

1 Enhanced electric resistivity and dielectric energy storage by  
2 vacancy defect complex

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30

1 **Abstract**

2 The presence of uncontrolled defects is a longstanding challenge for achieving high electric  
3 resistivity and high energy storage density in dielectric capacitors. In this study, opposite to  
4 conventional strategies to suppress defects, a new approach, i.e., constructing defects with deeper  
5 energy levels, is demonstrated to address the inferior resistivity of BiFeO<sub>3</sub>-based dielectric films.  
6 Deep-level vacancy complexes with high charge carrier activation energies are realized via  
7 deliberate incorporation of oxygen vacancies and bismuth vacancies in low-oxygen-pressure  
8 deposited films. This method dramatically increases the resistivity by ~4 orders of magnitude  
9 and the breakdown strength by ~150%, leading to a ~460% enhancement of energy density (from  
10 14 to 79 J cm<sup>-3</sup>), as well as improved efficiency and performance reliability. This work reveals  
11 the significance of rational design and precise control of defects for high-performance dielectric  
12 energy storage. The deep-level vacancy complex approach is generalizable to wide ranges of  
13 dielectric systems and functional applications.

14 **Keywords:** dielectric energy storage, defect, vacancy complex, resistivity, energy density,  
15 BiFeO<sub>3</sub>

16

## 1 **1. Introduction**

2 Dielectric capacitors with ultrafast charging-discharging speed are fundamental energy  
3 storage components in electronics and electrical power systems [1,2]. To realize device  
4 miniaturization, cost reduction and performance enhancement, dielectrics with high energy  
5 storage densities have been extensively pursued [3–6]. In the development of energy-storage  
6 dielectrics, material defect emerges as a longstanding obstacle [7,8]. Defects usually act as  
7 shallow charge carrier traps and severely degrade the electrical resistivity of dielectrics, which  
8 cause large leakage current, limit the breakdown strength, increase dielectric loss and destabilize  
9 dielectric polarization, consequentially deteriorating the energy density and operation reliability  
10 [9,10].

11 Fine control of defects, i.e., defect engineering, is thus highly demanded in dielectrics, yet  
12 the progress is rather limited compared with that in the semiconductor industry [11]. The  
13 challenges lie in the complexity of defects in dielectrics, for example, oxide dielectrics such as  
14  $\text{PbTiO}_3$  (PTO) and  $\text{BiFeO}_3$  (BFO) contain complicated defect configurations including but not  
15 limited to cation vacancies, oxygen vacancy, ion valence variation, etc [12]. Moreover, the  
16 category, concentration and distribution of the defects can be sensitively altered by the specific  
17 composition, synthesis process and thermal history. These defects, even with a very low  
18 concentration, can make a remarkable impact on the electric properties of the insulating  
19 dielectrics. The widely used defect-engineering strategies in semiconductors, e.g., chemical  
20 doping to interact with or counteract existing defects, are less effective and sometimes act in  
21 contradictory ways in oxide dielectrics. For example, Ti doping in BFO in different studies  
22 separately reported increased and decreased resistivity [13,14], which was linked to the varied  
23 dominant defects in the pristine BFO, i.e., oxygen vacancies ( $V_{\text{O}}$ ) [15,16] and bismuth vacancies

1 ( $V_{Bi}$ ) [17,18] that result in  $n$ -type and  $p$ -type conduction, respectively. Therefore, more  
2 determinative defect engineering strategies are desirable to enhance the resistivity and energy  
3 storage performance of dielectrics.

4 Considering that the defects in oxide dielectrics are rather shallow in the bandgap (e.g,  
5  $\sim 0.13$  eV for  $V_{Bi}$  and  $\sim 0.60$  eV for  $V_O$ ) [19,20] but eliminating them is thermodynamically  
6 unpractical, transforming the shallow defects into deeper ones can be a feasible approach to  
7 reduce the charge carrier activation and to increase the resistivity. To this end, defect complexes  
8 that consist of inversely charged defects with electrostatic mutual attraction can be promising  
9 strategies. It has been reported that  $He^{2+}$  bombardment of oxide films such as BFO [21], PTO  
10 [22], and  $Pb(Mg_{1/3}Nb_{2/3})O_3$ - $PbTiO_3$  (PMN-PT) [4] can induce cation and oxygen vacancies  
11 simultaneously that form deep-energy-level defect complexes. For example, in the case of PTO  
12 films,  $V_O^{\cdot\cdot} - V_{Pb}^{\prime\prime}$  and  $V_O^{\cdot\cdot} - V_{Ti}^{\prime\prime\prime} - V_O^{\cdot\cdot}$  with energy levels of  $\sim 1.0$  eV were induced, while the  
13 shallow-level defects  $Pb^{3+}$  ( $\sim 0.26$  eV) and  $V_{Pb}^{\prime\prime}$  ( $\sim 0.56$  eV) were suppressed. Thus, the films  
14 realize enhancements of resistivity by orders of magnitude [22].

15 In this work, we demonstrate a milder, less costly, yet efficient way to introduce defect  
16 complexes via in-situ control of the material fabrication, so as to improve the resistivity and  
17 energy storage performance of dielectrics. We employ the pulsed laser deposition (PLD)  
18 technique, in which multiple parameters can be precisely controlled to modulate the defects in  
19 deposited films [23–26]. By controlling the oxygen partial pressure ( $pO_2$ ) during deposition, we  
20 construct  $V_{Bi} - V_O$  defect complexes in BFO-based dielectric films. Contrary to the general  
21 perception that high  $pO_2$  ( $>10^{-1}$  Torr) is favourable to suppress the  $V_O$  concentration and  
22 increase the resistivity of BFO, we deliberately utilize lower  $pO_2$ s at the level of  $10^{-2}$  Torr to  
23 introduce both  $V_O$  and  $V_{Bi}$ , thus forming  $V_{Bi} - V_O$  complexes with higher carrier activation

1 energies. The film resistivity is thus enhanced by  $\sim 4$  orders of magnitude, which increases the  
2 breakdown strength by  $\sim 150\%$  (to  $\sim 4 \text{ MV cm}^{-1}$ ) and thus breaks the bottleneck of BFO-based  
3 films for high-voltage dielectric energy storage. Ultimately, great enhancement of energy density  
4 from  $14$  to  $79 \text{ J cm}^{-3}$ , together with improved efficiency, performance reliability and thermal  
5 stability, is achieved in the optimized BFO-based dielectrics with  $V_{Bi} - V_O$  complexes.

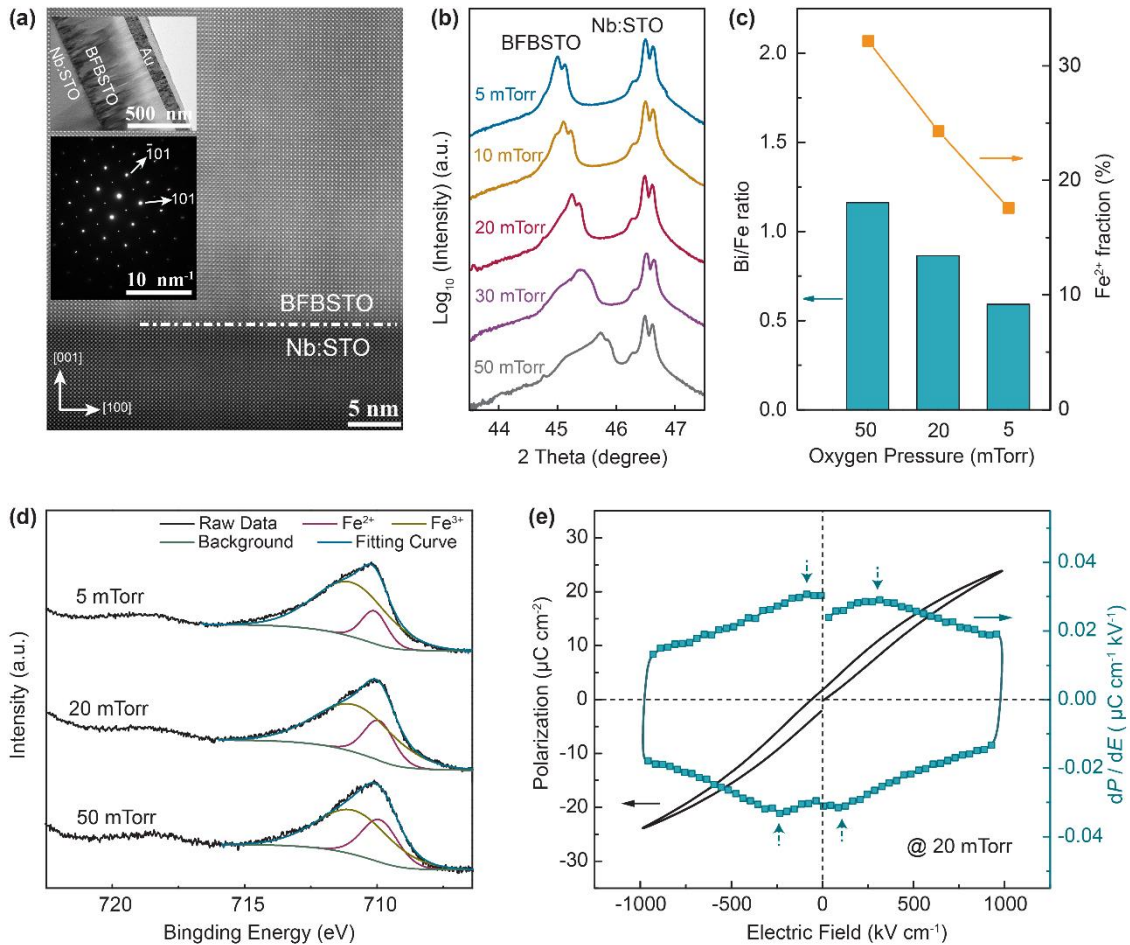
## 6 **2. Results and discussion**

7 BFO-based films were grown on (001)-oriented 0.7 wt% Nb-doped SrTiO<sub>3</sub> (Nb:STO)  
8 substrates with  $pO_2$  varied from 50 to 5 mTorr. The composition of the films was designed as  
9 30BFO-35BaTiO<sub>3</sub> (BTO)-35SrTiO<sub>3</sub> (STO) (denoted as BFBSTO) because it forms a relaxor  
10 ferroelectric (RFE) with unique polymorphic nanodomain structures and great energy storage  
11 potential [3]. All the BFBSTO films are high-quality epitaxial single-crystal perovskites, as  
12 affirmed by the results of scanning transmission electron microscopy (STEM, Fig. 1a) and X-ray  
13 diffraction (XRD, Fig. S1). With the decrease of  $pO_2$ , the films show a clear shift of the (002)  
14 XRD peak towards a smaller angle (Fig. 1b), corresponding to a  $c$ -axis lattice expansion from  
15  $3.96$  to  $4.03 \text{ \AA}$ . As the films have identical nominal compositions from the same PLD target, the  
16 lattice expansion is ascribed to ion vacancies induced during film deposition, which reduce the  
17 electrostatic constraint on the lattice [27]. In the BFBSTO films grown in reduced  $pO_2$ , more  $V_O$   
18 should be introduced jointly by the BFO, BTO and STO components, in accordance with  
19 previous reports [15,28,29]. As for the possible cation vacancies, a reduction of the Bi/Fe ratio  
20 with decreased  $pO_2$  is observed in the films via the inductively coupled plasma mass  
21 spectrometry (ICP-MS, Fig. 1c). The film composition transforms from slight Bi-excess (Bi/Fe  
22  $\sim 1.15$ ) at 50 mTorr to slight Bi-deficiency (Bi/Fe  $\sim 0.86$ ) at 20 mTorr, and to strong Bi-deficiency  
23 (Bi/Fe  $\sim 0.59$ ) at 5 mTorr. This evolution agrees with previous reports of PLD-fabricated BFO

1 films in which the Bi/Fe ratio was reduced from 1 to ~0.6 as  $pO_2$  decreases to 1 mTorr [30].  
2 Energy dispersive X-ray spectroscopy (EDS) shows a similar trend of Bi deficiency with  
3 lowered  $pO_2$  in our films but a negligible variation of the (Ba+Sr)/Ti ratio near ~1.0 (Fig. S2a),  
4 which is ascribed to the much stronger volatility of the Bi than other elements. Therefore,  $V_{Bi}$  is  
5 introduced along with  $V_O$  during the low- $pO_2$  deposition, which can be understood from two  
6 aspects. For one thing,  $V_{Bi}$  is triggered to maintain charge compensation in the films, as  
7 expressed by  $2Bi_{Bi} + 3O_O \rightarrow 2V_{Bi} + 3V_O + 2Bi(g) + \frac{3}{2}O_2$ . For another, the lowered  $pO_2$   
8 during PLD leads to less scattering and higher kinetic energy of the adatoms arriving the  
9 substrate, which results in knock-on damage on the deposited films (similar to the ion  
10 bombardment) [22], facilitates the formation and alignment of defect complexes, and, thus,  
11 causes lattice expansion [31].

12 It is generally regarded that in ferrite or titanate materials,  $V_O$  donates electrons and induces  
13 valence reduction of neighbouring Fe and Ti atoms ( $Fe^{3+}$  to  $Fe^{2+}$ , and  $Ti^{4+}$  to  $Ti^{3+}$ ) [28,32].  
14 However, in our BFBSTO films, it is found that as the  $V_O$  concentration increases (with  $pO_2$   
15 lowered from 50 to 5 mTorr), the  $Fe^{2+}$  fraction does not increase but in fact decreases from 32%  
16 to 18% (Fig. 1c), while the Ti element valence remains +4, as demonstrated by X-ray  
17 photoelectron spectroscopy (XPS, Fig. 1d and Fig. S2b). This indicates that the  $V_O$  in the  
18 BFBSTO films is more likely to interact with  $V_{Bi}$ , forming  $V_{Bi} - V_O$  defect complexes. Direct  
19 observation of defect complexes is so far technically not possible; yet their existence can be  
20 indirectly inferred from polarization-electric field ( $P$ - $E$ ) loops. Conventional RFE dielectrics  
21 present  $P$ - $E$  loops with two  $dP/dE$  (differential of  $P$  regarding  $E$ ) peaks, which correspond to the  
22 switching of polarization and FE domains. While in RFE dielectrics containing defect  
23 complexes, which are equivalent to dipoles consisting of positive and negative charges, FE

1 domains can be electrostatically pinned by the defect complexes [33]. This interaction, also  
 2 called ageing, hinders the domain switching and causes pinched  $P-E$  loops with 4 peaks in the  
 3  $dP/dE$  curve [34]. The BFBSTO film grown under the  $pO_2$  of 50 mTorr exhibits a normal  $P-E$   
 4 loop without pinching (Fig. S2), agreeing with the finding that it contains mainly  $V_O$ . For films  
 5 grown under  $pO_2$ s of 20 and 5 mTorr, the  $P-E$  loops become pinched with 4  $dP/dE$  peaks  
 6 presented (Fig. 1e and Fig. S3), indicating the existence of  $V_{Bi} - V_O$  complexes. Similar  
 7 pinched  $P-E$  loops have also reported in pure BFO [35]. We notice that the pinch feature in our  
 8 BFBSTO films is not as prominent as in pure BFO; this is because the nanodomains in the RFE  
 9 BFBSTO are more dynamic and thus less strongly pinned by the defect complexes, compared  
 10 with the ordinary FE domains in BFO.



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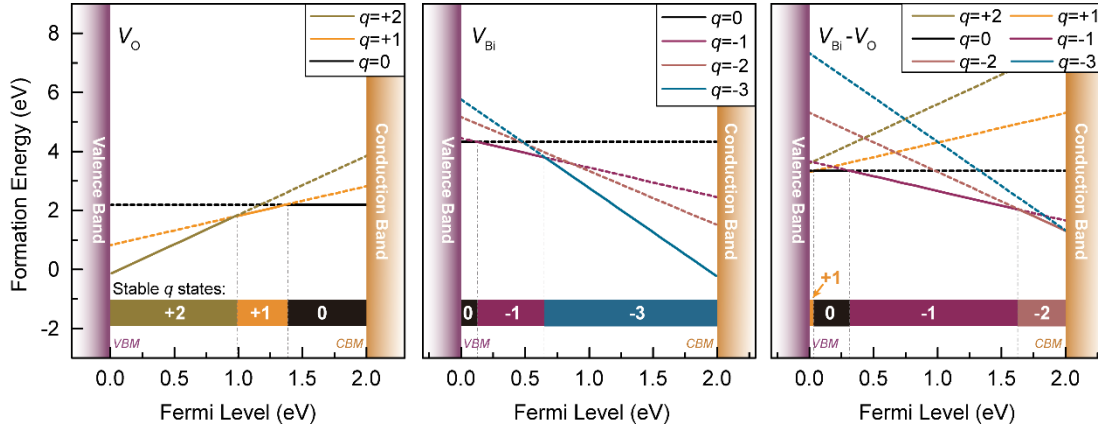
2 **Fig. 1.** (a) STEM atomic image of a BFBSTO film grown under a  $pO_2$  of 20 mTorr. The insets  
3 are the low-magnification morphology of the film and the selected area electron diffraction at the  
4 BFBSTO/Nb:STO interface. (b) XRD patterns of the (002) diffraction peaks of BFBSTO films  
5 grown under varied  $pO_2$ s. (c) Evolution of the Bi/Fe ratio and the  $Fe^{2+}$  fraction with  $pO_2$ . (d)  
6 XPS spectra of Fe  $2p_{3/2}$  peak and fitting of Fe valences. (e) Pinched  $P$ - $E$  loop of the BFBSTO  
7 film with the  $pO_2$  of 20 mTorr.

8

9 To understand the formation mechanism of the  $V_{Bi} - V_O$  complex and its influence on  
10 electrical properties, we conducted first-principles calculations based on density functional  
11 theory. To simplify the calculation, the BTO and STO components were reasonably omitted, and  
12 a  $2 \times 2 \times 2$  supercell of BFO was employed (Fig. S4a). A Bi and an O atom were removed to  
13 simulate  $V_{Bi}$  and  $V_O$ , respectively. Details of the calculation can be found in the Experimental  
14 section. We first demonstrated that in all configurations of the  $V_{Bi} - V_O$  complex, the nearest  
15 neighbouring configuration, with the smallest  $V_{Bi} - V_O$  distance of 2.31 Å, has the lowest total  
16 energy (Fig. S4b). The electrostatic attraction between the vacancies is revealed to be the key  
17 driving force for the formation of the defect complexes [36]. This is evidenced by the negative  
18 association energy of the  $V_{Bi} - V_O$  complex regardless of how its charge state ( $q$ ) varies  
19 between +2 and -3 (Fig. S5). The largest association energy (-1.95 eV), corresponding to the  
20 strongest electrostatic force, is obtained when  $q = -1$ , i.e., the configuration with fully ionized  
21 vacancies  $V_{Bi}''' - V_O''$ . Moreover, we derived the energy levels ( $\epsilon$ ) and charge carrier activation  
22 energies ( $E_a$ ) of the individual  $V_{Bi}$ ,  $V_O$ , and the  $V_{Bi} - V_O$  complex, from the calculated plots of

1 defect formation energies between the valence band maximum (VBM) and conduction band  
 2 minimum (CBM) (Fig. 2). Here,  $\epsilon_{\text{VBM}}$  is set as 0 eV and the  $\epsilon_{\text{CBM}}$  of BFO (i.e., bandgap) is  
 3 calculated to be 2.0 eV. For a defect  $D$  that transforms from  $q$  to  $q'$ , the energy level  $\epsilon_D(q/q')$  is  
 4 given by the intersection of corresponding plots of formation energy. For example,  $\epsilon_{V_O}(0/  
 5 +1) = 1.37$  eV and  $\epsilon_{V_{Bi}}(0/-1) = 0.12$  eV. The carrier activation energy  $E_{a,D}(q/q')$  equals  
 6  $\epsilon_{\text{CBM}} - \epsilon_D(q/q')$  for donors and  $\epsilon_D(q/q') - \epsilon_{\text{VBM}}$  for acceptors, respectively. Therefore, the  
 7 activation energies of the  $V_O$  donor  $E_{a,V_O}(0/+1)$  and the  $V_{Bi}$  acceptor  $E_{a,V_{Bi}}(0/-1)$  are  
 8 0.63 and 0.12 eV, respectively, in good agreement with previous theoretical calculations [36,37].  
 9 As for the  $V_{Bi} - V_O$  complex, one can first notice that the  $q = -1$  state possesses a much larger  
 10 range of stability with the lowest forming energy compared with other states, in accordance with  
 11 the largest association energy of the configuration  $V_{Bi}''' - V_O''$ . As a result, the activation energy  
 12 levels of  $V_{Bi}''' - V_O''$  are shifted deeper into the bandgap. More specifically, if  $V_{Bi}''' - V_O''$  behaves  
 13 as a donor, the activation energy should be  $E_{a,V_{Bi}-V_O}(-1/0) = \epsilon_{\text{VBM}} - \epsilon_{V_{Bi}-V_O}(-1/0) = 2.0$   
 14  $- 0.32$  eV = 1.68 eV, which is greatly enhanced compared with that of the individual  $V_O$  (0.63  
 15 eV). Similarly, considering it as an acceptor,  $E_{a,V_{Bi}-V_O}(-1/-2) = \epsilon_{V_{Bi}-V_O}(-1/-2) - \epsilon_{\text{VBM}}$   
 16  $= 1.65$  eV, which is also much larger than that of the individual  $V_{Bi}$  (0.12 eV). Therefore, the  
 17 activation of both electron and hole carriers are greatly suppressed by the  $V_{Bi} - V_O$  vacancy  
 18 complex, which can significantly improve the resistivity of BFO-based materials.

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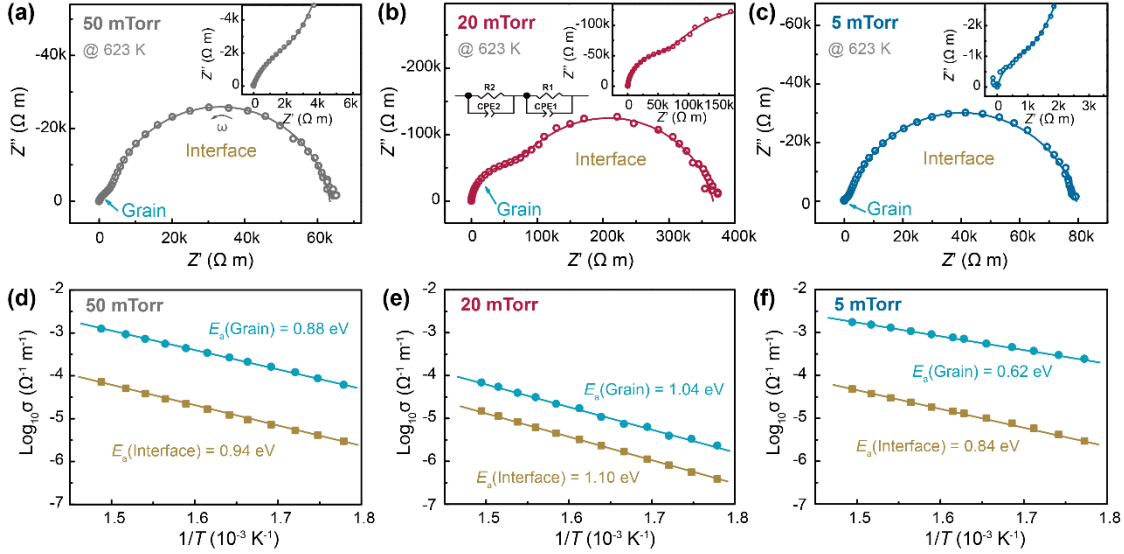
2 **Fig. 2.** First-principles calculations of the formation energies of  $V_O$ ,  $V_{Bi}$  and  $V_{Bi} - V_O$  defects  
 3 as a function of the Fermi level for different charge states. The solid lines indicate the charge  
 4 state  $q$  with the lowest formation energy, i.e., the highest stability, while the dash lines indicate  
 5 states with higher formation energy and lower stability. The stable ranges of the defect charge  
 6 state  $q$  between VBM and CBM are displayed.

7

8 To experimentally characterize the enhancement of resistivity by  $V_{Bi} - V_O$  complexes,  
 9 impedance spectra of the BFBSTO films were measured, as shown in Fig. 3a-c. The insets show  
 10 the fine spectrum structures at the high frequency ( $\omega$ ) range. All the spectra can be fitted with  
 11 two in-series R-CPE units (inset of Fig. 3b), with each R-CPE unit consisting of an ideal resistor  
 12 (R) and a constant-phase element (CPE) connected in parallel. Because the films are single  
 13 crystalline, the effect of grain boundaries is negligible. The lower- $\omega$  semi-circle in the impedance  
 14 plots is ascribed to the film-electrode interfaces, and the higher- $\omega$  one to the grains in the film  
 15 [38]. For the 50-mTorr film at 623 K (Fig. 3a), the fitted interfacial resistivity R1 and grain  
 16 resistivity R2 are 60.5 and 2.9 k $\Omega$  m, respectively, indicating that the interfaces are the main

1 contributors to the film resistivity, while the grains are much less resistive due to the existence of  
2  $V_O$ . This result agrees well with previous observations in BFO films [38]. The reciprocals of R1  
3 and R2 (i.e., conductivity  $\sigma$ ) are fitted regarding the temperature based on the Arrhenius law  
4 (Fig. 3d). The activation energies  $E_a$  for the interface and grain conductivity are 0.94 and 0.88  
5 eV, respectively, consistent with the reported values of  $V_O$  [39,40]. For the 20-mTorr film, R1  
6 and R2 increase remarkably to 308.9 and 58.2 k $\Omega$  m (Fig. 3b), which are 5 and 20 times the  
7 values of the 50-mTorr film, respectively. Besides,  $E_a$  of the interface and grain conductivity  
8 increases to 1.10 and 1.04 eV, respectively (Fig. 3e). The impedance becomes jointly contributed  
9 from the grain and interface in the 20-mTorr film. This is because of the formation of  $V_{Bi} - V_O$   
10 complexes, which, for one thing, greatly increases the grain resistivity by compensating  $V_O$  and  
11 suppressing charge activation; and for another, improves the interface resistivity possibly by  
12 reducing the accumulation of oxygen vacancy near the interfaces and increasing the Schottky  
13 barrier height [41]. As the  $pO_2$  further reduces to 5 mTorr (Fig. 3c), the resistivity is again  
14 degraded (R1 = 68.4 and R2 = 1.3 k $\Omega$  m), and the activation energies also decrease to 0.84 and  
15 0.62 eV, respectively (Fig. 3f). We propose that there is an upper limit for the stable formation of  
16  $V_{Bi} - V_O$  complexes. The excessive  $V_O$  defect induced by the excessively low  $pO_2$  become  
17 uncompensated, which decreases the resistivity. This proposition is supported by the enhanced  
18 resistivity of the 5-mTorr BFBSTO film after further annealing in high-pressure  $O_2$  (Fig. S6).

1



2 **Fig. 3.** Impedance spectra and conduction activation energy fitting of BFBSTO films. Complex  
 3 impedance ( $Z' - Z''$ ) spectra measured at 623 K for films grown under  $pO_2$ s of (a) 50 mTorr, (b)  
 4 20 mTorr, and (c) 5 mTorr. The insets show the fine spectrum structures in the high frequency  
 5 range. The circles are experimental results, and the lines are fitting results based on the  
 6 equivalent circuit with two R-CPE units in-series (inset of Fig. 3b). Fittings of the interface and  
 7 grain conduction regarding the temperature  $T$  of BFBSTO films with  $pO_2$ s of (d) 50 mTorr, (e)  
 8 20 mTorr and (f) 5 mTorr.

9

10 The DC leakage current of the BFBSTO films was also measured (Fig. 4a), whose evolution  
 11 agrees with the impedance spectra. At an electric field of  $1 \text{ MV cm}^{-1}$ , the 50-mTorr film shows a  
 12 high leakage current of  $2 \times 10^{-2} \text{ A cm}^{-2}$ , comparable with those of previously reported BFO-based  
 13 films [42]. As the  $pO_2$  decreases to 20 mTorr, the leakage current is distinctly suppressed by  
 14 almost 4 orders of magnitude. The ultralow current density of  $5 \times 10^{-6} \text{ A cm}^{-2}$  (at  $1 \text{ MV cm}^{-1}$ )  
 15 reaches the lowest level of perovskite dielectric films, which demonstrates the effectiveness of

1  $V_{Bi} - V_O$  complexes in improving the resistivity. With the  $pO_2$  further reducing to 5 mTorr, the  
2 leakage current increases again to  $3 \times 10^{-3}$  A  $cm^{-2}$  due to the excess  $V_O$ . Mechanisms of the  
3 leakage currents were also analysed (Fig. S7). At lower fields, the 20-mTorr film follows Ohm's  
4 law, while the 50- and 5-mTorr films exhibit space-charge-limited conduction, which is ascribed  
5 to charge carriers from  $V_O$ . At higher fields, the leakage current of the 50- and 5-mTorr films is  
6 dominated by the interfacial Schottky emission. While in the 20-mTorr film, interfacial Schottky  
7 emission and bulk Poole-Frenkel emission take effect jointly [43]. These phenomena match well  
8 with the impedance spectra (Fig. 3), indicating that  $V_{Bi} - V_O$  complex increases both the grain  
9 and interface resistivity. In addition, in BFO-based films of other compositions we observed  
10 similar trends of leakage current with  $pO_2$  (Fig. S8). In 40BFO-60STO films, the lowest leakage  
11 current is realized at a  $pO_2$  of 10 mTorr; while in 25BFO-75BTO films, the best  $pO_2$  is 25 mTorr.  
12 These results prove that the construction of deep  $V_{Bi} - V_O$  complexes in the optimized  $pO_2$  is a  
13 universal technique for resistivity enhancement in BFO-based films.

14 Since low resistivity is known to deteriorate both electrical and thermal breakdown of  
15 dielectrics [44], it is not surprising that the 50-mTorr film exhibits a low characteristic  
16 breakdown field  $E_b$  of only  $1.6$  MV  $cm^{-1}$ , as derived from the two-parameter Weibull distribution  
17 (Fig. 4b). With the improvement of resistivity by vacancy complexes, the 20-mTorr film exhibits  
18 a remarkable enhancement of  $E_b$  by 150%, reaching a maximum value of  $4.0$  MV  $cm^{-1}$ . This  
19 value is comparable with those of other insulating perovskite dielectric films, e.g., PZT and BZT  
20 [45,46]. The great improvements of resistivity and breakdown field overcome the bottleneck of  
21 BFO-based dielectrics for high-voltage energy storage applications.

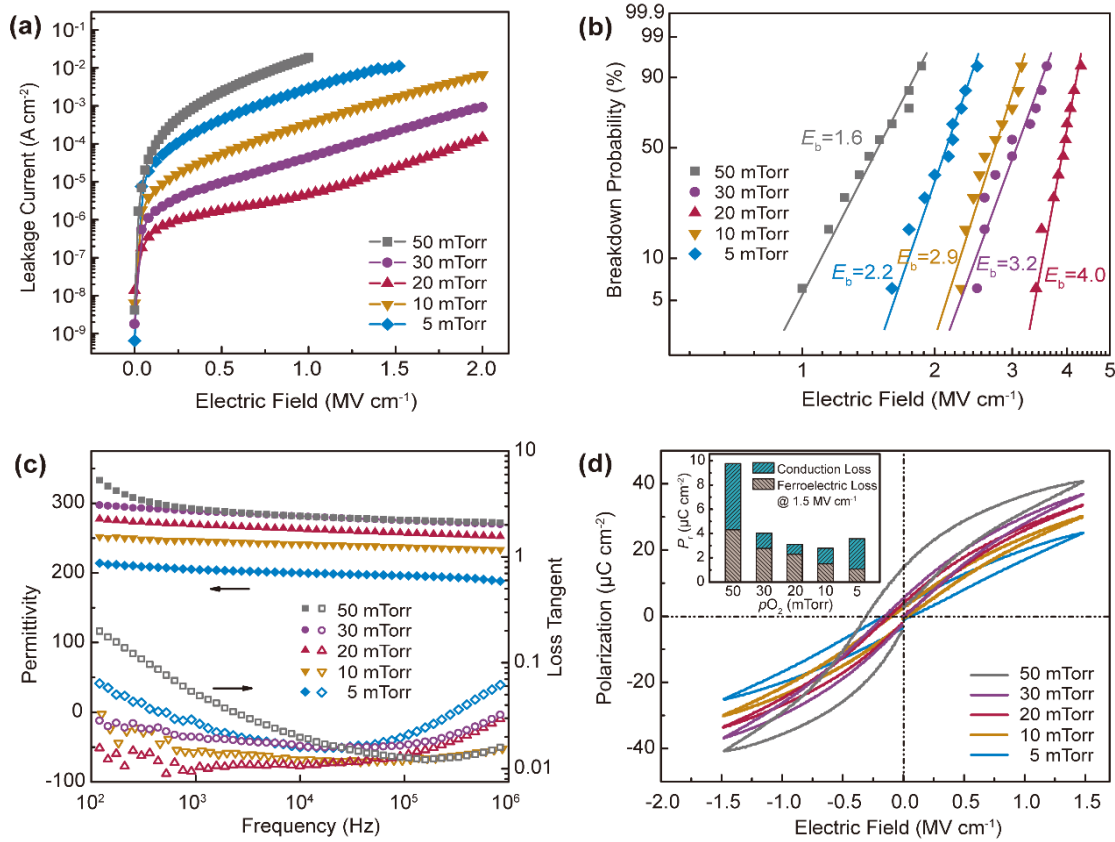
22 Because dielectric energy storage is closely linked to not only the breakdown strength but  
23 also the dielectric polarization (see schematic in Fig. S9), we investigated the dielectric

1 properties of the BFBSTO films. As shown in Fig. 4c, the dielectric permittivity decreases from  
2 293 to 205 (at 1 kHz) as the  $pO_2$  decreases. This can be ascribed to the increased Bi deficiency,  
3 as the Bi  $6s^2$  lone electron pair is one of the main contributors to the ferroelectric polarization of  
4 BFO [47]. The 50-mTorr and 5-mTorr films exhibit high dielectric loss at low frequencies, e.g.,  
5 0.23 and 0.07 at 100 Hz, respectively, which originates from the  $V_O$ -related space charge [44].  
6 The 20-mTorr film with  $V_{Bi} - V_O$  complexes greatly suppresses the space charge and thus  
7 exhibits a loss tangent of as low as 0.02, which reaches the best level of perovskite dielectric  
8 films [45]. The temperature dependence of the dielectric properties was also shown in Fig. S10.  
9 The 50- and 5-mTorr films exhibit remarkable elevation of permittivity and deterioration of loss  
10 tangent with increased temperature, which is ascribed to the thermal activation of the  $V_O$ -related  
11 space charges. As comparison, the 20-mTorr film maintains stable permittivity and low loss  
12 tangent at temperatures even up to 480 K, benefitting from the large activation energy of the  
13  $V_{Bi} - V_O$  complexes.

14 Fig. 4d exhibits the polarization behaviours of the BFBSTO films. A drop of the maximum  
15 polarization ( $P_m$ ) from 41 to 25  $\mu C\ cm^{-2}$  at the electric field of 1.5  $MV\ cm^{-1}$  is observed,  
16 associated with the Bi deficiency at lower  $pO_2$ s. The ferroelectric nonlinearity is also weakened:  
17 the  $P-E$  loop of the 5-mTorr film becomes similar to a paraelectric compared with that of the  
18 50-mTorr film. The remnant polarization ( $P_r$ , inset of Fig. 4d) shows a similar trend as the  
19 low-frequency dielectric loss (Fig. 3c) and the leakage current (Fig. 4a). This is reasonable as  $P_r$   
20 is contributed by both the ferroelectric loss and conduction loss [32]. The portion of  $P_r$  from the  
21 ferroelectric loss is characterized by the positive-up-negative-down (PUND) technique (Fig. S11  
22 and Table S1), which shows a monotonous decrease with  $pO_2$  (inset of Fig. 4d), again proving  
23 the depression of ferroelectricity by Bi deficiency. The other part of  $P_r$  from conduction loss can

1 then be obtained, which is minimal in the 20-mTorr film but visible in the 50- and 5-mTorr  
 2 films. These observations suggest that the formation of  $V_{Bi} - V_O$  complex is at the slight  
 3 sacrifice of polarization. This is, however, a worthwhile compromise as the dielectric loss and  $P_r$   
 4 are greatly suppressed, which is critical for high-performance energy storage.

5



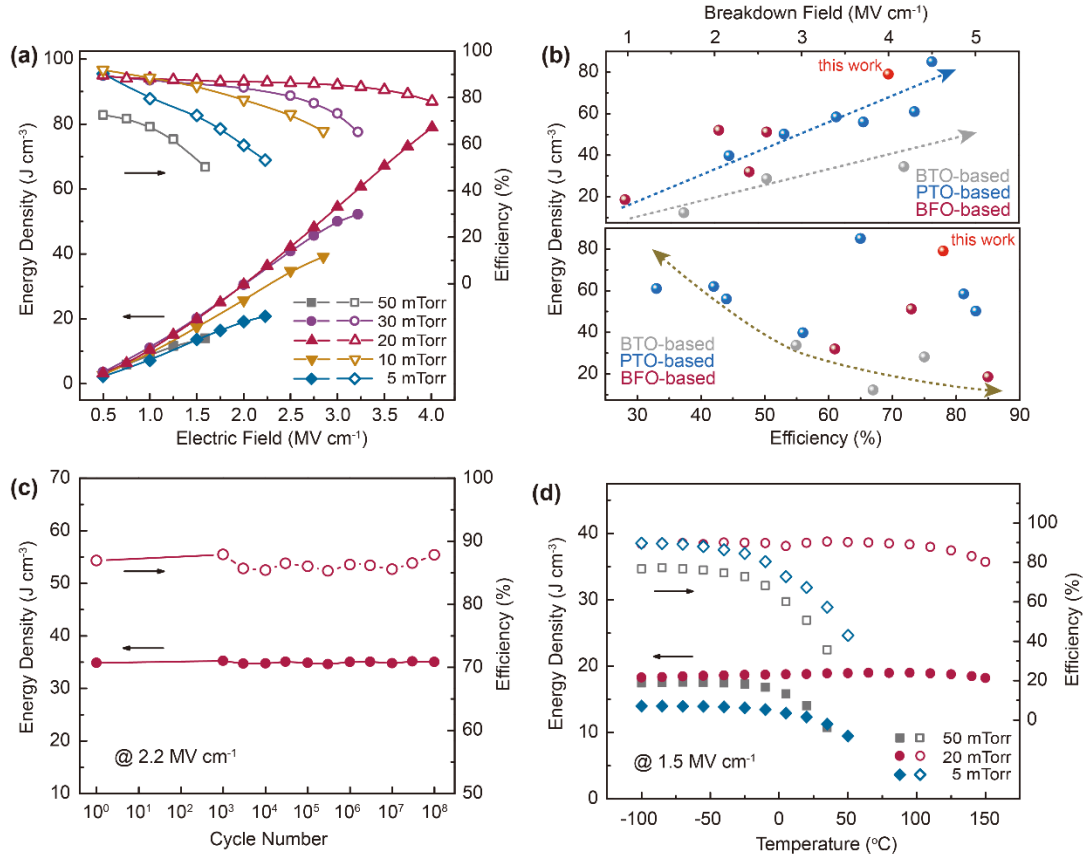
6 **Fig. 4.** Electric and dielectric properties of BFBSTO films. (a) Leakage current density as a  
 7 function of biased DC electric field. (b) Two-parameter Weibull distribution analysis of the  
 8 breakdown strength. (c) Frequency-dependent dielectric permittivity and loss tangent; (d)  $P-E$   
 9 loops. The inset shows the  $P_r$  contributed by ferroelectric loss and conduction loss of the  
 10 BFBSTO films.

11

1 To obtain the energy storage performance, we measured the  $P$ - $E$  loops of the BFBSTO films  
2 at fields up to  $E_b$  (Fig. S12). The energy density  $U_e$  and efficiency  $\eta$  are calculated based on the  
3 equations  $U_e = \int_{P_r}^{P_m} P dE$  and  $\eta = U_e / (U_e + U_{\text{loss}})$  (Fig. S9), where  $U_{\text{loss}}$  is the hysteresis energy  
4 loss (including ferroelectric loss and conduction loss) in the charging-discharging cycle. The  
5 results are displayed in Fig. 5a. At the same field of  $1.5 \text{ MV cm}^{-1}$ , the 20-mTorr film exhibits an  
6 optimally suppressed conduction loss, and thus realizes greatly enhanced  $U_e$  of  $20 \text{ J cm}^{-3}$  and  $\eta$  of  
7 87% compared with other films (e.g.,  $U_e$  of  $13 \text{ J cm}^{-3}$  and  $\eta$  of 53% in the 50-mTorr film).  
8 Moreover, the enhanced  $E_b$  enabled by resistivity improvement further raises the ceiling of  
9 energy storage capability for the 20-mTorr film, which realizes an ultrahigh  $U_e$  of  $79 \text{ J cm}^{-3}$  at  
10 the  $E_b$  of  $4.0 \text{ MV cm}^{-1}$ , representing a 4.6-time enhancement compared with the 50-mTorr film  
11 ( $14 \text{ J cm}^{-3}$  at the  $E_b$  of  $1.6 \text{ MV cm}^{-1}$ ). In addition, while other films experience a sharp decline of  
12  $\eta$  due to the deteriorated  $U_{\text{loss}}$  at high  $E$ , a high  $\eta$  of  $>78\%$  is maintained in the 20-mTorr film.  
13 These enhancements in energy storage performance clearly demonstrate the significance of  
14 improved resistivity by vacancy complexes. A comparison of the energy performance of our  
15 BFBSTO film with representative BTO-based, PTO-based and other BFO-based films is shown  
16 in Fig. 5b [45,48–58]. Note that the growth trend of  $U_e$  with  $E$  in BFO-based films is at the same  
17 level as PTO-based films and is more rapid than that of BTO-based films. This is because the  
18 spontaneous polarization of BFO ( $\sim 90 \mu\text{C cm}^{-2}$ ) is comparable with that of PTO ( $78 \mu\text{C cm}^{-2}$ )  
19 and is much larger than that of BTO ( $26 \mu\text{C cm}^{-2}$ ), rendering BFO-based films promising as  
20 Pb-free energy-storage dielectrics. The vacancy complex strategy in this work breaks the  
21 limitations of low resistivity and small  $E_b$  of BFO-based films, making their  $U_e$  competitive with  
22 Pb-based films. Furthermore, the vacancy complex strategy also facilitates the breach of the

1 adverse coupling between  $U_e$  and  $\eta$ , realizing concurrent optimization and making the  
2 BFO-based dielectric films even more outstanding for practical utilization (Fig. 5b).

3 The performance reliability with respect to the charging-discharging cycles and working  
4 temperatures was also characterized. The 20-mTorr BFBSTO film undergoes a  
5 charging-discharging process over  $1 \times 10^8$  cycles at an electric field of  $2.2 \text{ MV cm}^{-1}$  with  
6 negligible degradation of energy performance (Fig. 5c and Fig. S13). This should be ascribed to  
7 the RFE nanodomain structures of BFBSTO that suppress domain pinning and polarization  
8 fatigue [3]. Besides, the high resistivity and high  $E_b$  of the 20-mTorr film prevents failure during  
9 the cycles, which happens easily in the 5-mTorr and 50-mTorr films. The energy storage  
10 performance of the BFBSTO films between  $-100$  and  $150 \text{ }^\circ\text{C}$  is also presented (Fig. 5d and Fig.  
11 S14). The 20-mTorr film shows stable  $U_e$  and  $\eta$ , with variations of  $<4\%$  and  $<10\%$ , respectively,  
12 over the whole temperature range. This excellent stability is first due to the RFE nature of  
13 BFBSTO films that realizes thermally stable dielectric and polarization behaviours. More  
14 importantly, the  $V_{Bi} - V_O$  complexes strongly trap charge carriers, reduce carrier activation, and  
15 mitigate the conduction loss at high temperatures. This is in stark contrast to the films of 50 and  
16 5 mTorr, in which the leakage current increases dramatically with temperature (Fig. S15),  
17 resulting in distorted  $P-E$  loops and degenerated energy storage performance [59].



1

2 **Fig. 5.** Dielectric energy storage performance of the BFBSTO films. (a) Energy density and  
 3 efficiency of the films as functions of the electric field; (b) Comparison of energy storage  
 4 performance of the BFBSTO film in this work with representative PTO-based, BTO-based, and  
 5 other BFO-based dielectric films; (c) Energy storage performance of the 20-mTorr BFBSTO film  
 6 at an electric field of 2.2 MV cm<sup>-1</sup> over 10<sup>8</sup> charging-discharging cycles; (d) Temperature  
 7 dependence of the energy storage performance of the BFBSTO films at a field of 1.5 MV cm<sup>-1</sup>.

### 8 3. Conclusions

9 To conclude, we demonstrate an effective approach of constructing deep-level  $V_{Bi} - V_O$   
 10 complexes to enhance the resistivity and dielectric energy storage performance of BFO-based

1 films. The  $V_{Bi} - V_O$  complexes are realized by deliberately introducing both O and Bi  
2 deficiencies with a low  $pO_2$  during film deposition. The charge carrier activation energies are  
3 greatly enhanced by the  $V_{Bi} - V_O$  complexes, as revealed by theoretical calculations and  
4 experimental characterizations. The film resistivity is thus improved by  $\sim 4$  orders of magnitude,  
5 which reduces the dielectric loss to as low as 0.02, improves the breakdown strength to 4.0 MV  
6  $cm^{-1}$  and thus gives rise to a  $\sim 460\%$  enhancement of energy density (from 14 to 79 J  $cm^{-3}$ ).  
7 Improved energy storage efficiency and operational reliability are also achieved. The vacancy  
8 complex offers a new, feasible and effective approach to address the longstanding bottleneck,  
9 i.e., inferior resistivity of BFO-based materials for high-voltage dielectric applications. This  
10 work could be a paradigm applicable for a wide range of materials (Pb-based, Bi-based,  
11 Na/K-based, *etc.*) in which vacancy complexes have a high possibility to form and to exert  
12 influence on dielectric and other functional properties.

## 13 **4. Experimental section**

### 14 *4.1 Film Preparation*

15 BFBSTO ceramic target was prepared with cold isostatic pressing and sintered by  
16 conventional solid-state reaction at 1100 °C for 2h. To compensate for the Bi volatilization  
17 during sintering, 15 mol% Bi was added in excess. The BFBSTO films were deposited in a PLD  
18 system with a KrF excimer laser. The laser wavelength was 248 nm, the repetition rate was 10  
19 Hz and the laser energy was  $\sim 1.5$  J  $cm^{-2}$ . (001)-oriented 0.7wt% Nb-doped SrTiO<sub>3</sub> (Nb:STO)  
20 single crystals were used as substrates and served as the bottom electrodes. The substrate  
21 temperature was kept at 700 °C during deposition. The BFBSTO films were grown under varied  
22  $pO_2$  in the range of 50 - 5 mTorr. After deposition, the films were in-situ annealed in 500 Torr O<sub>2</sub>  
23 for 30 min at 500 °C to reduce excess oxygen deficiency and then cooled down at a rate of 10 °C  
24  $min^{-1}$ . The film thicknesses were  $\sim 500$  nm, as confirmed by scanning electron microscopy (SEM,

1 Merlin, Zeiss). Circular Au top electrodes (~300  $\mu\text{m}$  in diameter and ~100 nm in thickness) were  
2 sputtered for electrical measurements.

### 3 4.2 Characterizations

4 Film crystal structure and epitaxial properties were characterized using an X-ray  
5 diffractometer (XRD; Smartlab, Rigaku) with Cu  $K_{\alpha}$  radiation ( $\lambda = 1.5418 \text{ \AA}$ ). The  
6 cross-sectional low-magnitude morphology and high-magnitude atomic images were obtained  
7 with the high-resolution scanning transmission electron microscopy (STEM; JEOL  
8 ARM200CF). The film composition was measured with inductively coupled plasma mass  
9 spectrometry (ICP-MS; iCAP Q, Thermo Fisher Scientific). X-ray photoelectron spectroscopy  
10 (XPS) was collected using a SPECS PHOIBOS 150 electron energy analyzer. Impedance  
11 properties were studied using an electrochemical impedance Spectroscopy (Solartron 1260 and  
12 1278) with an AC voltage of 100 mV, in the frequency range of 1 Hz to 5 MHz and temperature  
13 range of 543 to 623 K. The temperature was controlled by a home-made heating system. The  
14 dielectric permittivity and loss tangent were obtained with HP 4294A (Agilent) and LCR-8105G  
15 (GW Instek) with a perturbation AC voltage of 500 mV. The ferroelectric and leakage current  
16 properties were measured with Precision Multiferroic II (Radiant Technologies). Bipolar and  
17 unipolar  $P$ - $E$  loops were collected with 1 kHz triangular voltage waves. The leakage currents  
18 were obtained with a series of DC voltages, with 200-ms soak time and 200-ms measure time for  
19 each voltage.

### 20 4.3. First-principles Calculations

21 The first-principles calculations were based on density functional theory (DFT) within the  
22 local density approximation (LDA) for exchange-correlation potential as implemented in Vienna  
23 *Ab initio* Simulation Package (VASP). The onsite Coulomb interaction was included in the  
24 LDA+ $U$  approach with a Hubbard parameter  $U_{\text{eff}} = 4 \text{ eV}$  for Fe  $3d$  electrons. Projector  
25 augmented wave methods (PAW) were used to describe the electron-ion interaction with  
26 Bi( $5d^{10}6s^26p^3$ ), Fe( $3p^63d^64s^2$ ) and O( $2s^22p^4$ ). A periodic  $R3c$  structure of a  $2 \times 2 \times 2$  BFO supercell  
27 was employed with G-type antiferromagnetic order, as shown in Fig. S4. We simulated single Bi,  
28 O vacancies and Bi-O vacancy complex with different charge states. All defective supercells  
29 were relaxed while keeping the lattice parameters fixed at the perfect rhombohedral BFO values.

1 The plane-wave cutoff was set at 500 eV and the Brillouin Zone was sampled with a  $3 \times 3 \times 3$   
2  $k$ -point mesh. All atoms were allowed to move until the Hellmann-Feynman force on each atom  
3 was smaller than 0.01 eV/Å and the total energy was converged to  $10^{-5}$  eV. To determine the  
4 most stable Bi-O divacancy structure, seven configurations with different Bi-O divacancy  
5 distances were modelled. The nearest neighboring configuration was revealed to possess the  
6 lowest DFT total energy (Fig. S4), which was thus adopted in the following calculations.

7 The formation energy of defect  $D$  with charge state  $q$  was calculated as follows:

$$8 \quad E_f(D, q) = E(D, q) - E(\text{perfect}) + \sum_i \mu_i + q(E_F + E_{VBM}) \quad (1)$$

9 where  $E(D, q)$  and  $E(\text{perfect})$  were the DFT total energies of the defective and perfect  
10 supercells, respectively. The chemical potentials  $\mu_i$  of the removed atoms were related by  $\mu_i =$   
11  $\mu_i^0 + \Delta\mu_i$ , where  $\mu_i^0$  was the chemical potential of the elemental solid and  $\Delta\mu_i$  was determined  
12 by the thermodynamical constraints. The Fermi level  $E_F$  was defined relative to the valence band  
13 maximum ( $E_{VBM}$ ) and could vary from zero up to the band gap of BFO.  $E_{VBM}$  was described as  
14  $E_{VBM} = E(\text{perfect}) - E(\text{perfect} + 1)$ , with  $E(\text{perfect} + 1)$  being the energy of a perfect  
15 supercell in which an electron had been removed. For charged defects, a neutralizing background  
16 charge was applied, and their energies were corrected by the monopole part of Makov-Payne and  
17 by the band alignment onto the perfect supercell with the shift of the orbital energy of the  
18 semi-core Fe 3s state [36]. Based on Eqn. (1), the association energy of the  $V_{Bi} - V_O$  complex  
19 with various charge states was obtained:

$$20 \quad E_{assoc}(V_{Bi} - V_O, q_1 + q_2) = E_f(V_{Bi} - V_O, q_1 + q_2) - E_f(V_{Bi}, q_1) - E_f(V_O, q_2) \quad (2)$$

21 Where  $q_1$  and  $q_2$  are the charge state of  $V_{Bi}$  and  $V_O$ , respectively. The results are shown  
22 in Fig. S5. The transition energy level  $\epsilon(q/q')$  representing the energy required to ionize a  
23 defect with charge state  $q$  to another charge state  $q'$  was given by

$$24 \quad \epsilon_D(q/q') = \frac{E_f(D, q) - E_f(D, q')}{q' - q} \quad (3)$$

25

#### 26 4.4. Weibull Distribution Analysis

1 The characterized dielectric breakdown strength  $E_b$  of the samples was obtained with a  
2 two-parameter Weibull distribution function:

$$3 \quad P(E) = 1 - \exp(- (E / E_b)^\beta) \quad (4)$$

4 where  $E$  is the measurement breakdown field,  $P(E)$  is the cumulative probability of electric  
5 breakdown at  $E$ . When  $E$  equals  $E_b$ ,  $P(E)$  equals 63.2%. The Weibull parameter  $\beta$  evaluates the  
6 distribution of  $E$ . In this work, the breakdown field data of 10 samples for each  $pO_2$  were  
7 collected for the fitting.

8

### 9 **Declaration of competing interest**

10 The authors declare no conflict of interest.

11

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19

### 20 **Appendix A. Supplementary material**

21 Supplementary data associated with this article can be found in the online version.

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