<table>
<thead>
<tr>
<th><strong>Title</strong></th>
<th>K10[Co4(H2O)2(B--SiW9O34H)2]-21H2O : a sandwich polyoxometalate based on the magnetically interesting element cobalt</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Author(s)</strong></td>
<td>Zhang, Li-Zhi; Gu, Wen; Liu, Xin; Dong, Zhili; Yang, Yong Sheng; Li, Bing; Liao, DaiZheng</td>
</tr>
<tr>
<td><strong>Citation</strong></td>
<td>Zhang, L. Z., Gu, W., Liu, X., Dong, Z., Yang, Y. S., Li, B., &amp; Liao, D. Z. (2007). K10[Co4(H2O)2(B--SiW9O34H)2]-21H2O: a sandwich polyoxometalate based on the magnetically interesting element cobalt. Inorganic Chemistry Communications, 10(11), 1378-1380.</td>
</tr>
<tr>
<td><strong>Date</strong></td>
<td>2007</td>
</tr>
<tr>
<td><strong>URL</strong></td>
<td><a href="http://hdl.handle.net/10220/8867">http://hdl.handle.net/10220/8867</a></td>
</tr>
<tr>
<td><strong>Rights</strong></td>
<td>© 2007 Elsevier B.V. This is the author created version of a work that has been peer reviewed and accepted for publication by Inorganic Chemistry Communications, Elsevier B.V. It incorporates referee’s comments but changes resulting from the publishing process, such as copyediting, structural formatting, may not be reflected in this document. The published version is available at: [DOI:dx.doi.org/10.1016/j.inoche.2007.08.025].</td>
</tr>
</tbody>
</table>
K\textsubscript{10}[Co\textsubscript{4}(H\textsubscript{2}O)\textsubscript{2}(B-α-SiW\textsubscript{9}O\textsubscript{34}H)\textsubscript{2}] \cdot 21H\textsubscript{2}O: A sandwich polyoxometalate based on the magnetically interesting element cobalt

Li-Zhi Zhang \textsuperscript{a,b}, Wen Gu \textsuperscript{a}, Xin Liu \textsuperscript{a,*}, Zhili Dong \textsuperscript{b,*}, Yong-Sheng Yang \textsuperscript{a}, Bing Li \textsuperscript{a,b}, Dai-Zheng Liao \textsuperscript{a}

\textsuperscript{a} Department of Chemistry, Nankai University, Tianjin 300071, PR China
\textsuperscript{b} School of Materials Science and Engineering, Nanyang Technological University, Singapore 639798, Singapore

* Corresponding authors. Tel.: +86 22 2350 5020; fax: +65 6790 6727. E-mail addresses: liuxin64@nankai.edu.cn (X. Liu), zldong@ntu.edu.sg (Z. Dong).

Abstract

The magnetism of a sandwich tetracobalt(II) substituted tungstosilicate K\textsubscript{10}[Co\textsubscript{4}(H\textsubscript{2}O)\textsubscript{2}(B-α-SiW\textsubscript{9}O\textsubscript{34}H)\textsubscript{2}] \cdot 21H\textsubscript{2}O has been studied based on an isotropic Heisenberg model supplemented by a zero-field splitting term.

Keywords: Cobalt(II); Sandwich; Polyoxometalate; Magnetism

As the largest subclass of transitional metal substituted polyoxometalates (TMSPs), sandwich-type compounds have received increasing interest for their magnetism in recent years [1]. Though their structures containing more than one paramagnetic transition metal ion in close proximity may exhibit exchange-coupled spins leading to large spin ground states, these compounds often display the weak magnetism because the magnetic sites are encapsulated in a diamagnetic molecular metal oxide cluster [1c,1d]. As a matter of fact, the special use of these compounds in magnetism area is that they offer ideal structural supports for the study of the interactions between paramagnetic metal atoms [2,3] as well as between delocalized electrons and paramagnetic metal atoms [4].

On the other hand, it is to be noted that the interaction between orbitally degenerate ions, as for example high-spin cobalt(II) in octahedral sites, remains an essentially unsolved problem [5]. The high symmetry of sandwich-type TMSPs and the small distortion of the octahedral CoO\textsubscript{6} sites should facilitate the theoretical treatment required to give a quantitative explanation of the cobalt–cobalt magnetic interaction.
Herein, we report the synthesis, structure and magnetism of a sandwich TMSPs based on the magnetically interesting element Co.

Reaction of CoCl₂ · 6H₂O (0.2 mmol, 0.048 g) and Na₂[β-SiW₁₀O₃₄H] · 23H₂O [6] (0.1 mmol, 0.28 g) in 10 mL of KAc buffer (pH 4.8) at 80 °C for 1 h followed by slow evaporation at room temperature resulted in red single crystals (0.15 g, yield 54%) [7]. X-ray analyses [8] reveal the compound, crystallized as K₁₀[Co₄(H₂O)₂(B-α-SiW₉O₃₄H)₂] · 21H₂O (1), belongs to monoclinic system. The dimeric polyoxoanion [Co₄(H₂O)₂(B-α-SiW₉O₃₄⁻ H)₂]₁₀⁻ consists of two lacunary [B-α-SiW₉O₃₄H]¹⁺ Keggin moieties linked by four Co²⁺ ions, leading to a sandwich-type structure with idealized C₂ᵥ symmetry (Fig. 1). The central belt of 1 is composed of a rhombic Co₄O₁₆ group consisting of four coplanar CoO₆ octahedral sharing edges (Fig. 2). The distances of Co···Co along the sides of the rhombus are 3.18 Å while along the diagonal of the rhombus are 3.30 Å and 5.44 Å, respectively. There is a narrow variation in the Co–O–Co angles, with an average value of 95.8°. These values for the metal–metal separation and the geometry of the bridging unit are very similar to that found in the phosphorus-containing polyanion [Co₄(H₂O)₂(α-PW₉O₃₄)₂]¹₀⁻ which is the first example of sandwich TMSPs as well as of cobalt-based sandwich TMSPs reported by Weakley et al. [9].

Although silicon-containing analogues SiMn₂W₉, SiCu₂W₉, and SiZn₂W₉ had been well reported by Kortz et al., [1c] which were synthesized by reacting M²⁺ ions with [γ-SiW₁₀O₃₆]⁶⁻ under conditions very similar to that used for the synthesis of 1, we were not able to synthesize [Co₄(H₂O)₂(B-α-SiW₉O₃₄H)₂]₁₀⁻ (SiCo₂W₉) starting from [γ-SiW₁₀O₃₆]⁶⁻ under the same conditions. Interestingly, the reaction starting from [β-SiW₉O₃₄H]⁹⁻ resulted in [B-α-SiW₉O₃₄H]⁹⁻ anions in compound 1, which meant that the mechanism of the metal insertion reaction must be accompanied by isomerization ([β-SiW₉O₃₄]¹₀⁻ → [B-α-SiW₉O₃₄]¹₀⁻). In our attempts to perform the reaction starting from Na₁₀[α-SiW₉O₃₄] [6], strangely, the crystals of the same compound could not be obtained. We speculated the Na₁₀[α-SiW₉O₃₄] synthesized according to Tézé’s method was a mixture of [A-α-SiW₉O₃₄]¹₀⁻ and [B-α-SiW₉O₃₄]¹₀⁻, and thus caused the desired compound hard to be crystallized. Also deserving attention, despite the synthesis distinction between compound 1 and the three compounds (SiMn₂W₉, SiCu₂W₉, and SiZn₂W₉), they display the same anion [B-α-SiW₉O₃₄]¹₀⁻, which indicates that formation of the (B-α-SiW₉O₃₄) fragment in a heated aqueous acidic medium is strongly favored.

Variable-temperature (2–300 K) magnetic susceptibility data at a magnetic field strength of 2 kg were collected for compound 1. The χₘ and μₑff vs. T plots are shown in Fig. 3. The μₑff at room temperature, 10.36 B.M., is larger than the spin-only value of 7.74 B.M. for the magnetically isolated high-spin octahedral Co₄ species.
due to the splitting of the $^4T_1$ term through spin–orbit coupling and local distortion of the octahedral sites. When the temperature is lowered, the value of $\mu_{\text{eff}}$ decreases smoothly, reaching a minimum at ca. 30 K ($\mu_{\text{eff}} = 9.43$ B.M.). Below 30 K, the $\mu_{\text{eff}}$ increases smoothly upon cooling further and reaches a maximum with 9.81 B.M. at 9 K. After this maximum, the $\mu_{\text{eff}}$ value decreases quickly, with 8.52 B.M. at 2 K. All the data measured from 2 to 300 K can be roughly fitted to a Curie–Weiss law with $C = 13.62$ cm$^3$ K mol$^{-1}$ and $\theta = -4.86$ K. Interestingly, here the plot of $\mu_{\text{eff}}$ vs. $T$ looks like that found in some irregular spin state complexes [10]. To the best of our knowledge, the case has not been previously reported in other sandwich-type TMSPs. However, the magnetic mechanism of 1 is completely distinct from that of those irregular spin state compounds.

In order to explain the observed magnetic property of the title compound, the following exchange scheme has been used:

![Exchange Scheme Diagram]

The interactions between the metal ions of the Co$_4$O$_{16}$ unit can be described by two exchange coupling constants $J$ and $J'$, where $J$ represents the interactions along the sides of the rhombus while $J'$ represents interactions along the short diagonal of the rhombus. As far as the existence of strong spin–orbit coupling for cobalt(II) ions was concerned, the magnetic data were analyzed by the following equations:

$$\chi_{\text{tet}} = \frac{N \beta^2}{3kT} \sum_i S_i (S_i + 1) (2S_i + 1) \exp(-E_i/KT)$$
$$\chi_{\text{Lig}} = \frac{N \beta^2}{3kT} \sum_i \frac{x(3.15x+3.18+2(2.84x+2.13)\exp(-3.75x)+4.7x-6.05)\exp(-6x)}{x[3+2\exp(-3.75x)+\exp(-6x)]}$$

$(x = \lambda/kT)$

$$\chi_M = \chi_{\text{tet}} + 4\chi_{\text{Lig}} - \frac{N \beta^2}{3kT} \left[ 4 \times \frac{3}{2} \left( \frac{3}{2} + 1 \right) \right]$$

Best fitting for the experimental data leads to $J = -0.77$ cm$^{-1}$, $J' = 3.12$ cm$^{-1}$, $g = 2.34$, $\lambda = -129$ cm$^{-1}$, with the agreement factor $R = \sum(\chi_{\text{obsd}} - \chi_{\text{calc}})^2 / \sum \chi_{\text{obsd}}^2 = 2.73 \times 10^{-4}$. The results reveal that magnetic interaction between Co1 and
Co1 is ferromagnetic while the Co1 and Co2 interaction is antiferromagnetic. The exchange interactions in the compound are slightly different from that found in the silicon-containing analogues [{SiM2W9O34(H2O)}2]12 (M = Mn2+, Cu2+) [1c], in which the central Mn4 unit exhibits antiferromagnetic (J = -1.77(5) cm⁻¹) as well as weak ferromagnetic (J’ = 0.08(2) cm⁻¹) Mn–Mn exchange interactions and the Cu–Cu exchange interactions are antiferromagnetic (J = -0.10(2) cm⁻¹, J’ = -0.29(2) cm⁻¹). It is believed that the major structural factors affecting exchange coupling in these compounds are the polyoxometalate matrixes since the polyoxometalate matrix may be considered a diamagnetic host encapsulating and thereby isolating magnetic clusters of transition metals.

In summary, we have synthesized a sandwich tetracobalt(II) substituted tungstosilicate which displays interesting isomerization ([β-SiW9O34]10⁻ → [B-α-SiW9O34]10⁻). The magnetism has been studied based on an isotropic Heisenberg model supplemented by a zero-field splitting term. It must be noted that it is an approximate method and usually insufficient for cobalt(II) exchange system due to the existence of spin anisotropy, and it is still a great challenge to well-understand the cobalt magnetism.

Acknowledgements

We gratefully acknowledge support of this work by National Natural Science Foundation of China (Nos. 20371027 and 20071020) and Tianjin Science Foundation (No. 033609211).

Appendix A. Supplementary material

X-ray crystallographic file in CIF format, complementary drawings, and complementary equations. Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.inoche.2007.08.025.
References


[7] IR: 987,951,904,876,850,757,723,532,512, and 486 cm$^{-1}$. Anal. Calcd (Found): K 7.1 (7.2), Co 4.3 (4.4), W 60.2 (59.7), Si 1.0 (1.1).

[8] Crystal data for the compound: Co$_{4}$H$_{48}$K$_{10}$O$_{91}$Si$_{2}$W$_{18}$, $M_r =$ 5496.58, monoclinic, $P2(1)/n$, $a = 12.296(3)$ Å, $b = 21.285(6)$ Å, $c =$ 15.792(4) Å, $\alpha =$ 90°, $\beta =$ 92.208(5)°, $\gamma =$ 90°, $V =$ 4130(2) Å$^3$, $Z = 2$, $\rho_{calc} =$ 4.420 g cm$^{-3}$, $2\theta_{max} =$ 26.39° (-12 ≤ h ≤ 15, -26 ≤ k ≤ 23, -19 ≤ l ≤ 19), $T =$ 294(2) K, $F(000) =$ 4868, GOF = 0.975, $R_1 =$ 0.0576, $wR_2 =$ 0.1492. CSD 416730.


List of Figures

Fig. 1. Ball and stick (left) and polyhedral (right) representations of 1. The color code is as follows: cobalt (purple), tungsten (black), silicon (blue), and oxygen (red). (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

Fig. 2. ORTEP drawing of the central Co₄O₁₆ fragment of 1; thermal ellipsoids are drawn at the 30% probability level.

Fig. 3. \(\chi_M(\Delta)\) vs. \(T\) and \(\mu_{eff}(O)\) vs. \(T\) plots for the compound 1; the solid lines represent the theoretical values based on the equations.
Fig. 1
Fig. 3