EUTECH Instruments, USA). The details of RO system were
189 described previously (Suwarno et al. 2016, Tay et al. 2018, Wu et al. 2013). RO membrane (BW-
190 30, DOW FilmTec, USA) with an effective membrane area of 0.0186 m^2 was employed in the
191 RO test cell.

192 The RO membrane was first compacted with ultrapure water at flux of $50 \text{ L/m}^2\text{h}$ and
193 crossflow velocity of 0.10 m/s for 1 h prior fouling test. The RO feed water in the RO fouling
194 test was the permeate collected from the NF-MBRs during Day 60-62. The collected NF-MBR
195 permeate was first concentrated using the RO by discharging the RO permeate until its
196 concentration equivalent to 90% recovery ratio, i.e., $\times 10$ original concentration, was attained.
197 Subsequently, the RO fouling test was performed with the concentrated test solution at flux of 20
198 $\text{L/m}^2\text{h}$ and crossflow velocity of 0.10 m/s . Full recycle of retentate and permeate to the feed tank

199 was adopted to simulate the RO fouling at high recovery of 90%. The pH of test solution was
200 maintained at pH 6.3 by acid dosing. The RO fouling test lasted for 5 days.

201 2.3. Analytical measurements

202 2.3.1. *Water quality analysis*

203 The feed water, mixed liquor and permeate of the NF-MBRs were sampled twice weekly
204 for analysis. The supernatant of the feed water and mixed liquor was obtained by centrifuging for
205 10 min at 4000 ×g and then filtering through a 0.45 μm membrane filter before analysis. Total
206 organic carbon (TOC) and total nitrogen (TN) were determined by a TOC/TN-V analyzer
207 (Shimadzu, Japan). Chemical oxygen demand (COD), ammonia, nitrite, nitrate and total
208 phosphorus were measured using the colorimetric method with a spectrophotometer (DR 3900,
209 HACH, USA). Concentrations of cations were determined by an inductively coupled plasma
210 optical emission spectrometry (ICP-OES, Optima 8000, Perkin Elmer, USA). The conductivity
211 and total dissolved solids (TDS) were measured using a multi-function meter (Eutech
212 Instruments, USA). Liquid chromatography-organic carbon detector (LC-OCD) analyzer (LC-
213 OCD Model 8, DOC-LABOR, Germany), a size-exclusion chromatography system coupled with
214 organic carbon detection and organic nitrogen detection was used to quantify the dissolved
215 organic fractions according to their molecular weights (MWs): biopolymers (typically MW > 10
216 kDa), humic substances and building blocks (HS&BB, MW 350 – 1000 Da), low molecular
217 weight organics (LMW, MW < 350 Da). Details of the LC-OCD system and analysis were
218 described in the literature (Huber et al. 2011).

219 2.3.2. *Biological analysis*

220 The total solid amount in the bioreactor was defined as the sum of mixed liquor
221 suspended solids (MLSS) and total attached solids on the bio-carriers. Similarly, the total
222 biomass amount in the bioreactor was defined as the sum of mixed liquor volatile suspended
223 solids (MLVSS) and total attached biomass on the bio-carriers. The attached solids/biomass on
224 the bio-carrier was extracted by sonication (10 min) and vortex (1 min). The total attached
225 solids/biomass was estimated by multiplying the attached solids/biomass amount extracted per
226 bio-carrier (i.e., 2 pieces of biocarriers were sampled for analysis each time) with the total
227 number of bio-carriers in the bioreactor. The total solids/biomass concentration was calculated
228 by dividing the total solids/biomass amount by the effective reactor volume. The MLSS and
229 MLVSS were measured according to Standard Methods (American Public Health et al. 2005).
230 Particle size of the bioflocs was measured by a Mastersizer (MS2000, Malvern, UK). Zeta
231 potential of the flocs was analyzed by a Zetasizer (ZEN3600, Malvern, UK).

232 The intracellular ATP (i.e., live cells) and extracellular ATP (i.e., dead cells)
233 concentrations were measured according to manufacturer's specifications (Kikkoman, Japan).
234 The viability of the activated sludge was determined by the ratio of intracellular ATP
235 concentrations to the total ATP concentrations. The microbial activity was assessed by
236 measuring the specific oxygen uptake rate (SOUR) of the activated sludge according to Standard
237 Method 1683 (American Public Health et al. 2005).

238 The separation of bound EPS and soluble EPS in the mixed liquor and foulant solution
239 was done by centrifuging the solution at 4000 $\times g$ for 10 min at 4 °C. The supernatant was
240 collected for soluble EPS measurement. The bound EPS was extracted by formaldehyde–NaOH
241 method (Liu and Fang 2002, Zhang et al. 2006). Briefly, the solid fraction was re-suspended in

242 10 mL of ultrapure water, followed by adding 12 μ L of formaldehyde (36.5%; Sigma Aldrich,
243 USA) and keeping for 1 h at 4 °C. After that, 0.8 mL of 1M NaOH was added and kept for 3 h at
244 4 °C. Subsequently, the solution was centrifuged at 13200 \times g for 20 min and the supernatant was
245 collected for bound EPS measurement. The EPS contents were determined by measurement of
246 polysaccharides and proteins concentrations. The polysaccharides concentration was determined
247 according to the phenol-sulfuric acid method with glucose as a standard and absorbance
248 measurement at 490 nm using a spectrometer (HACH, USA) (Dubois et al., 1956). The protein
249 concentration was determined by Bradford method with bovine serum albumin (BSA) as
250 standard and absorbance measurement at 595 nm (Bradford 1976).

251 To examine the microbial community compositions, the suspended activated sludge
252 samples in the NF-MBRs were collected during steady operation period and the cake layer
253 foulants on NF2 membranes were collected at the end of operation. The genomic DNA of the
254 microbial community was extracted from the samples by PowerSoil[®] DNA isolation kit (MO bio,
255 USA). The 16S rRNA genes were sequenced by Illumina MiSeq platform using primers 357wF
256 (CCTACGGGGNGGCWGCAG) and 785R (GACTACHVGGGTATCTAATCC). The
257 sequencing results were analyzed by QIIME using the standard de novo operational taxonomic
258 unit (OTU)-based approach.

259 2.3.3. *NF membrane foulant solution preparation*

260 At the end of NF-MBR operation, the fouled NF modules were removed for autopsy
261 studies. The foulants were detached from the membrane surface (20 pieces of hollow fibers, 3
262 cm in length, total membrane area of 14.14 cm²) by soaking the membrane in 20 mL of ultrapure
263 water and ultra-sonicating for 30 min. The foulant solution (10 mL) was centrifuged at 4000 \times g
264 for 10 min and then filtered through a 0.45 μ m membrane filter. The supernatant (i.e., soluble

265 foulants) was collected for EPS and LC-OCD analysis. In addition, 10 mL of foulant solution
266 was digested by adding 2% nitric acid. The digested foulant solution was centrifuged at 4000 ×g
267 for 10 min and then filtered through a 0.45 µm membrane filter. The supernatant was collected
268 for inorganic element analysis by ICP-OES.

269 2.3.4. Confocal laser scanning microscopy (CLSM)

270 CLSM (Leica, German) was employed to record the biofilm morphology after the
271 membrane sample (1.3 mm × 0.45 mm) was stained with dyes. In detail, the biofilm on the
272 membrane surface was stained with LIVE/DEAD BacLight™ Bacterial Viability kits (Molecular
273 Probes, USA) according to manufacturer's instruction. The live and dead cells in the biofilm
274 were analyzed and the biovolume was calculated using IMARIS software (version 7.1.3,
275 Bitplane, Switzerland).

276 The EPS staining procedure was conducted according to a reported procedure (Adav et al.
277 2008). Briefly, 100 µL of 0.1M sodium bicarbonate buffer was added to the sample to maintain
278 the amino groups in non-protonated status. Then 10 µL of 10 g/L fluorescein isothiocyanate
279 (FITC, Molecular Probes, USA) solution was added to bind with proteins. The stained sample
280 was placed on a shaker for 1 h. After that, 100 µL of 250 mg/L concanavalin A (Con A)
281 conjugated with tetramethyl rhodamine (Molecular Probes, USA) was added to the sample to
282 bind with α-d-glucopyranose-D polysaccharides and incubated at room temperature for 30 min.
283 The stained sample was then washed twice with PBS to remove unbound dye. After that, 100 µL
284 of 300 mg/L Calcofluor white (Sigma-Aldrich, USA) was added to the sample for 30 min to
285 stain the β-polysaccharides and subsequently the sample was washed twice with PBS to remove
286 excess dye. Lastly, 60 µL of 100 mg/L of Nile red (Molecular Probes, USA) was employed to
287 bind with the intracellular lipids, followed by washing twice with PBS prior to CLSM analysis.

288 **3. Results and Discussion**

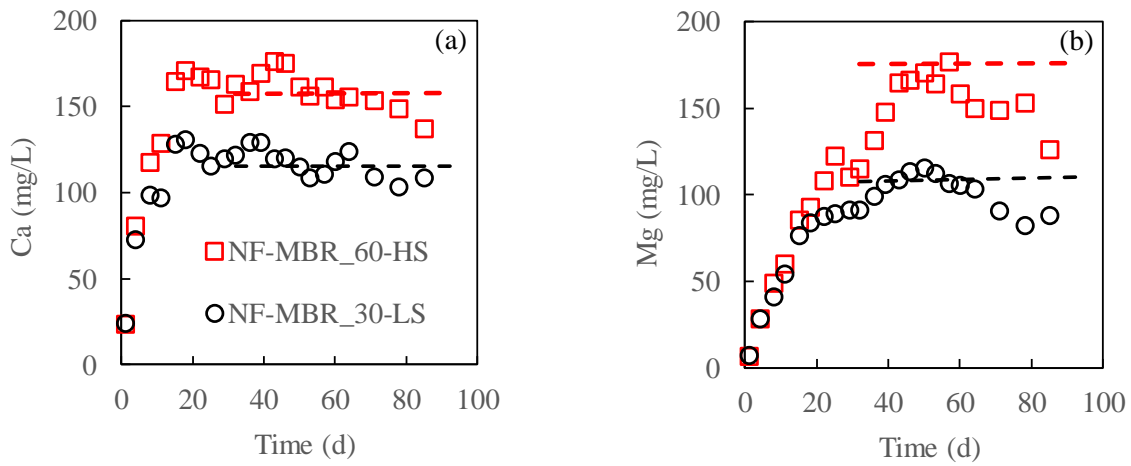
289 3.1. Impact of salt accumulation on NF-MBR performance

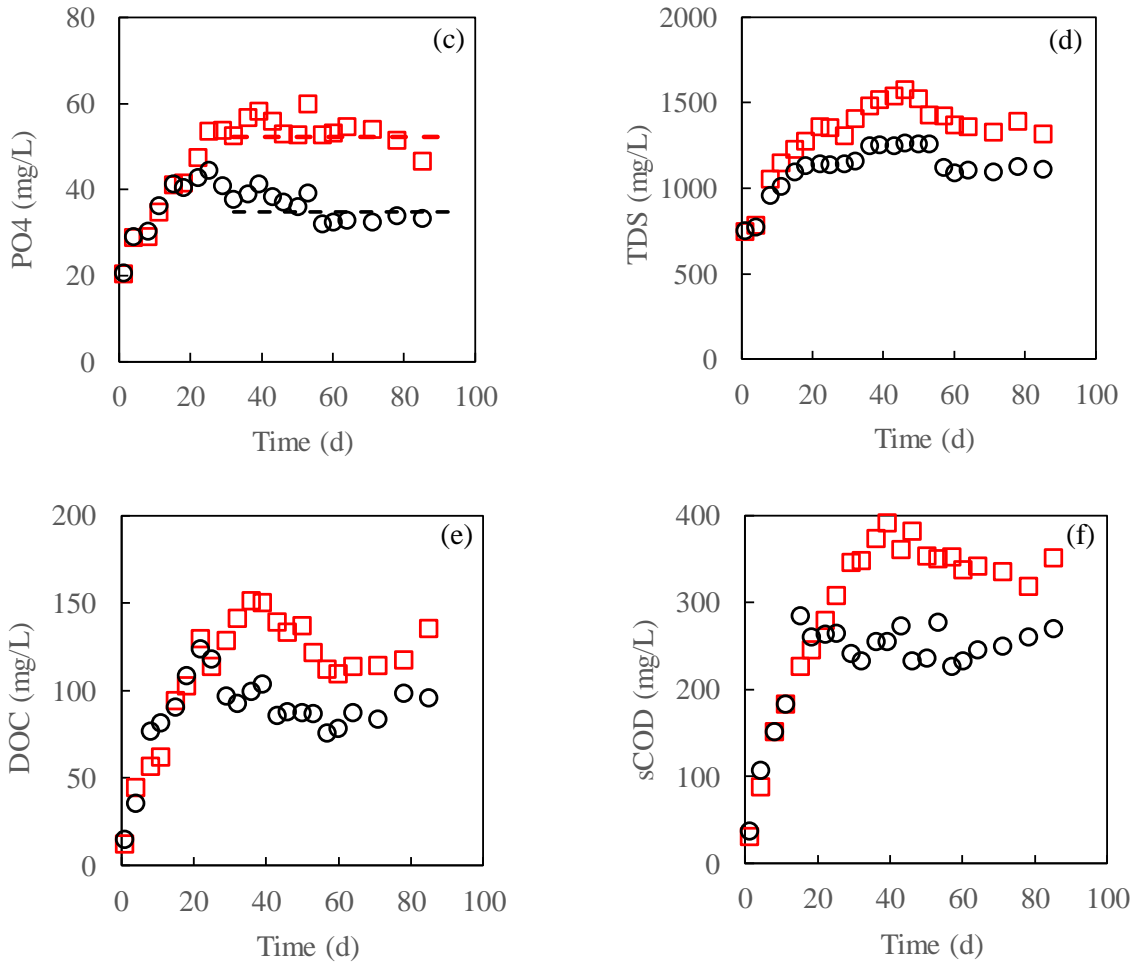
290 During 90-day operation of NF-MBRs, a gradual build-up and accumulation of scale-
291 forming ions such as Ca, Mg, and PO₄, as well as TDS, DOC, and sCOD in the bioreactors was
292 observed as expected due to their rejection by the NF membranes (Figure 1). The water quality
293 of feed, mixed liquor ad permeate of NF-MBRs at steady state are summarized in Figure 2.

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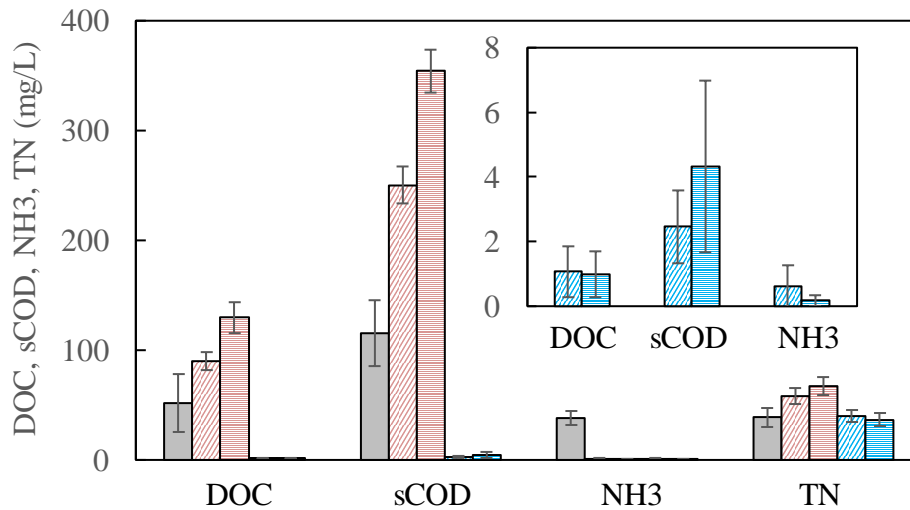
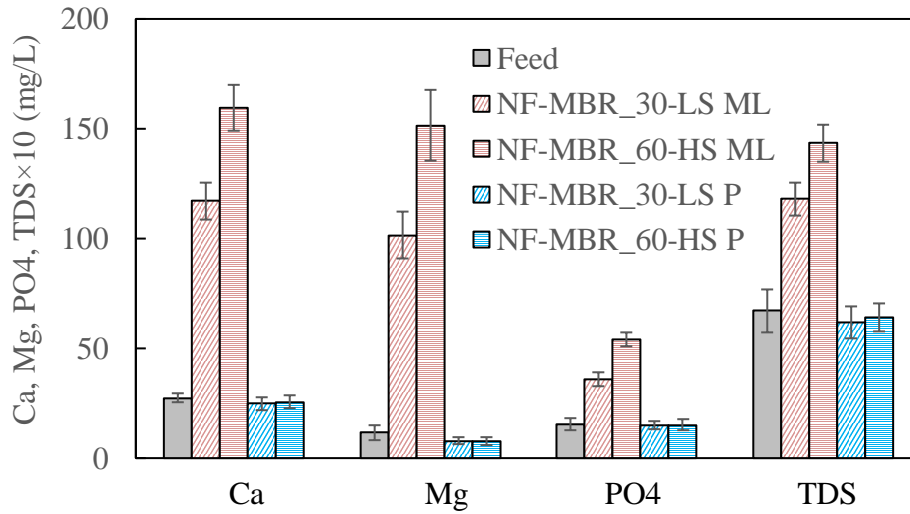
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296





297 **Figure 1. (a) Ca, (b) Mg, (c) PO₄, (d) TDS, (e) DOC, and (f) sCOD in the mixed liquor of**
 298 **NF-MBRs. Dotted-lines in the plot for Ca, Mg, PO₄ indicate the concentrations at steady**
 299 **state as predicted by Eq. (1).**



300

301

302 **Figure 2. Water quality of feed, mixed liquor, and permeate of NF-MBRs at steady state.**

303

304 Increasing the SRT, the measured concentrations of Ca (159.5 ± 10.6 mg/L), Mg
 305 (151.4 ± 16.3 mg/L), PO₄ (53.9 ± 3.3 mg/L), and TDS (1434.2 ± 85.3 mg/L) in the mixed liquor of
 306 NF-MBR_60-HS during steady state were approximately 36, 49, 51, and 22% higher than those
 307 in the NF-MBR_30-LS (Ca at 116.9 ± 8.3 mg/L, Mg at 101.4 ± 10.7 mg/L, PO₄ at 35.7 ± 3.1 mg/L,
 308 and TDS at 1179.0 ± 74.5 mg/L), respectively (Figure 1). To predict the mixed liquor salt
 309 concentrations, C_{ml} at steady state, Eq. (1) was developed based on the mass balance of the NF-

310 MBR system which relates the SRT, HRT and NF membrane rejection, η_{NF} (refer to Appendix
311 for details):

$$C_{ml}(t) = C_f \left[\left(1 - \frac{1}{\alpha} \cdot \frac{1}{HRT} \right) \cdot e^{-\alpha \cdot t} + \frac{1}{\alpha} \cdot \frac{1}{HRT} \right] \text{ where } \alpha = \left[\left(\frac{1}{HRT} - \frac{1}{SRT} \right) (1 - \eta_{NF}) + \frac{1}{SRT} \right]$$

312 Eq. (1)

313 (1)

314 Eq. (1) differs from the previous work that assumes the membrane (i.e., FO membrane) has
315 perfect rejection (Lay et al. 2010). Eq. (1) assumes no salt precipitation in the bioreactor and
316 scaling on the membrane surface, as well as no adsorption of inorganic ions onto the biomass or
317 biofilm on the biocarriers. The measured concentrations correlated well within 15% error with
318 the predicted concentrations, i.e., Ca = 115.3 and 157.7 mg/L, Mg = 108.6 and 175.6 mg/L, PO₄
319 = 34.8 and 52.2 mg/L for NF-MBR_30-LS and NF-MBR_60-HS, respectively. The
320 discrepancies observed between the predicted and measured values could be due to these
321 reasons.

322 Despite the good rejection properties of NF membrane, i.e., Ca = 79 - 84%, Mg = 92 -
323 95%, PO₄ = 58 - 72%, the overall removal attained for Ca, Mg, PO₄ was only <10, <35, and
324 <2%, respectively, thus the NF-MBR permeate has almost similar concentrations of Ca and PO₄
325 as the incoming feed. Further optimization work such as chemical assisted precipitation is
326 required to increase the removal efficiency. Tuning the NF membrane to get higher rejection,
327 i.e., increase the number of polyelectrolyte layers or make a tighter membrane, is a complicated
328 issue as on one hand excellent removal of divalent ions is desired but on the other hand increase
329 rejection of monovalent ions can cause an increase in osmotic pressure.

330 The soluble organics in terms of DOC and sCOD in the mixed liquors also increased
331 gradually before reaching steady values of DOC = 89.7±7.9 and 129.2±14.3 mg/L, sCOD =

332 250.1±17.1 and 354.1±19.7 mg/L for NF-MBR_30-LS and NF-MBR_60-HS, respectively
333 (Figure 1). The accumulation of soluble organics in the NF-MBRs could be attributed to their
334 rejections by the NF membrane and their slowly-biodegradable or non-biodegradable natures.
335 Similar to inorganic ions, greater accumulation of soluble organics was found in the NF-MBR
336 with longer SRT. The impact of salt accumulation on the biodegradation efficiency of NF-MBR
337 can be assessed by estimating the biodegradation rate at steady state, which was defined as:

$$338 \text{ biodegradation rate (\%)} = (1 - C_{\text{ml, measured}}/C_{\text{ml, predicted}}) \times 100\% \quad \text{Eq.}$$

339 (2)

340 $C_{\text{ml, measured}}$ is the measured concentration of organics in the mixed liquor while $C_{\text{ml, predicted}}$ is the
341 concentration of organics in the bioreactor assuming no biodegradation and negligible adsorption
342 onto the membrane, which can be obtained using Eq. (1). The results showed that the
343 biodegradation rate of DOC attained in NF-MBR_30-LS and NF-MBR_60-HS were ~92.0%
344 ($C_{\text{ml, predicted}} = 1115.5$ mg/L) and ~92.3% ($C_{\text{ml, predicted}} = 1679.8$ mg/L), respectively, showing
345 insignificant effect from salt accumulation at different SRT ($p > 0.05$).

346 Both NF-MBRs produced permeates with comparable DOC contents (1.1±0.8 mg/L in
347 NF-MBR_30-LS vs. 1.0±0.7 mg/L in NF-MBR_60-HS; $p > 0.05$), achieving >97% of DOC
348 removal (Figure 2). The NF-MBR_60-HS produced a permeate with slightly higher COD level
349 compared to the NF-MBR_30-LS (4.3±2.7 mg/L vs. 2.5±1.1 mg/L; $p < 0.05$). Nevertheless, both
350 NF-MBRs could achieve COD removal of >99% (Figure 2). Not surprisingly, the superior
351 permeate quality was owing to the excellent rejection properties of NF membranes toward
352 organic matters, i.e., >98.5%, thus no obvious impact of salt accumulation on permeate quality in
353 terms of DOC and sCOD could be observed, unlike in the case of dissolved ions. In addition,
354 both NF-MBRs achieved excellent nitrification levels (ammonia removal ratios >98%) and

355 produced permeates with very low NH₃ levels (<1 mg/L) despite the elevated salinity condition
 356 in the bioreactors. In contrast, the inhibition of nitrification process was commonly observed in
 357 FO-MBR and MD-MBR due to the accumulation of NaCl (Qiu and Ting 2013, Wang et al. 2016,
 358 Wijekoon et al. 2014). The nitrifiers responsible for the two-step bioprocess, i.e., ammonia
 359 oxidizing bacteria (AOB) and nitrite oxidizing bacteria (NOB), are sensitive to high salt
 360 environment (Campos et al. 2002, Moussa et al. 2006). In the NF-MBR system applied in this
 361 study, the negative impact arisen from the accumulation of NaCl was alleviated by allowing the
 362 passage of monovalent ions. Our findings suggested that the accumulation of Ca, Mg, and PO₄
 363 has no adverse influence on the nitrification process.

364 The organic compositions of feed, mixed liquor, and permeate of NF-MBRs were
 365 analyzed by LC-OCD (Table 3). In both NF-MBRs, it was observed that (1) NF membranes
 366 demonstrated complete rejection of biopolymers, >99% rejection of HS&BB, and >95%
 367 rejection of LWM; (2) HS&BB accounted for ~75% of total soluble organics in the bioreactors
 368 compared to ~31% in the feed, indicating their larger accumulation rates, i.e., highest
 369 concentration factor, due to their slowly-biodegradable or non-biodegradable nature; (3) Limited
 370 accumulation of biopolymers (~10%) and LMW (~14%) in the bioreactors due to their easily-
 371 biodegradable nature. In addition, the biodegradation rate of each fraction was estimated as
 372 summarized in Table 3, and no obvious impact of salt accumulation was observed since both NF-
 373 MBRs had similar biodegradation efficiencies for all components.

374

375 **Table 3. Compositions of soluble organics in the feed, mixed liquor, and permeate of NF-**
 376 **MBRs analyzed by LC-OCD (n=5).**

	Biopolymers	HS&BB	LMW
Feed (mg/L)	5.2 ± 0.9	11.0 ± 4.9	18.9 ± 2.9

		(14.9%)	(31.1%)	(54.0%)
	Mixed Liquor (mg/L)	8.1 ± 0.5	55.3 ± 5.0	10.9 ± 1.5
		(10.3%)	(75.7%)	(14.0%)
NF-MBR_30-LS	CF ^b	1.55	5.02	0.57
	Biodegradation rate (%)	94.37	77.73	95.38
	Permeate (mg/L)	N.D. ^a	0.1 ± 0.0	0.5 ± 0.3
		(0%)	(14.3%)	(85.7%)
	Mixed Liquor (mg/L)	12.4 ± 2.3	83.5 ± 3.8	16.5 ± 2.3
		(11.0%)	(74.4%)	(14.6%)
NF-MBR_60-HS	CF	2.38	7.59	0.87
	Biodegradation rate (%)	94.13	75.81	94.21
	Permeate (mg/L)	N.D.	0.1 ± 0.0	0.4 ± 0.1
		(0%)	(20%)	(80%)

377 ^a N.D. represents not detectable;

378 ^b CF represents concentration factor, $CF = C_{ml}/C_f$.

379 3.2. Impact of salt accumulation on biomass characteristics

380 3.2.1. Short-term spike test

381
382 Short-term impact of rapid exposure of high concentration of divalent ions, i.e., relative
383 to the feed concentration, on biomass characteristics was performed by spiking of salts into the
384 bioreactors at different concentrations, namely control (no addition of salts), 200 and 400 mg/L
385 Ca, 100 and 200 mg/L Mg. The physiological responses in terms of cell viability, microbial
386 activity, and EPS production on Day 0, 4 and 7 are summarized in Figure S3 (Appendix).
387 Interestingly, the abrupt change in salt concentrations did not show any negative impact on the
388 biomass characteristics. As compared to control, the addition of Ca increased the viability of
389 biomass by 52 and 33% for 200 and 400 mg/L Ca, respectively, whereas no changes for samples
390 spiked with Mg. In addition, the microbial activity as reflected by SOUR analysis showed 29%
391 increase at 200 mg/L Ca while no notable change for 400 mg/L Ca, 100 and 200 mg/L Mg. The

392 improved viability and SOUR could be attributed to the important role of calcium ions in
393 maintaining the cell structure, cell division, and signal transduction that subsequently promote
394 cell growth and increase microbial activity (Dominguez 2004, Yu and Margolin 1997).
395 Furthermore, the EPS production was increased by 35 and 34% when spiked with 100 and 200
396 mg/L Mg, respectively, but comparable EPS production as control when spiked with 200 and
397 400 mg/L Ca, which indicated an increase in Ca had negligible impact on EPS production in NF-
398 MBRs.

399 3.2.2. Microbial activity and viability

400 The characteristics of biomass in mixed liquor (i.e., suspended biomass) and on
401 biocarriers (i.e., attached biomass) during long term operation of NF-MBRs, i.e., gradual
402 accumulation of salts until steady state, are summarized in Table 4. The average MLSS (i.e., total
403 suspended solids) concentrations in NF-MBR_30-LS and NF-MBR_60-HS were relatively
404 comparable at 1277 ± 286 mg/L and 1282 ± 240 mg/L ($p>0.05$), respectively. In contrast, the
405 attached solids on biocarriers of NF-MBR_60-HS (594 ± 132 mg/L) was higher than NF-
406 MBR_30-LS (515 ± 134 mg/L; $p<0.05$), leading to a higher total solids amount ($p<0.05$). Both
407 NF-MBRs also had similar MLVSS/MLSS ratio (0.79-0.89; $p>0.05$). The viability and activity
408 of the suspended biomass were examined by ATP and SOUR analysis (Table 5). In both NF-
409 MBRs, the viability increased gradually with time, but SOUR maintained relatively constant
410 (Figure S4 in Appendix). Averagely, in NF-MBR_30-LS, the biomass had viable ratio of $59\pm 9\%$
411 and SOUR of 11.2 ± 1.9 mgO₂/gMLVSS·h, whereas the viability and SOUR in NF-MBR_60-HS
412 were $60\pm 12\%$ and 12.4 ± 2.4 mgO₂/gMLVSS·h, $p> 0.05$, respectively. Although the viability and
413 SOUR in NF-MBR_60-HS were slightly higher, the values were not statistically significant,

414 suggested that the impact of Ca accumulation was not remarkable in the long-term operation of
415 NF-MBR.

416 3.2.3. EPS production

417 In addition, the soluble and bound EPS productions of suspended biomass in both NF-
418 MBRs experienced increase during initial stage, followed by slow decrease, regardless of SRT
419 (Figure S5 in Appendix). The results agreed with the spike test above, which showed Mg had
420 greater impact on EPS production. It was also noted that soluble EPS accounted for almost 50%
421 of the total EPS in our NF-MBRs and such ratio was higher than those reported in conventional
422 MBRs (Wu et al. 2011, Zhang et al. 2010). Table 4 further compares the compositions of EPS
423 (i.e., polysaccharides and protein) in both NF-MBRs at steady state. The amount of protein was
424 almost three folds of polysaccharides for both soluble and bound EPS, regardless of the SRT. In
425 addition, higher soluble polysaccharides (i.e., ~35%) were present in NF-MBR_60-HS than NF-
426 MBR_30-LS ($p<0.05$), while comparable soluble proteins were found in both NF-MBRs
427 ($p>0.05$). Both NF-MBRs produced comparable amount of bound proteins ($p>0.05$) and bound
428 polysaccharides ($p>0.05$). The results suggested that SRT could influence soluble EPS
429 production in NF-MBRs, especially soluble polysaccharides.

430 3.2.4. Particle size and zeta potential

431 Furthermore, the suspended biomass in NF-MBR_60-HS had larger floc sizes ($p<0.05$)
432 and less negative surface charges ($p<0.05$) compared to those in NF-MBR-30-LS, (Table 4). This
433 could be attributed to the presence of higher concentrations of divalent cations under longer
434 SRT, which enhanced the bridging between negatively-charged functional groups of the bioflocs,
435 in particular the charge neutralization effect and decrease in zeta potential to become less
436 negative was more pronounced for Ca (Sobeck and Higgins 2002, Zhang et al. 2016). It has been

437 reported that biofloculation could be beneficial to membrane fouling reduction in conventional
 438 MBRs (Van den Broeck et al. 2012), its impact in NF-MBRs will be discussed in Section 3.3.

439

440

441

442

Table 4. Characteristics of biomass in NF-MBRs (n=11).

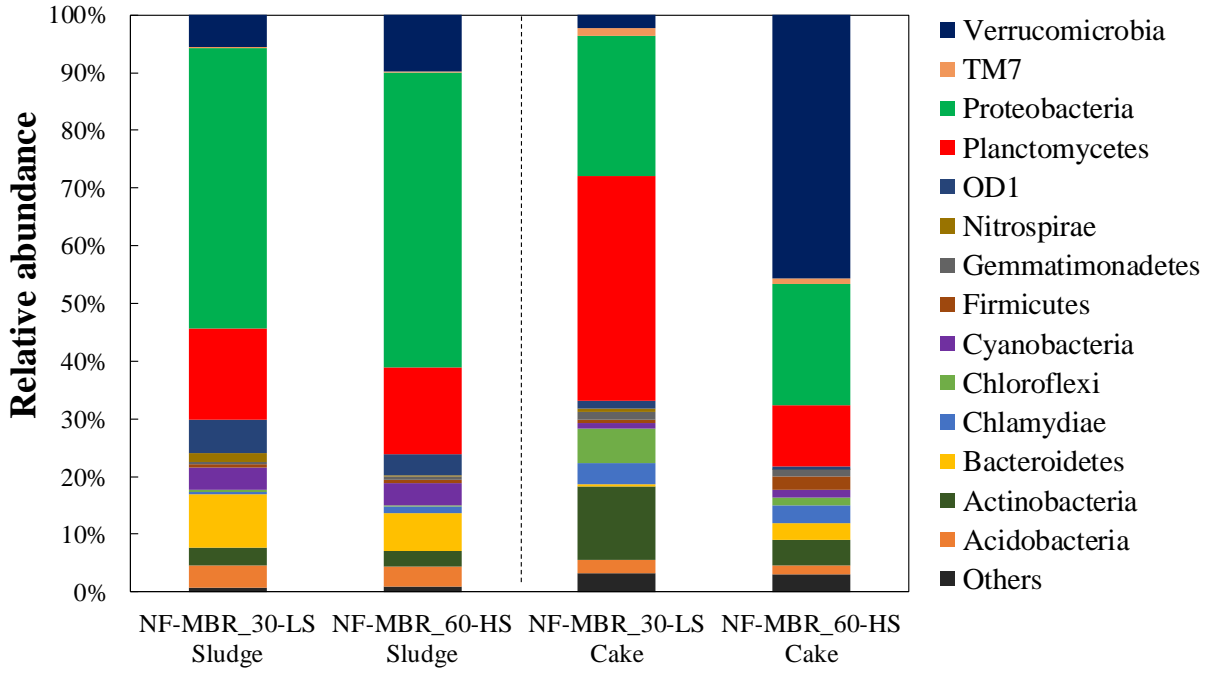
Biomass characteristics		NF-MBR_30-LS	NF-MBR_60-HS
Attached biomass	Total solids (mg/L)	515±134	594±132
	Total biomass (mg/L)	453±117	515±107
Suspended biomass	Total solids (MLSS) (mg/L)	1277±286	1282±240
	Total biomass (MLVSS) (mg/L)	1084±274	1061±223
	Viability (%)	59±9	60±12
	SOUR (mgO ₂ /gMLVSS·h)	11.2±1.9	12.4±2.4
	Soluble protein (mg/gMLVSS)	56±10	63±9
	Soluble polysaccharides (mg/gMLVSS)	23±5	31±4
	Bound protein (mg/gMLVSS)	69±7	70±10
	Bound polysaccharides (mg/gMLVSS)	23±6	25±4
	Particle size (µm)	95±4	114±5
	Zeta potential (mV)	-11.7±0.8	-9.9±1.0

443

444 3.2.5. Microbial community

445 In addition, the microbial community compositions and relative abundances of bacterial
 446 populations at phylum level of suspended activated sludge in both NF-MBRs are illustrated
 447 in Figure 3. It was found that at phylum level, Proteobacteria (48.7% in NF-MBR_30-LS vs.
 448 51.2% in NF-MBR_60-HS) and Planctomycetes (15.7% in NF-MBR_30-LS vs. 14.9% in NF-
 449 MBR_60-HS) were the predominant microbial communities. This observation was similar to the
 450 reported findings in conventional MBRs (Jo et al. 2016, Takimoto et al. 2018, Xia et al. 2012).
 451 Also, the microbial communities in both NF-MBRs had comparable relative abundances (Figure

452 3) with similar observed OTU, Chao1, Shannon and Simpson diversity indexes (Table 5). It
 453 implied that the microbial communities in NF-MBRs were not significantly affected by the
 454 increased organic and salt accumulations with extending SRT.
 455



456
 457 **Figure 3. Prokaryotic community compositions of activated sludge and cake layers from**
 458 **NF membranes at the phylum level. All classified taxa with relative abundance lower than**
 459 **1% and unclassified taxa were assigned to “Others”.**

460
 461
 462 **Table 5. The observed OTU, Chao1, Shannon and Simpson diversity indexes of prokaryotic**
 463 **communities in NF-MBRs.**

Sample	Observed OTU	Chao1	Shannon	Simpson
NF-MBR_30-LS sludge	555	585.9	4.129	0.943
NF-MBR_60-HS sludge	635	659.5	4.165	0.940
NF-MBR_30-LS cake	791	829.7	5.095	0.983

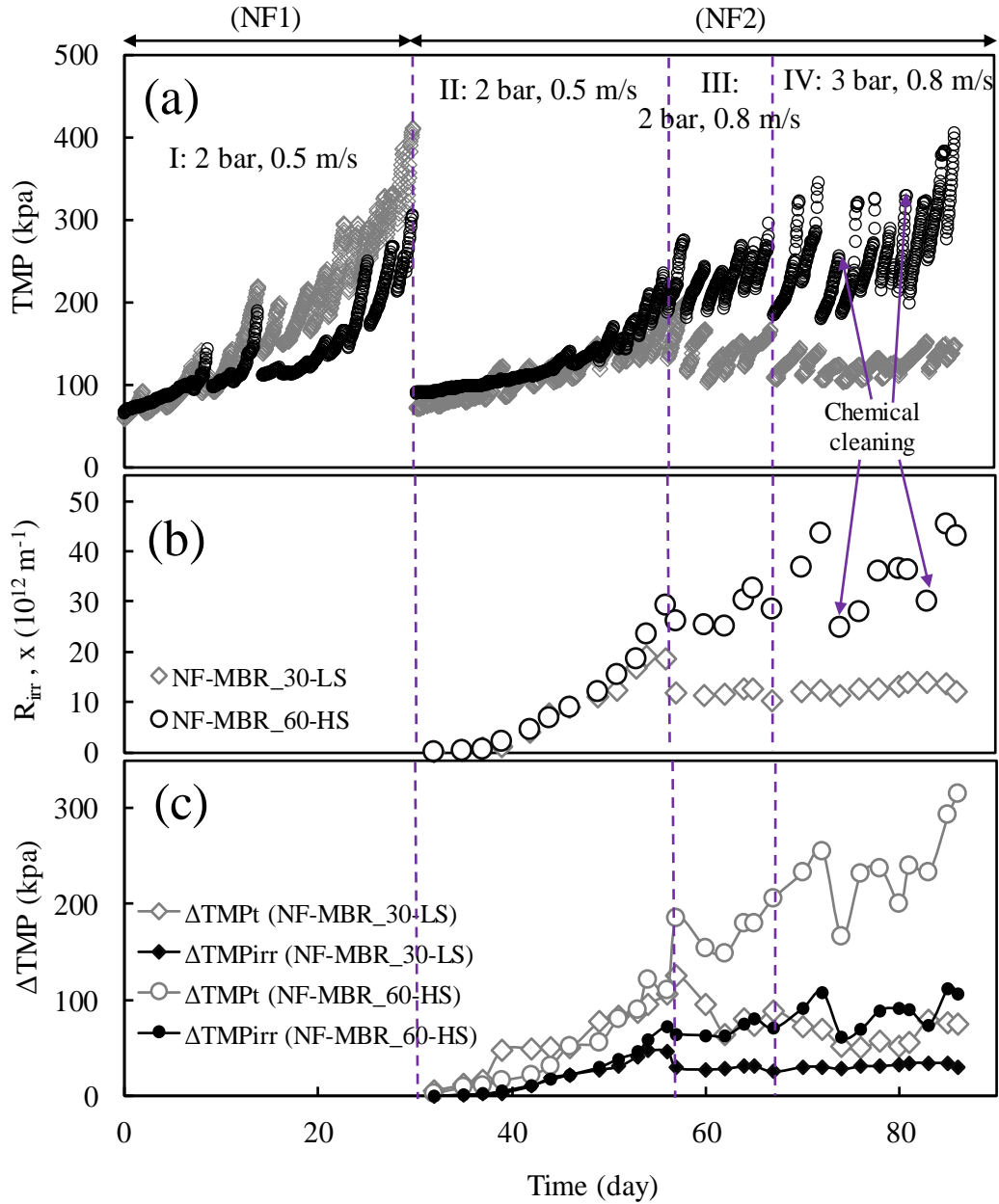
464

465 3.3. Impact of salt accumulation on NF membrane fouling

466 3.3.1. TMP and permeability of NF-MBRs

467 Figure 4a describes the TMP profile of NF-MBRs. During initial stage of operation
468 before the NF-MBRs reached steady operation and without applying the sponge column (from
469 Day 0 to 30, NF1), severe fouling accompanied by rapid increase in TMP was observed due to
470 high loading of bioflocs directly to the NF module, i.e., clogging of lumen. Hence, the discussion
471 will be focused on the later stages (II, III and IV) of operation for NF2. In general, the TMP
472 profile displayed a ‘sawtooth’ pattern, in particular during stage III and IV, which was associated
473 with the interplay between reversible, irreversible fouling and cake-enhanced osmotic pressure
474 effect (CEOP) (Tay et al. 2018). To further examine the NF membrane fouling mechanisms, the
475 irreversible fouling resistance, $R_{irr}(t)$, $\Delta TMP_t = (TMP(t) - TMP(0))$ and $\Delta TMP_{irr} = J\mu R_{irr}(t)$, as
476 shown in Figure 4b and c, were calculated from the clean membrane pure water permeability,
477 membrane pure water permeability after each physical cleaning cycle (Figure S6) and TMP
478 profile (Figure 4a) according to the model developed in our previous publication (refer to
479 Appendix) (Tay et al. 2018). During Stage II operation, at backwashing pressure of 2 bar and
480 flushing crossflow velocity of 0.5 m/s, there was an initial slow build-up (i.e., first 10 days of
481 operation of NF2) followed by an exponential increase of irreversible fouling, where $R_{irr} = \sim 20$
482 and $30 \times 10^{12} \text{ m}^{-1}$ after 25 days of operation of NF2 for NF-MBR_30-LS and NF-MBR_60-HS,
483 respectively. In Stage III, the flushing crossflow velocity was increased from 0.5 m/s to 0.8 m/s
484 with a fixed backwashing pressure of 2 bar in order to improve the cleaning efficiency, where the
485 R_{irr} dropped to 12 and $27 \times 10^{12} \text{ m}^{-1}$ for NF-MBR_30-LS and NF-MBR_60-HS, respectively.
486 The cleaning protocol appeared to be effective for the case of NF-MBR_30-LS as it prevented

487 further increase in R_{irr} , but not for the case of NF-MBR_60-HS as there was a linear (average)
488 increase in R_{irr} . In Stage IV, the backwashing pressure was increased from 2 to 3 bar with a fixed
489 flushing crossflow velocity of 0.8 m/s, but it did not further remove the irreversible fouling for
490 NF-MBR_30-LS, only maintained R_{irr} at the same level. Meanwhile, the R_{irr} continued to
491 increase linearly (average) for NF-MBR_60-HS, hence physical cleaning was not effective in
492 this case. These observations implied that the cake layer morphology in NF-MBRs may be
493 different ($\times 3.5$ higher in final R_{irr} value), i.e., more compact cake layer in NF-MBR_60-HS
494 compared to NF-MBR_30-LS, as a result of salt accumulation effect (discussed below). In
495 addition, chemical cleaning was performed using citric acid after 43 and 52 days of operation of
496 NF2 for both NF-MBRs. Apparently, the membrane permeability of NF-MBR_60-HS could be
497 restored by $\sim 12-42\%$, suggesting partial irreversible fouling was removed by chemical cleaning.
498 However, the effect of chemical cleaning in NF-MBR_30-LS was not remarkable ($< 5\%$
499 restoration), indicating the limitations of chemical cleaning using citric acid in the restoration of
500 membrane permeability. Therefore, further optimization of chemical cleaning protocol is
501 crucially important in future studies on NF-MBR. The NF membrane (layer-by-layer
502 polyelectrolytes of PAH and PSS) used in this work cannot tolerate $pH > 8$, one option is to apply
503 chemical-enhanced backwashing with weak acid, EDTA, or SDS (Lateef et al. 2013).



504

505

Figure 4. (a) TMP profile, (b) Irreversible fouling resistance, R_{irr} , and (c) ΔTMP_t and

506

ΔTMP_{irr} for NF-MBRs.

507

508

The CEOP effect, approximated by $\Delta \text{TMP}_{CEOP} \approx (\Delta \text{TMP}_t - \Delta \text{TMP}_{irr})$, is associated with

509

the cake thickness, porosity and tortuosity (Chong et al. 2008). Therefore, there was no

510

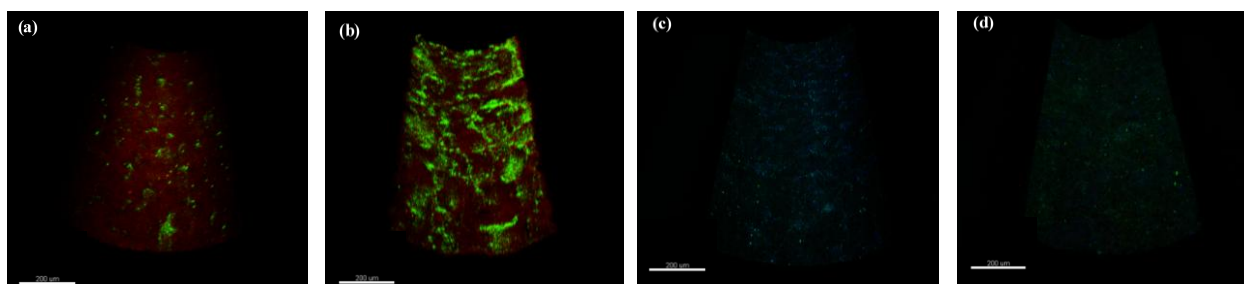
immediate and sharp rise in TMP (i.e., the upward part of the ‘sawtooth’) during initial stage

511 (first 10 days) of NF2 operation, but an obvious TMP increase was noticed when a substantial
512 cake layer was formed during later stage in Stage II as well as throughout Stage III and IV. As
513 shown in Figure 4c, the CEOP effect was the predominant contributor and more severe in TMP
514 increase for NF-MBR_60-HS, which was attributed to the thicker and more compact cake layer
515 as well as higher salt concentration (i.e., osmotic pressure), as compared to NF-MBR_30-LS.

516 3.3.2. *NF membrane foulants*

517 (1) Morphology of cake layer foulants

518 The CLSM images of cells (live cells and dead cells) and EPS (proteins, α -d-
519 glucopyranose-D polysaccharides, β -polysaccharides and intracellular lipids) are shown in
520 Figure 5 and the calculated biovolumes are summarized in Table 6. The biovolumes of live cells
521 (predominantly present on the cake layer surface), dead cells (abundantly distributed in both
522 inner and outer regions of the cake layers), and total EPS in the NF foulants of NF-MBR_60-HS
523 were higher than NF-MBR_30-LS. First, the higher amount of live cells and EPS production
524 were due to greater availability of DOC and no inhibition of salt accumulation to support the
525 biofilm formation. Second, the higher amount of EPS, especially polysaccharides, was associated
526 with the greater amount of Ca that formed polysaccharides-Ca complex with strong gelation
527 properties (Meng et al. 2011), which could not be easily removed by physical cleaning. Third,
528 the higher amount of dead cells could be due to the greater operating pressure, which led to the
529 formation of a compact cake layer that hindered the oxygen and nutrient transfer to the inner
530 region of cake layer.



531
 532 **Figure 5. CLSM images of cake layer foulants on NF membranes. Images of live/dead cells**
 533 **in (a) NF-MBR_30-LS and (b) NF-MBR_60-HS; Images of proteins (FITC), α -d-**
 534 **glucopyranose polysaccharides (Con A), β -polysaccharides (calcofluor white), and**
 535 **intracellular lipids (Nile red) in (c) NF-MBR_30-LS and (d) NF-MBR_60-HS.**

536

537 **Table 6. Characteristics of cake layer foulants on NF membranes in NF-MBRs (n=2).**

Foulants characteristics		NF-MBR_30-LS	NF-MBR_60-HS
CLSM image analysis	Live cells ($\mu\text{m}^3/\mu\text{m}^2$)	0.23 ± 0.02	1.75 ± 0.07
	Dead cells ($\mu\text{m}^3/\mu\text{m}^2$)	0.93 ± 0.43	1.69 ± 0.11
	Protein ($10^5 \mu\text{m}^3/\text{cm}^2$)	2.86 ± 0.18	2.72 ± 0.22
	α -d-glucopyranose polysaccharides ($10^5 \mu\text{m}^3/\text{cm}^2$)	3.36 ± 2.10	29.23 ± 11.34
	β -polysaccharides ($10^5 \mu\text{m}^3/\text{cm}^2$)	8.77 ± 3.33	34.27 ± 7.7
	Intracellular lipids ($10^5 \mu\text{m}^3/\text{cm}^2$)	N.D. ^a	N.D.
	LC-OCD analysis	Biopolymers (mg/m^2)	200.6 ± 10.3
HS & BB (mg/m^2)		93.1 ± 11.4	235.36 ± 22.4
LMW (mg/m^2)		31.8 ± 1.1	57.1 ± 3.0
ICP-OES analysis	Ca (mg/m^2)	160.6 ± 21.6	226.1 ± 14.1
	Mg (mg/m^2)	49.2 ± 2.7	97.6 ± 7.1
	P (mg/m^2)	103.0 ± 13.2	142.2 ± 31.5
	S (mg/m^2)	30.0 ± 20.6	48.8 ± 37.8
	Si (mg/m^2)	21.0 ± 12.4	19.1 ± 5.4

538 ^a N.D. represents "not detectable".

539 (2) Organic compositions

540 In both NF-MBRs, the biopolymers appeared to be the predominant species, accounting
541 to 51-61%. The results agreed well with our previous study on NF-MBRs operated at different
542 conditions (Tay et al. 2018). The organic fractions (i.e., biopolymers, HS&BB, and LMW) in the
543 cake layer foulants of NF-MBR_60-HS were greater than NF-MBR_30-LS. Possibly, this was
544 due to more adsorption from the higher concentration in the bulk solution as well as more dead
545 cell lysis in the cake layer biofilm.

546 (3) Inorganic compositions

547 Higher accumulations of Ca (226.1 mg/m²), Mg (97.6 mg/m²), P (173.6 mg/m²), S (48.8
548 mg/m²), and Si (19.1 mg/m²) were found on the NF membrane in NF-MBR_60-HS (vs. Ca of
549 160.8 mg/m², Mg of 49.2 mg/m², P of 103.0 mg/m², S of 30.0 mg/m², and Si of 21.0 mg/m² for
550 NF-MBR_30-LS) owing to the higher concentrations of salts in the mixed liquor, in particular,
551 calcium phosphate accounted for more than 60% of total inorganic foulants on the NF
552 membranes. Note that the solubility limit of Ca₃(PO₄)₂, i.e. 2.07×10^{-33} , has been far exceeded in
553 the bioreactors (refer to Appendix).

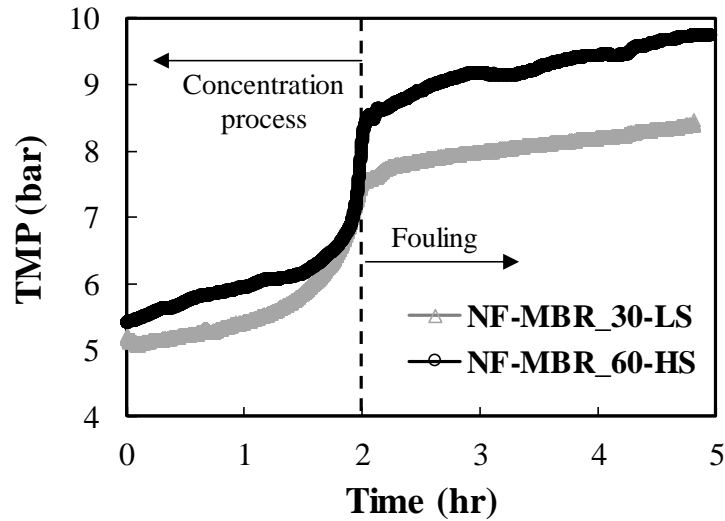
554 (4) Microbial community compositions

555 In the cake layer foulants of NF-MBR_30-LS, Planctomycetes and Proteobacteria were
556 the dominant phyla, accounting for 38.9% and 24.4%, respectively (Figure 4), similar to
557 conventional aerobic MBRs (Chen et al. 2016). In contrast, the dominant microbial community
558 in the cake layer foulants of NF-MBR_60-HS was Verrucomicrobia (45.7%), which showed a
559 low relative abundance in the suspended activated sludge (Figure 3). It has been reported that
560 Verrucomicrobia has the ability to hydrolyze high-molecular-weight substrates such as
561 polysaccharides (Cardman et al. 2014). Nevertheless, it suggested that salt accumulation could

562 affect the composition of microbial communities in the cake layer foulants of NF-MBRs. In
563 addition, the observed OTU and Chao1, Shannon, and Simpson diversity indexes of microbial
564 communities in the cake layers of NF-MBR_60-HS were 55.9%, 56.5%, 32.8%, and 16.5%
565 lower than NF-MBR_30-LS, respectively (Table 5). The differences in microbial community
566 structure may be attributed to factors, such as operating pressure (Gao et al. 2013), compositions
567 and contents of EPS on the membrane surface (Tsuneda et al. 2003).

568 3.4. Impact of salt accumulation in NF-MBRs on performance of downstream RO membranes

569 As shown in Figure 6, during the concentration stage (i.e., when recovery was increased
570 from 0 to 90%), the TMP increased from ~5.3 to 7.5 and 8.5 bar; while during continuous
571 operation at recovery equivalent to 90%, the fouling rates (i.e., dTMP/dt) were ~0.25 and ~0.38
572 bar/d for RO fed with permeate of NF-MBR_30-LS and NF-MBR_60-HS, respectively. In
573 conventional MF/UF-MBR+RO systems, it had been demonstrated that the higher organic levels
574 (in terms of TOC, COD, EPS) in MBR permeate could lead to more serious RO membrane
575 fouling (Wu et al. 2013). However, in this study, the permeates of both NF-MBRs had
576 comparable quality (except higher sCOD for permeate of NF-MBR_60-HS), but resulted in
577 relatively higher fouling rate for permeate of NF-MBR_60-HS. In addition, previous studies had
578 emphasized that assimilable organic carbon (AOC) was an important parameter in determining
579 the biofouling potential of RO feed water in seawater desalination (Lee et al. 2019, Wu et al.
580 2017), which had very low DOC concentration (< 2 mg/L) similar to the permeate of NF-MBR.
581 Therefore, it was unclear the difference in fouling rate was due the difference in sCOD level or
582 rather the compositions of organics. Further study will be focused on the effect of AOC in NF-
583 MBR permeate on RO membrane performance in order to further advance the NF-MBR+RO for
584 water reclamation.



585

586

Figure 6. Performance of RO fed with different NF-MBR permeates.

587

588 **4. Conclusions**

589 This study compared the bioreactor and membrane performances in parallel NF-
 590 MBR+RO systems for water reclamation application under different salt accumulation
 591 conditions in NF-MBRs, which were operated under SRT of 30 and 60 days, respectively. The
 592 following conclusions can be drawn:

593 (1) The concentration factor is a function of SRT, HRT and membrane rejection. Increasing the
 594 SRT resulted in greater accumulation of scale-forming inorganic salts (i.e., Ca, Mg, PO₄) and
 595 organic compounds (i.e., DOC, sCOD, and especially HS&BB due to its poor or non-
 596 biodegradable nature). The accumulation of divalent ions did not pose adverse impact to the
 597 biodegradation efficiency and both NF-MBRs achieved superior organic removal (>97%) and
 598 ammonia removal (>98%).

599 (2) The elevated divalent ions level promoted biofloculation, however, the effects on microbial
 600 activity and viability were not statistically significant in the long-term experiment. Salt

601 accumulation did not influence the microbial community structure of activated sludge in NF-
602 MBRs, but led to different predominant microbial communities in the cake layer foulants of NF
603 membranes.

604 (3) More severe membrane fouling in the NF-MBR at longer SRT (i.e., higher salt accumulation)
605 due to combined organic and inorganic fouling by calcium phosphate scaling and Ca-
606 polysaccharides complex that was not easily removed by physical cleaning. The thicker and
607 more compact irreversible fouling layer as well as higher salt concentration caused more severe
608 cake-enhanced osmotic pressure (CEOP) effect in the NF-MBR with longer SRT. Physical
609 cleaning (i.e., backwash and flushing) can be applied to maintain sustainable operation of the
610 NF-MBR at low SRT of 30 d.

611 (4) Higher RO membrane fouling rate was observed when fed with permeate of NF-MBR with
612 longer SRT, despite having similar water quality, i.e., DOC of <2 mg/L. One possible
613 explanation was the role of AOC towards RO fouling at low DOC level, which required further
614 investigation in future studies.

615

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621

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