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<th><strong>Title</strong></th>
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A Magnetic Field-enriched Surface-enhanced Resonance Raman Spectroscopy Strategy Towards the Early Diagnosis of Malaria

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ABSTRACT

Early malaria diagnosis is important because malaria disease can develop into fatal illness within hours upon the appearance of the first symptom. The low concentration of the diagnosis biomarker, hemozoin, at the early stage of malaria disease makes early diagnosis difficult. In this paper, we present a magnetic field-enriched surface-enhanced resonance Raman spectroscopy (SERRS) strategy for the sensitive detection of \( \beta \)-hematin crystals, which is equivalent to hemozoin in the characteristics of Raman spectrum, by using magnetic nanoparticles. We observe several orders of magnitude enhancement in the SERRS signal of enriched \( \beta \)-hematin in comparison to the Raman signal of \( \beta \)-hematin in the cases of SERRS alone or magnetic enrichment alone, showing the great potential of this method towards early malaria diagnosis.

Keywords: Hemozoin, Raman spectroscopy, surface-enhanced Raman scattering, resonance Raman, hematin, malaria, magnetic nanoparticles.

1. INTRODUCTION

Human malaria disease is a worldwide disease.\(^1\) Early diagnosis is critical to reducing morbidity and mortality rates since fatal illness can be resulted within hours after the development of the first malaria disease symptom.\(^2\) In malaria diagnosis, Raman spectroscopy has shown great potential for the detection of a malaria biomarker, hemozoin.\(^3\) Compared to blood smear examination, which is the current "gold standard" method, Raman diagnosis is faster and less labor-intensive; moreover, it requires minimal expertise for data interpretation.\(^4\) However, one significant disadvantage of Raman spectroscopy is the intrinsically weak Raman signal,\(^5\) which is aggravated by the low concentration of the malaria diagnosis biomarker at the early stage of malaria infection.\(^6\)

In this paper, we propose a magnetic field-enriched surface-enhanced resonance Raman spectroscopy (SERRS) strategy to enrich the \( \beta \)-hematin crystals, which is equivalent to hemozoin in the characteristics of Raman spectra,\(^7\) by using Fe\(_3\)O\(_4@Ag\) nanoparticles to enhance the SERRS signal. We will also compare the different Raman strategies for the detection of \( \beta \)-hematin crystals: 1) magnetic field-enriched SERRS, 2) SERRS, 3) magnetic field-enriched RRS, and 4) RRS to demonstrate the potential for integrating SERRS and magnetic enrichment for the sensitive detection of hemozoin towards early malaria diagnosis.

2. MATERIALS AND METHODS

2.1 Synthesis of \( \beta \)-hematin crystals

An acid-catalyzed method was used to fabricate the \( \beta \)-hematin crystals.\(^8\) 7.9 mM of Ferriprotoporphyrin IX chloride was dissolved in a 0.1-M NaOH solution under constant heating at 60 °C and stirring at 150 rpm. After 10 min and 14 min, 1.45 ml of HCl (1 M) and 8.825 ml of acetate solution were mixed into the solution, respectively. The heating was continued for another 46 min, after which the heating and stirring were stopped, and the mixture was left undisturbed in a dark environment for 24 h. Finally, the solute was washed with methanol and then by deionized water. The solute was collected by filtration with 0.2 \( \mu \)m supor filter and dried at room temperature over P\(_2\)O\(_5\) for 48 h.

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2.2 Fabrication of Fe₃O₄@Ag magnetic nanoparticles

The nanoparticles with iron oxide core and silver shell were synthesized by using the seed-growth reduction method.⁹,¹⁰ A mixture of Fe₃O₄ nanoparticles at concentration of 16.2 mM in 20 ml of ethanol was added drop-wise to ethanol (80 ml) with polyacrylic acid (0.15 g), prior to the sonication of the mixture for 15 min. Subsequently, the Fe₃O₄ nanoparticles were collected by using a magnet and washed with ethanol. A mixture of deionized water and ethanol (19.4:80.6: % v/v) with 2.8-mM AgNO₃ was used to re-dispersed Fe₃O₄ nanoparticles (2.1 mM) in an ultrasonic bath for 30 min. Then, 4.1 mM of hydroxylamine hydrochloride and 8.1 mM NaOH in triton X-100, ethanol and deionized water (9.0:70.8:28.3% v/v/v) was added drop-wise (5.88 μl/sec) to the suspension of Fe₃O₄ nanoparticles to reduce the adsorbed Ag⁺ salt. Finally, 19.4 mM of AgNO₃ in triton X-100, ethanol and deionized water (2:65.3:32.7% v/v/v) was added drop-wise (5.88 μl/sec) to the mixture. This second AgNO₃ addition was to ensure growth of Ag on the Ag seeds formed on the Fe₃O₄ core at the first introduction of AgNO₃. The Fe₃O₄ nanoparticles were collected using a magnet, washed, suspended in ethanol (15 ml) and followed by the filtering with 0.2-μm supor filters.

2.3 Raman Instrumentation

All samples were analyzed using an inVia Renishaw Raman system with 633-nm excitation wavelength. Excitation power of 10 mW for ordinary Raman spectroscopy and 0.1 mW for SERS study of R6G and β–hematin at different concentrations.

2.4 Field emission scanning electron microscope (FESEM) analysis

The nanoparticles and β–hematin were imaged with a JOEL JSM-6700F field emission scanning electron microscope (FESEM) with an acceleration voltage of 5 kV.

3. RESULTS

Figure 1 shows the field emission scanning electron microscope (FESEM) image of fabricated β–hematin crystals. These β–hematin crystals have sizes comparable to that of the hemozoin biocrystals reported in the literature.¹¹ Hence, similar SERRS enhancement effect would be resulted in the two different types of crystals due to the close resemblance in the spatial dimensions.

Figure 1. FESEM image of β–hematin crystals.
Figure 2(a) gives FESEM image of the raw Fe₃O₄ nanoparticles before coating with silver shell. The Fe₃O₄ have a mean diameter of 50 nm and aggregated together [Fig. 2(b)]. In contrast, the Fe₃O₄@Ag nanoparticles are well-dispersed with the use of surfactant to prevent the nanoparticles from aggregating together [Fig. 2(c)]. Each Fe₃O₄@Ag nanoparticles has a mean diameter of about 140 nm [Fig. 2(d)].

![FESEM images of raw Fe₃O₄ and Fe₃O₄@Ag nanoparticles](image)

Figure 2. (a) FESEM and (b) zoomed in FESEM image of raw Fe₃O₄ nanoparticles. (c) FESEM and (d) zoomed in FESEM image of Fe₃O₄@Ag nanoparticles.

Figure 3 illustrates the magnetic Fe₃O₄@Ag nanoparticles suspended in a solution (a) without and (b) with the influence of an external magnet. Without an external magnet [Fig. 3(a)], the magnetic Fe₃O₄@Ag nanoparticles are suspended in the solution, giving a uniform color. On the contrary, these magnetic nanoparticles are attracted to the external magnet to render a clear solution on the side of the vial without the influence of a magnet [Fig. 3(b)].

![Images of magnetic Fe₃O₄@Ag nanoparticles with and without magnet](image)

Figure 3. (a) Fe₃O₄@Ag nanoparticles and (b) attracted to the wall of vial in a solution next to an external magnet.
Figure 4 shows the SERS spectra of aqueous R6G solution at concentrations ranging from $10^{-6}$ M to $10^{-9}$ M absorbed on the fabricated Fe$_3$O$_4$@Ag nanoparticles. Prominent Raman peaks, such as C – C – C ring in-plane bending (611 cm$^{-1}$), CH out-of-plane bending (772 cm$^{-1}$), C – O – C stretching (1181 cm$^{-1}$), C – C/ C – N stretching (1308 and 1360 cm$^{-1}$), and aromatic C – C stretching (at 1507 cm$^{-1}$), can be clearly observed in the SERS spectra of R6G absorbed on the Fe$_3$O$_4$@Ag nanoparticles. These results demonstrate that the fabricated SERS-active nanoparticles are effective in the sensitive detection of chemical test molecules and thus, the sensitive detection of biocrystals, $\beta$ – hematin, at low concentrations could also be feasible with the use of these Fe$_3$O$_4$@Ag nanoparticles.

![SERS spectra of R6G](image)

Figure 4. SERS spectra of R6G at a range of concentrations ($10^{-6}$ to $10^{-9}$ M) adsorbed on Fe$_3$O$_4$@Ag nanoparticles. The SERS spectrum of R6G at concentration $10^{-9}$ M is scaled by a factor of 10 for the purpose of comparison.

![SERRS and RRS spectra of $\beta$ – hematin](image)

Figure 5. (a) Magnetic field-enriched SERRS spectrum of $\beta$ – hematin (concentrations of $10^{-7}$ M). (b) SERRS spectrum of $\beta$ – hematin (concentrations of $10^{-5}$ M). (c) Magnetic field-enriched RRS spectrum of $\beta$ – hematin (concentrations of $10^{-6}$ M). (d) RRS spectrum of $\beta$ – hematin (concentrations of $10^{-4}$ M). The excitation power was 0.1 mW for SERRS spectra and 10 mW for RRS spectra.
Figure 5 shows the (a) magnetic field-enriched SERRS spectrum, (b) SERRS spectrum, (c) magnetic field-enriched RRS spectrum, and (d) RRS spectrum of $\beta$–hematin. Prominent Raman peaks, such as $\nu_6$ (345 cm$^{-1}$, based on the tetragonal $D_{4h}$ system electron spin and crystallographic coordination notation for resonance Raman peaks studies on myoglobin), $\gamma_6$ (367 cm$^{-1}$), $\nu_{15}$ (754 cm$^{-1}$), $\nu_{22}$ (1120 cm$^{-1}$), $\nu_1$ (1551 cm$^{-1}$), $\nu_2$ (1570 cm$^{-1}$), and $\nu_{10}$ (1628 cm$^{-1}$), are observed in these spectra of $\beta$–hematin. Among the four different strategies, the magnetic field-enhanced SERRS measurement is the most effective, although low $\beta$–hematin concentration of $10^{-7}$ M and laser excitation power of 0.1 mW has been used. Therefore, the descending order of detection sensitivity for the four different strategies is magnetic field-enriched SERRS, SERRS, magnetic field-enriched RRS, and RRS. More details of this study can be found in a paper in press.

4. CONCLUSIONS

In conclusion, the magnetic field-enriched SERRS strategy can enhance the Raman signal of $\beta$–hematin using Fe$_3$O$_4$@Ag nanoparticles under the influence of a magnetic field. This strategy shows promises for the early diagnosis of malaria disease.

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