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Supplementary Information

Hydrophobic bioadhesive composites for human motion detection

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CaproGlu Synthesis

4-[3-(Trifluoromethyl)-3H-diazirin-3-yl] benzoic acid (230.14 g.mol⁻¹ Dz-COOH) was synthesized from Dz-CH₂OH. The solution of Dz-CH₂OH (1 mol. eq.) and potassium hydroxide (1.55 mol. eq.; 0.6 M KOH solution) was dissolved in a mixture of water/dioxane (5/1), cooled down to 0 °C in an ice bath and potassium permanganate (1.5 eq.) was added portion-wise. The biphasic reaction mixture was vigorously stirred at room temperature overnight (18 hours). The suspension was filtered through a pad of Celite®. The filtrate was cooled in an ice bath and acidified with 2M HCl to pH=1. The obtained white precipitate was filtered off and dissolved in ethanol. The organic layer was washed 5 times with water and one time with brine, dried with anhydrous MgSO₄, and the solvent was evaporated in a vacuum to yield pure Dz-COOH as a pale-yellow solid. The Proprietary CaproGlu synthesis method was the esterification reaction between PCLT (polyol) and Dz-COOH (acid) with 1,1-carbonyldiimidazole (CDI) used as a coupling agent. The molar ratio of Dz-COOH/PCLT = 2/1 was deliberately chosen to yield ~50% diazirine conjugation. The reactants were placed in a round-bottomed flask with a magnetic stir bar. The flask was immersed in a water bath placed on the top of the heater equipped with a magnetic stirrer. DCM was added dropwise and stirred at room temperature for 1 h. The solution of PCLT in DCM was added into the reaction mixture slowly and the round bottom flask was fitted with a condenser. The reaction mixture was heated to 40 °C and stirred for 18 h. After completion DCM was removed on rotavap and the product CaproGlu was purified by water / Et₂O extraction (the water phase contained saturated NaHCO₃ to remove unreacted Dz-COOH). The organic layer was dried with anhydrous MgSO₄

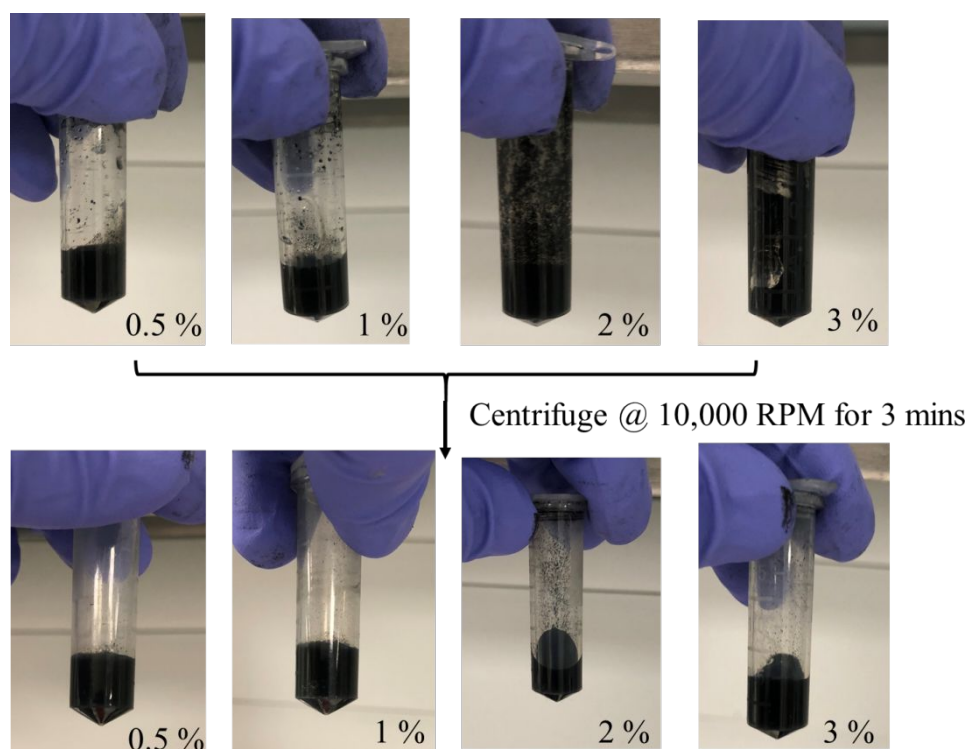
31 and filtrated. Et₂O was removed with rotavap. Traces of solvent from CaproGlu product were
32 removed on a high vacuum for 6 h at 40 °C.

33

34 **Technical details on the CNTs, as provided by the manufacturer ¹:**

Properties of CNT	Value
Average diameter	9.5 x 10 ⁻⁹ m
Average length	1.5 μm
Carbon purity	90 %
Surface Area	250-300 m ² /g
Volume resistivity	10 ⁻⁴ Ω cm
¹ NC7000™ - Technical Data Sheet (DM-TI-02-TDS-NC7000-V08). https://www.nanocyl.com/download/tds-nc7000/	

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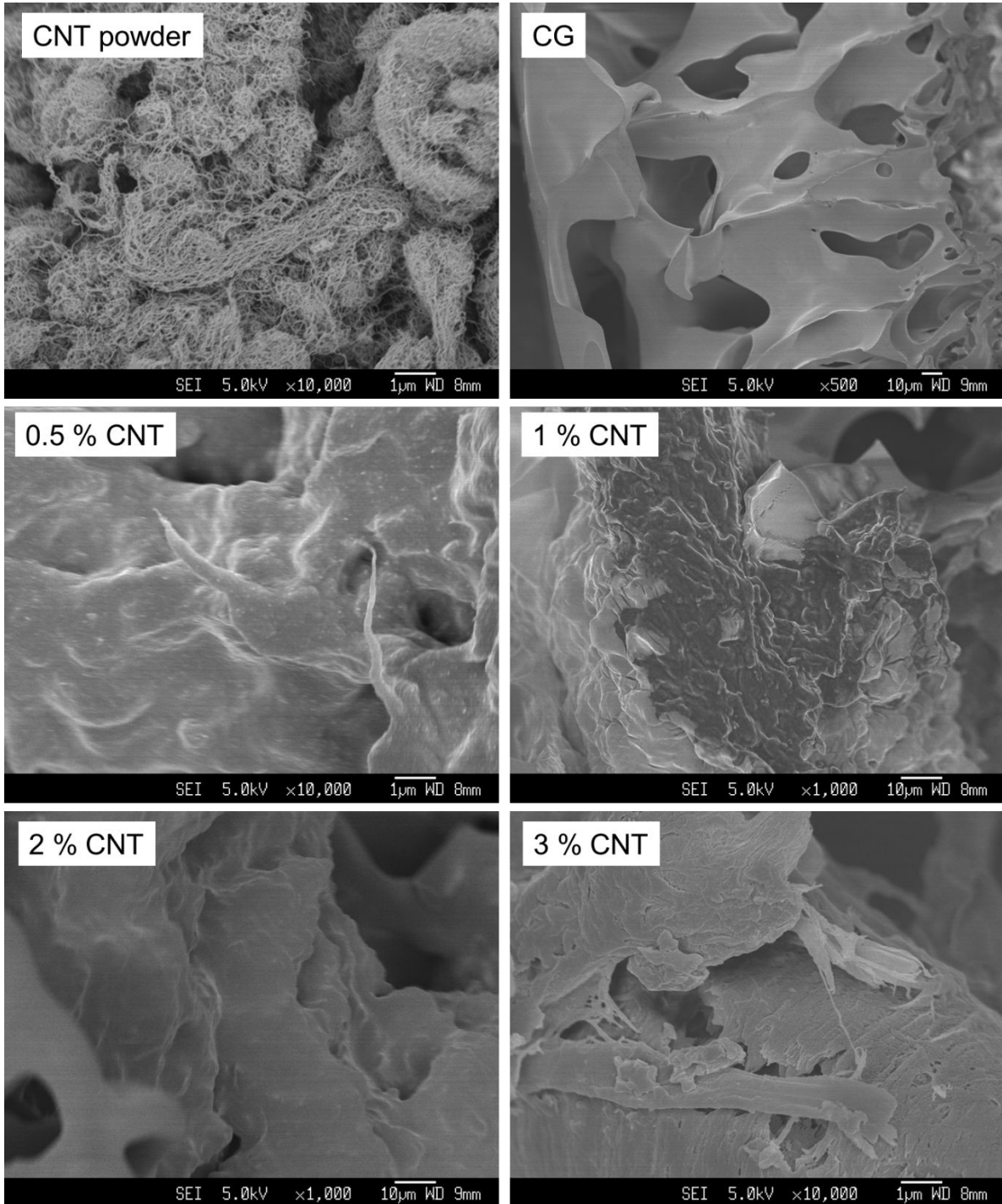
37 **Figure S1:** CNTs were physically mixed and remained dispersed in CaproGlu even after
38 centrifuging at 10000 RPM for 3 minutes.

39

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41 **Scanning Electron Microscopy (SEM)**

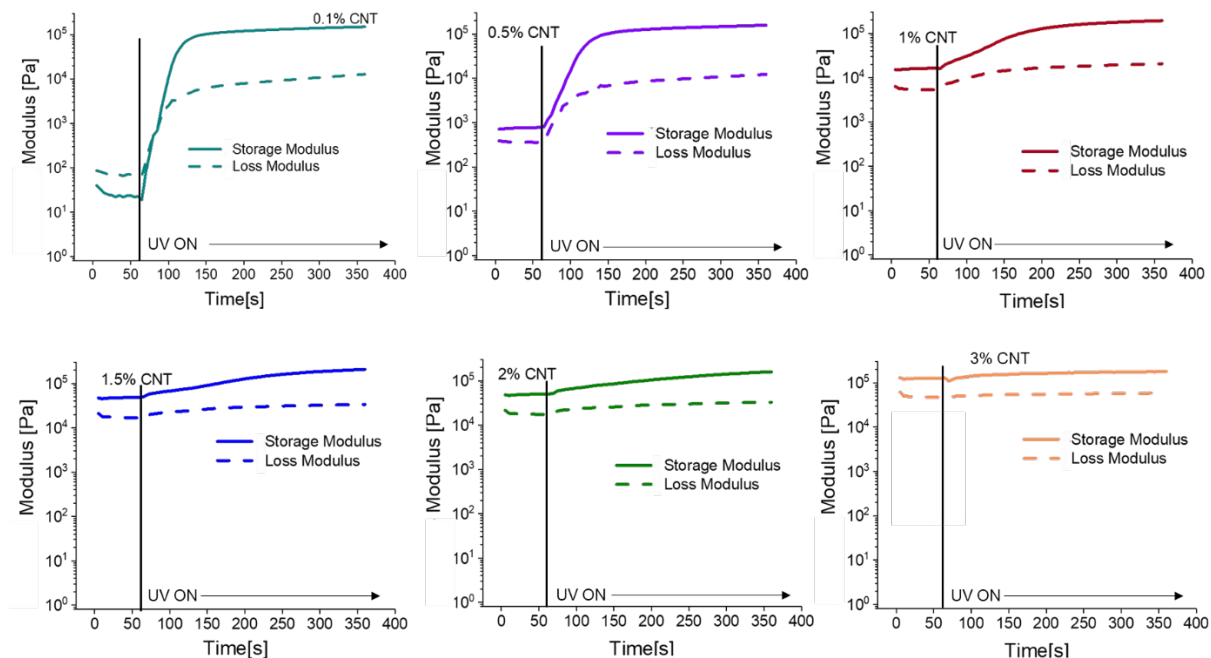
42 For qualitative analysis of CNTs dispersion in the UV-cured CaproGlu matrix, SEM analyzed
43 the morphology of the composites. Cured samples were subjected to platinum coating (90 s,
44 chamber pressure <5 Pa at 20 mA). Images were obtained at an acceleration voltage of 5 kV
45 and a working distance of 8-9 mm (**Figure S4**).



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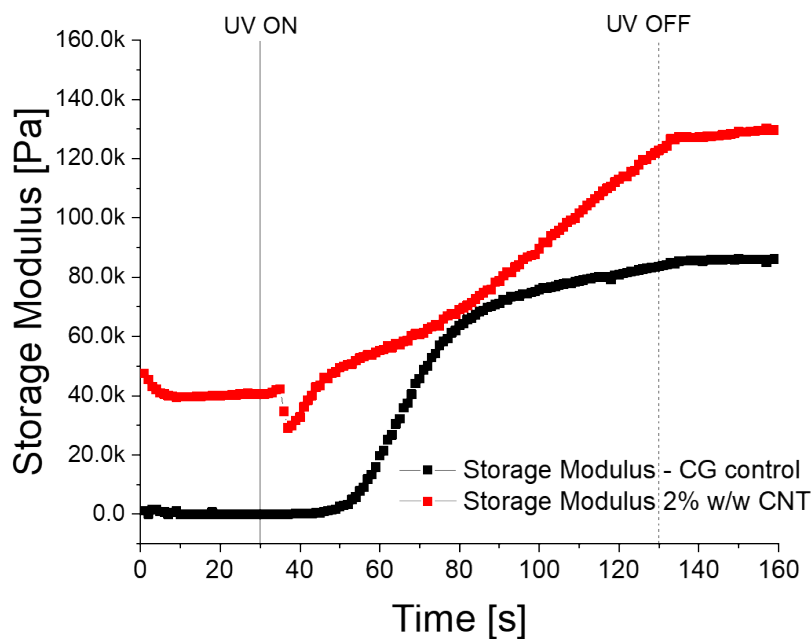
47 **Figure S2:** Cross-sectional morphology of CaproGlu composites under SEM. The surface of
 48 CaproGlu control (CG) seems smooth, whereas the cross-sectional morphology of the
 49 composites shows rough-textured dispersion of CNTs in the matrix.

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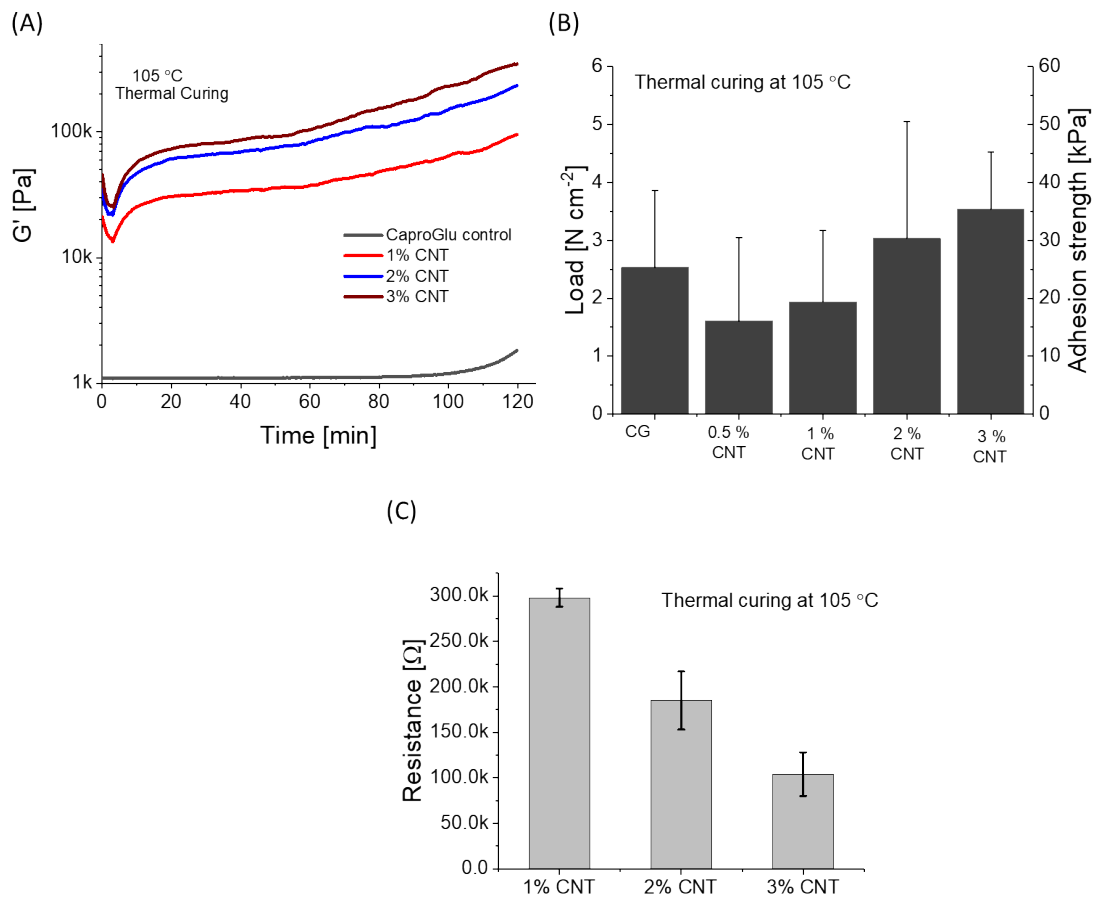
52 **Figure S3:** Real-time stimuli-based oscillatory rheology for CaproGlu composites showing G'
 53 and G'' .



54

55 **Figure S4:** Real-time stimuli-based oscillatory rheology for CaproGlu composites showing G'
 56 stabilizes once UV is turned off.

57 **Mechanical, adhesion, and electrical properties of the composites under thermal curing**
58 **at 105 °C**



59

60 **Figure S5:** Evaluation of CaproGlu composites under thermal curing at 105 °C. (A)
61 Comparison of real-time stimuli-based oscillatory rheology (G') for CaproGlu composites. (B)
62 Comparison of maximum lap-shear adhesion strength at a strain rate of 3 mm min⁻¹. (C)
63 Comparison of resistance measured by the two-point probe method.